

## LITERATURE CITED

- Akiba, M. and A. S. Hashim. 1997. Vulcanization and crosslinking in elastomers. **Prog. Polym. Sci.** 22: 475-521.
- Blow, C. M. 1971. **Rubber Technology and Macnufacture**. Butterworth, London.
- \_\_\_\_\_ and C. Hepburn. 1985. **Rubber Technology and Manufacture**. 2<sup>nd</sup> ed. Butterworth, Heinemann.
- Britt, J.H., R.G. Scott, J.D. Armstrong and M.D. Whitacre. 1986. Determinants of estrous behavior in lactating Holstein cows. **J. Dairy Sci.** 69: 2195-2202.
- Bryk, M.T. 1991. **Degradation of filled polymers**. Ellis Horwood, England.
- Cho, K. and D. Lee. 2001. Effect of molecular weight between cross-links on the abrasion behavior of rubber by a blade abrader. **Polymer**. 41: 133-140.
- Costa, H. M., L.L.Y. Visconte, R.C.R. Nunes and C.R.G. Furtado. 2003. Rice-husk-ash-filled natural rubber. I. Overall Rate Constant Determination for the Vulcanization Process from Rheometric Data **J. Appl. Polym. Sci.** 87: 1194-1203.
- \_\_\_\_\_, \_\_\_\_\_, \_\_\_\_\_ and \_\_\_\_\_. 2003. Rice-husk-ash-filled natural rubber. II. Partial replacement of commercial fillers and the effect on the vulcanization process. **J. Appl. Polym. Sci.** 87: 1405-1413.
- Diskin, M.G. and J.M. Sreenan. 2000. Expression and detection of estrus in cattle. **Reprod. Nutr. Dev.** 40: 481-491.
- Dyer, A. 1988. **An Introduction to Zeolite Molecular Sieves**. John Wiley & Sons Ltd, United state of America.

Fred, W.B. 1993. **Rubber Compounding: Principles, Material, and Techniques.** 2<sup>nd</sup> ed. Marcel Dekker, New York.

Fukahori, Y. and H. Yamazaki. 1994a. Mechanism of rubber abrasion Part 1 Abrasion pattern formation in natural rubber vulcanizate. **Wear.** 171: 195-202.

\_\_\_\_\_ and \_\_\_\_\_. 1994b. Mechanism of rubber abrasion Part 2 General rule in abrasion pattern formation in rubber-like materials. **Wear.** 178: 109-116.

\_\_\_\_\_ and \_\_\_\_\_. 1995. Mechanism of rubber abrasion Part 3 How is friction linked to fracture in rubber abrasion. **Wear.** 188: 19-26.

Hofmann, W. 1989. **Rubber Technology Handbook.** Hanser Publishers, Munic Vienna, New York.

Ismail, H., P.K. Freakloy, I. Sutherland and E. Sheng. 1995. Effects of multifunctional additive on mechanical properties of silica filled natural rubber compound. **Eur. Polym. J.** 31: 1109-1117.

\_\_\_\_\_, R. Nordin and A.M. Noor. 2002. Cure characteristics tensile properties and swelling behavior of recycled rubber powder-filled natural rubber compounds. **Polymer Testing.** 21: 565-569.

Jame, E. M., B. Erman and R. E. Frederick. 1994. **Science and Technology of Rubber.** 2<sup>nd</sup> ed. Academic Press.

Korkuna, O., R. Leboda, J. Skubiszewska-Zięba, T. Vrublevs'ka, V.M. Gun\_ko, and J. Ryczkowski. 2005. Structural and physicochemical properties of natural zeolites: clinoptilolite and mordenite. **Microporous and Mesoporous Materials.** 87: 243–254.

Lauriente, D.H. and Y. Inoguchi. n.d. **CEH Marketing Research Report, Zeolites.**

Morton, M. 1959. **Introduction to Rubber Technology.** Reinhold Publisher Corporation, New York.

Özmihçi, F., D. Balköse and S. Ülku. 2001. Natural zeolite polypropylene composite film preparation and characterization. **J. Appl. Polym. Sci.** 82: 2913-2921.

Pal, P.K., A.K. Bhowmick and S.K. De. 1982. Interplay of HAF black filler and vulcanization system on technical properties and network structure of natural rubber. **Rubb. Chem. Technol.** 55: 23.

