Saengdoen Daungdaw 2009: Preparation of LaCoO₃ Perovskite for CO Gas Sensor. Master of Science (Chemistry), Major Field: Chemistry, Department of Chemistry. Thesis Advisor: Ms. Pinsuda Viravathana, Ph.D. 141 pages.

The LaCoO₃-ss (1:1, 1:5 (high Co), and 5:1 (high La)) perovskites were prepared by solid state reaction method. All LaCoO₃ samples were prepared and characterized by X-ray diffraction (XRD), X-ray absorption near-edge structure (XANES), BET surface area, and scanning electron microscopy (SEM). Their properties were compared with those of the samples prepared by co-precipitation (LaCoO₃-cp) and modified wet powder dispersion (LaCoO₃-wd) methods. The main crystalline phase of all prepared LaCoO₃ was a rhombohedral perovskite structure. The crystallite sizes of LaCoO₃-ss (1:1, high Co, and high La) at 850°C were 122, 42, and 73 nm and the crystallite sizes at 700°C of LaCoO₃-wd and LaCoO₃-cp were 28.5 and 28.2 nm, respectively. The surface areas were 1.6, 3.9, and 5.2 m^2g^{-1} for LaCoO₃-ss (1:1, high Co, and high La), respectively. From co-precipitation and modified wet powder dispersion methods, the surface areas were 11.8 and 11.7 m^2g^{-1} , respectively. From XANES spectra, the average Co oxidation state of the LaCoO₃-ss (1:1 and high La) could be \approx 3 and the average oxidation state of LaCoO₃-ss (high Co) was from 2 to 8/3. Moreover, SEM images showed the difference in morphology which strongly depended on the preparation methods. At the voltages of 7.5 volts, corresponded to temperature of 133°C measured at the LaCoO₃ surface, the resistance of prepared LaCoO₃ was lowest (highest conductivity). The fabricated sensors had been tested as CO gas sensors with the CO concentration of 1.98-19.84% v/v, 0.39-99.2% v/v, and 0.006-0.039% v/v at 7.5 volts. The LaCoO₃-ss (1:1) sensor performed the highest CO sensing capability compared to LaCoO₃-cp and LaCoO₃-ss (high Co). The detection limits of LaCoO₃-ss (1:1) and LaCoO₃-cp on CO sensing were of 0.006% (60 ppm) for CO concentration with the sensitivity of 1.007 and 1.013, respectively.

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