Research Title

Gas Adsorption on Activated Carbon at Low Temperature (80-200K)

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Abstract

This research has studied about gas-adsorption on activated carbon by passing sample gases through the activated carbon at low temperature (80 - 200 K) and detect gas-adsorption signals from the TCD (Thermal Conductivity Detector). These sample gases are argon, oxygen, nitrogen and carbon dioxide. Found that the quantity of gas-adsorption depend on temperature of activated carbon and physical property of gas. At low temperature(80 - 90 K), gas will be more adsorped than high temperature. But in this case of carbon dioxide adsorption with activated carbon, it can't be detected the response signal from the detector.

To calculate the surface of gas-adsorption of argon, oxygen, and nitrogen on activated carbon at 30 K, according to BET theorem, the existance value are 4280 $\rm m^2 g^{-1}$, 2456 $\rm m^2 g^{-1}$ and 58 $\rm m^2 g^{-1}$ respectively. And the surface of gas-adsorption of nitrogen on activated carbon at liquid nitrogen temperature is 1343 $\rm m^2 g^{-1}$