Pehlivan, H., D. Balköse, S. Ülkü and F. Tihminlioğlu. 2006. Effect of zeolite filler on the thermal degradation kinetics of polypropylene. **J. Appl. Polym. Sci.** 101: 143-148.

Perdana, S.S.. n.d. **Design and Molecular Engineering of Nanostructured Zeolites and Mesomorphous Materials – Advancing Through the Pore.** n.p.

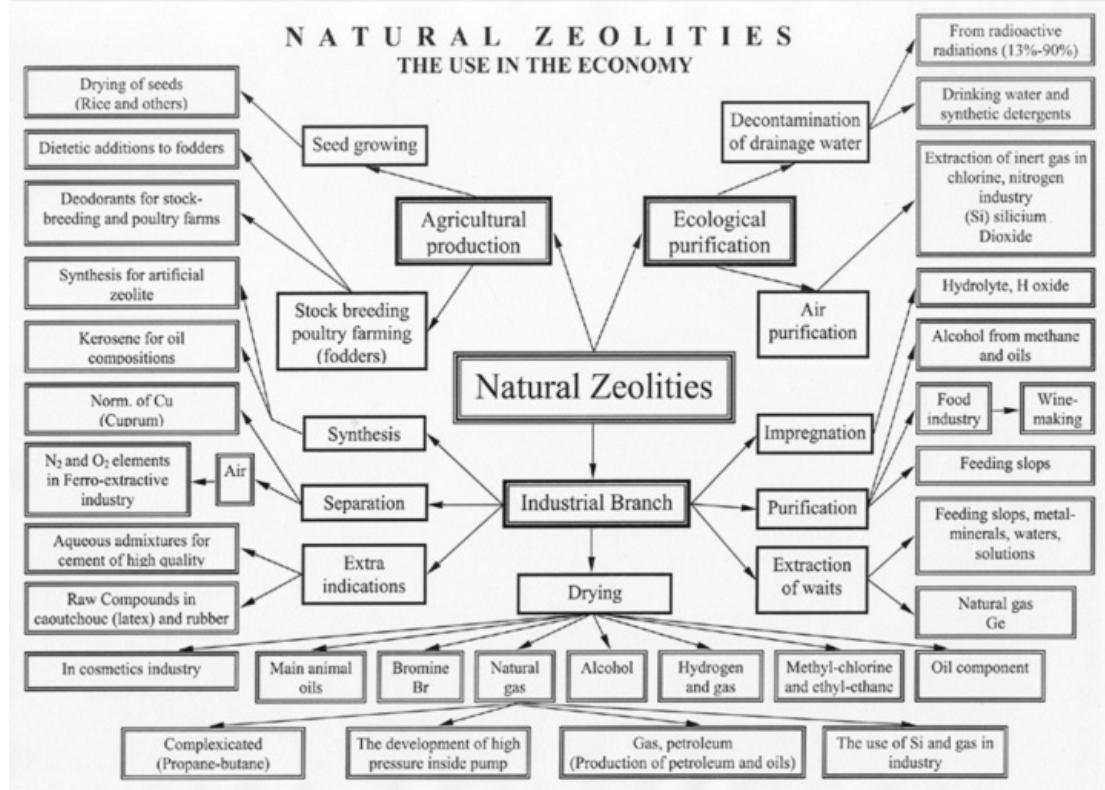
Rattanasom, N., A. Poonsuk and T. Makmoon. 2005. Effect of curing system on the mechanical properties and heat aging resistance of natural rubber/tire tread reclaimed rubber blends. **Polymer Testing.** 24: 728-732.

Rodtian,P., G. King, S. Subrod and P. Pongpiachan. 1996. Oestrus behaviour of Holstein cows during cooler and hotter tropical seasons. **Anim. Reprod. Sci.** 45: 47-58.

