TABLE OF CONTENTS

| | Page |
|--|------|
| TABLE OF CONTENTS | i |
| LIST OF TABLES | ii |
| LIST OF FIGURES | iii |
| LIST OF ABBREVIATIONS | V |
| INTRODUCTION | 1 |
| LITERATURE REVIEW | 11 |
| METHODS OF CALCULATIONS | 25 |
| RESULTS AND DISCUSSION | |
| Chapter I. Theoretical study on structural and electronic | |
| properties of the BN-doped Carbon Nanotubes grafted with | |
| N-nucleophiles | 36 |
| Chapter II. Theoretical study on structural and electronic | |
| properties of the BN-doped Fullerene grafted with | |
| heterocyclic N-nucleophiles | 50 |
| CONCLUSION | 60 |
| LITERATURE CITED | |

LIST OF TABLES

| Γable | | Page |
|-------|--|------|
| 1 | Overview of the important synthesis procedures for single-walled | 4 |
| | carbon nanotubes. | |
| 2 | Structural parameters of the BN-doped SWCNT | |
| | grafted with N-nucleophiles | 39 |
| 3 | Interaction parameters between the BN-doped SWCNT and | |
| | various N-nucleophiles | 40 |
| 4 | Relative energy gap reduction of systems studied | |
| | compared to C_{60} ($E_{gap} = 1.64 \text{ eV}$) | 50 |
| 5 | Molecular orbital contributions | 59 |

LIST OF FIGURES

| Figure | | Page |
|--------|---|------|
| 1 | Schematics of an individual (A) SWCNT and (B) MWCNT | 2 |
| 2 | Determination of the chiral vector | 3 |
| 3 | Peptide formation from the reaction of amino acid and carboxylic | |
| | acid attached on side-SWCNTs | 6 |
| 4 | Positions of mono & tri BN-substitution on side-wall (5,5) | |
| | armchair SWCNTs | 8 |
| 5 | Geometry of the BN-doped fullerene | 10 |
| 6 | Position of R-group functionalization on side-wall (5,5) SWCNT | 32 |
| 7 | Structure of Nitrogen nucleophiles | 32 |
| 8 | Illustrations of a) Fullerene (C ₆₀), b) BN-doped fullerene | |
| | $(C_{58}BN)$, c) 2,6- Naphyridine $(C_8H_6N_2, 1)$, d) 3,8-Phenanthroline | |
| | $(C_{12}H_8N_2,\textbf{2})$ and e) 2,6-Diazaantracene $(C_{12}H_8N_2,\textbf{3})$ structures. | 34 |
| 9 | Bond lengths and charge distribution of BN-doped (5,5)SWCNT | 38 |
| 10 | Position of R groups, B atom, N atom, C1 atom, C2 atom on | |
| | side-wall (5,5) SWCNT | 39 |
| 11 | Correlation between binding energies and pKa values of BN- | |
| | doped SWCNT/N-nucleophiles complexes | 41 |
| 12 | Charge distribution of isolated m-nitroaniline and m-nitroaniline | |
| | complexed with BN-doped SWCNT | 42 |
| 13 | Charge distribution of isolated pyridine and pyridine complexed | |
| | with BN-doped SWCNT | 43 |
| 14 | Partial density of states of pyridine complexed with BN-doped | |
| | SWCNT | 44 |
| 15 | Partial density of states of m-nitroaniline complexed with BN- | |
| | doped SWCNT | 44 |
| 16 | Partial density of states of guanidine complexed with BN-doped | |
| | SWCNT | 45 |

LIST OF FIGURES (Continued)

| Figure | | Page |
|--------|---|------|
| 17 | HOMO and LUMO orbital energies of isolated pyridine and | |
| | pyridine complexed with BN-doped SWCNT | 48 |
| 18 | HOMO and LUMO orbital energies of isolated m-nitroaniline | |
| | and m-nitroaniline complexed with BN-doped SWCNT | 49 |
| 19 | HOMO and LUMO orbital energies and energy gap of isolated | |
| | 2,6-diazanaphthalene (1) and $C_{58}BN/1/C_{58}BN$ complex | 53 |
| 20 | HOMO and LUMO orbital energies and energy gap of isolated | |
| | 3,8-phenanthroline (2) and C ₅₈ BN/2/C ₅₈ BN complex | 54 |
| 21 | HOMO and LUMO orbital energies and energy gap of isolated | |
| | 2,6- diazaanthracene (3) and C ₅₈ BN/3/C ₅₈ BN complex | 55 |
| 22 | Illustrations of mono- and di-C ₆₀ complexes with 2,6 | |
| | diazanaphthalene (1), 3,8-diazaphenanthrene (2) and 2,6- | |
| | diazaanthracene (3) | 57 |
| 23 | Molecular orbital contour plots for C ₅₈ BN/1/C ₅₈ BN, | |
| | C ₅₈ BN/2/C ₅₈ BN and C ₅₈ BN/3/C ₅₈ BN systems | 58 |

LIST OF ABBREVIATIONS

B = Boron

B3LYP = Beck's three-parameter hybrid functional

using the LYP correlation functional

CNTs = Carbon nanotubes

CVD = Chemical vapor deposition

DOSs = Density of states

 E_g = Energy gap

eV = electron volt

HOMO = Highest Occupied Molecular Orbital

 I_h = Icosahedral

IPR = Isolated-pentagon rule

kcal/mol = kilocalorie per mole

LUMO = Lowest Occupied Molecular Orbital

MWCNTs = Multi-walled carbon nanotubes

N = Nitrogen

NPA = Natural Population Analysis

OPT = Optimization

PAPR = Pentagon adjacency penalty rule

PEG = Poly(ethylene glycol)

SWCNTs = Single-walled carbon nanotubes