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The adsorptions of CO and NO on Ag-ZSM-5 have been studied by using the

Surface Charge Representation of External Embedded Potential (SCREEP) embedded

cluster model at the density functional theory. It was found that CO adsorbs on the Ag+

ion in the linear configuration with the bond distance of 1.130 Å and binding energy of

18.63 kcal/mol. Adsorbed NO, on the other hand has a bend configuration with the

AgNO angle of 127.5° and the Ag-N bond distance of 1.150 Å. The binding energy for

NO is 9.24 kcal/mol. By using natural bond orbital analysis (NBO), it could be seen that

the former has stronger a σ donation and also a greater π -back donation. The Madelung

potential has only a small perturbation on the structure and binding energy of the

adsorption complexes. This environmental effect, however, improves the agreement

between the calculated frequency shifts and those of experimental observations.

The interaction of a gold atom with single-wall carbon nanohorn (SWNH) and

single-wall carbon nanotube (SWNT) tips and their complexes with a CO molecule were

studied using the density functional theory (DFT). The analysis of the pyramidalization

angle (θ_n) as well as the π -orbital misalignment angles (ϕ) lead to an elucidation that

there would be many reactive carbon sites on the tips of both SWNH and SWNT, the

former provides reactive sites that can more selectively interact with the target atom. The

CO probe molecule binds tightest with the Au/SWNH tip and the Au/SWNT tip with the

interaction energies of -22.34 and -18.29 kcal/mol, respectively. The main contribution of

the metal interacted with both carbon nanostructures stemmed from the interactions of σ -

donation and π -back bonding. The results suggest the potential of the use of an SWNH

for the development of a high-specific nanosensor.

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