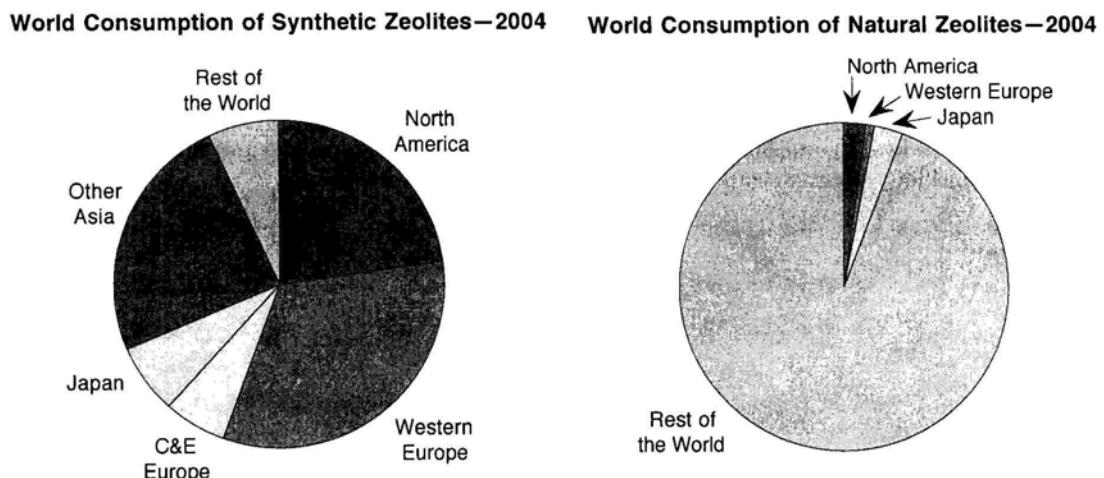
Sae-oui, P., U. Thepsuwan and K. Hatthapanit. 2004. Effect of curing system on reinforcing efficiency of silane coupling agent. **Polymer Testing.** 23: 397-403.

- Shah, V. 1998. **Handbook of Plastic Testing Technology.** Wiley, USA.
- Thavamani, P. and A.K. Bhowmick. 1993. Abrasion of vulcanizates of NR and SBR at high temperatures. **Plast. Rubb. Compos. Process. Appl.** 20: 239.
- Tudsri, S. and S. Sawasdipanit. 1998. Managerial approach to pasture production in Thailand. 103-109. In C.P. Chen and C. Satjianon, eds. **Strategies for Suitable Forage-Based Livestock Production in Southeast Asia.** The 3<sup>rd</sup> meeting of regional working group on grazing and feed resources of Southeast Asia, Khon Khaen.
- Vinod, V.S., S. Varghese and B. Kuriakose. 2002. Degradation behaviour of natural rubber–aluminium powder composites: effect of heat, ozone and high energy radiation. **Polymer Degradation and Stability.** 75: 405-412.
- Xanthos, M. 2005. **Functional Fillers for Plastics.** Wiley – VCH, Weinheim.

## **APPENDIX**



**Appendix Figure 1** Representative utilizations of zeolite



**Appendix Figure 2** World Consumption of Synthetic and Natural Zeolites in 2004

**Appendix Table 1** Global consumption of zeolites (2000)

Zeolite type	Consumption in thousands of metric tons per annum			
	United States	Western Europe	Japan	Other Global Regions
<b>Total</b>				
Natural zeolites	32	30	120	3800
Synthetic zeolites	470	606	110	176
Detergents	365	555	95	145
1160(85%)				
Catalysts	70	26	10	11
Absorbents	35	25	5	20
<b>Total</b>	<b>502</b>	<b>636</b>	<b>230</b>	<b>3976</b>
				<b>5344</b>



Request No. 55/49 MATERIALS TECHNOLOGY DEPARTMENT (MTD) Lab No. 59/49

Date March 31, 2006

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### REPORT ON TESTING AND ANALYSIS

FOR

KASETSART UNIVERSITY

Testing/analysis of :- Zeolite, CaCO<sub>3</sub>, RHA (30) and RHA (AA)

Method of testing/analysis :- Surface Area Analysis (BET); Model: Autosorb 1 and Particle Size Analysis (Wet Sieve)

Result of testing/analysis :-

The results of Particle size by wet sieving of the samples are shown in the table on page 2. The results of surface area of the samples are shown in the table below. The details are shown on pages 3-14.

Sample	Single point BET (m <sup>2</sup> /g)	Multipoint BET (m <sup>2</sup> /g)	Remark
Zeolite	14.30	14.59	Pages 3-5
CaCO <sub>3</sub>	2.99	3.07	Pages 6-8
RHA (30)	78.54	79.92	Pages 9-11
RHA (AA)	76.58	77.04	Pages 12-14

Tested/analysed by

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FM-MTD-GEN 02-02 Rev.0

Remark : The above results are valid exclusively for tested/analysis sample as mentioned in the report  
Publicity of the results on testing and analysis is prohibited unless written permission is obtained from the governor of TISTR

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**Appendix Figure 3 Report on Surface area of various fillers**



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The amount of residue on sieve.

	% Residue on sieve			
	+150µm	+75µm	+45µm	-45µm
Zeolite	8.13	16.76	10.54	64.57
CaCO <sub>3</sub>	0.51	0.41	0.33	98.75
RHA (30)	70.25	15.32	6.38	8.05
RHA (AA)	68.07	17.72	6.91	7.30



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**Appendix Figure 4** Report on Particle size of various fillers

Date: 04/03/2006

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Quantachrome Corporation  
 Quantachrome Autosorb Automated Gas Sorption System Report  
 Autosorb for Windows® Version 1.19

Sample ID	Zeolite
Description	TISTR
Comments	
Sample Weight	0.4116 g
Adsorbate	NITROGEN
Cross-Sec Area	16.2 Å <sup>2</sup> /molecule
NonIdeality	6.580E-05
Molecular Wt	28.0134 g/mol
Station #	1
	Outgas Temp 300.0 °C
	Outgas Time 3.0 hrs
	P/PO Toler 0
	Equil Time 3
	Bath Temp. 77.40
Operator	RUNGRUENG
Analysis Time	64.8 min
End of Run	01/03/1980 0:
File Name	AS010450.RAW

## AREA-VOLUME-PORE SIZE SUMMARY

## SURFACE AREA DATA

Multipoint BET.....	1.459E+01	m <sup>2</sup> /g
Single Point BET.....	1.430E+01	m <sup>2</sup> /g
Langmuir Surface Area.....	2.294E+01	m <sup>2</sup> /g
t-Method External Surface Area.....	1.366E+01	m <sup>2</sup> /g
t-Method Micro Pore Surface Area.....	9.273E-01	m <sup>2</sup> /g
DR Method Micro Pore Area.....	1.875E+01	m <sup>2</sup> /g

## PORE VOLUME DATA

t-Method Micro Pore Volume.....	3.825E-04	cc/g
DR Method Micro Pore Volume.....	6.662E-03	cc/g
HK Method Cumulative Pore Volume.....	5.906E-03	cc/g
SF Method Cumulative Pore Volume.....	6.039E-03	cc/g

## PORE SIZE DATA

DR Method Micro Pore Width .....	9.669E+01	Å
DA Method Pore Diameter (Mode).....	1.780E+01	Å
HK Method Pore Width (Mode).....	1.402E+01	Å
SF Method Pore Diameter (Mode).....	2.639E+01	Å

## DATA REDUCTION PARAMETERS

Thermal Transpiration : ON  
 Effective Molecule Diameter (D) 3.5400 Å  
 Effective Cell Stem Inner Diameter (d) 4.0000 mm  
 Last Po Acquired 794.10 mm Hg  
 Additional Initialization Information Not Recorded

BJH/DH Moving Average Size : 1

Interaction Constant (K) 2.9600 nm<sup>3</sup> x kJ/mol TISTR

Appendix Figure 5 Report on Pore volume of natural zeolite

Date: 04/03/2006

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Quantachrome Corporation  
 Quantachrome Autosorb Automated Gas Sorption System Report  
 Autosorb for Windows® Version 1.19

Sample ID	CaCO3
Description	TISTR
Comments	
Sample Weight	0.2364 g
Adsorbate	NITROGEN
Cross-Sec Area	16.2 Å <sup>2</sup> /molecule
NonIdeality	6.580E-05
Molecular Wt	28.0134 g/mol
Station #	1
Outgas Temp	300.0 °C
Outgas Time	3.0 hrs
P/Po Toler	0
Equil Time	3
Bath Temp.	77.40
Operator	RUNGRUENG
Analysis Time	60.6 min
End of Run	03/20/2006 15
File Name	AS010449.RAW

## AREA-VOLUME-PORE SIZE SUMMARY

## SURFACE AREA DATA

Multipoint BET.....	3.078E+00	m <sup>2</sup> /g
Single Point BET.....	2.994E+00	m <sup>2</sup> /g
Langmuir Surface Area.....	4.792E+00	m <sup>2</sup> /g
t-Method External Surface Area.....	2.582E+00	m <sup>2</sup> /g
t-Method Micro Pore Surface Area.....	4.963E-01	m <sup>2</sup> /g
DR Method Micro Pore Area.....	4.045E+00	m <sup>2</sup> /g

## PORE VOLUME DATA

t-Method Micro Pore Volume.....	2.541E-04	cc/g
DR Method Micro Pore Volume.....	1.437E-03	cc/g
HK Method Cumulative Pore Volume.....	1.308E-03	cc/g
SF Method Cumulative Pore Volume.....	1.338E-03	cc/g

## PORE SIZE DATA

DR Method Micro Pore Width .....	8.998E+01	Å
DA Method Pore Diameter (Mode).....	1.780E+01	Å
HK Method Pore Width (Mode).....	1.773E+01	Å
SF Method Pore Diameter (Mode).....	3.368E+01	Å

## DATA REDUCTION PARAMETERS

Thermal Transpiration : ON  
 Effective Molecule Diameter (D) 3.5100 Å  
 Effective Cell Stem Inner Diameter (d) 4.0000 mm  
 Last Po Acquired 798.09 mm Hg  
 Additional Initialization Information Not Recorded.

BJH/DH Moving Average Size : 1

Interaction Constant (K) 2.9600 nm<sup>3</sup> x kJ/mol**TISTR**Appendix Figure 6 Report on Pore volume of CaCO<sub>3</sub>

Date: 04/03/2006

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Quantachrome Corporation  
 Quantachrome Autosorb Automated Gas Sorption System Report  
 Autosorb for Windows® Version 1.19

Sample ID	RHA(30)
Description	TISTR
Comments	
Sample Weight	0.1361 g
Adsorbate	NITROGEN
Cross-Sec Area	16.2 Å <sup>2</sup> /molecule
NonIdeality	6.580E-05
Molecular Wt	28.0134 g/mol
Station #	1
	Outgas Temp 300.0 °C
	Outgas Time 3.0 hrs
	P/Po Toler 0
	Equil Time 3
	Bath Temp. 77.40
	Operator RUNGRUENG
	Analysis Time 94.2 min
	End of Run 03/29/2006 1:
	File Name AS010452.RAW

## AREA-VOLUME-PORE SIZE SUMMARY

## SURFACE AREA DATA

Multipoint BET.....	7.992E+01	m <sup>2</sup> /g
Single Point BET.....	7.854E+01	m <sup>2</sup> /g
Langmuir Surface Area.....	1.237E+02	m <sup>2</sup> /g
t-Method External Surface Area.....	6.630E+01	m <sup>2</sup> /g
t-Method Micro Pore Surface Area.....	1.361E+01	m <sup>2</sup> /g
DR Method Micro Pore Area.....	1.067E+02	m <sup>2</sup> /g

## PORE VOLUME DATA

t-Method Micro Pore Volume.....	6.870E-03	cc/g
DR Method Micro Pore Volume.....	3.793E-02	cc/g
HK Method Cumulative Pore Volume.....	3.388E-02	cc/g
SF Method Cumulative Pore Volume.....	3.453E-02	cc/g

## PORE SIZE DATA

DR Method Micro Pore Width .....	9.311E+01	Å
DA Method Pore Diameter (Mode).....	1.800E+01	Å
HK Method Pore Width (Mode).....	1.412E+01	Å
SF Method Pore Diameter (Mode).....	2.648E+01	Å

## DATA REDUCTION PARAMETERS

Thermal Transpiration : ON  
 Effective Molecule Diameter (D) 3.5400 Å  
 Effective Cell Stem Inner Diameter (d) 4.0000 mm  
 Last Po Acquired 788.57 mm Hg  
 Additional Initialization Information Not Recorded

BJH/DH Moving Average Size : 1 **TISTR**Interaction Constant (K) 2.9600 nm<sup>3</sup> x kJ/mol

Appendix Figure 7 Report on Pore volume of RHA

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**SCHOLARSHIP** : Thai Research Fund (TRF) 2005-2006