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ริทยามิหาย์มียื่มส่วนหนึ่งของการที่กษาตามหลักสูตรปริญญาวิทยาการตรมหายันที่ต สายาวิชาบีโตยเคยี่มละวิทยาตาสตร์สอธิเมอร์ อัการก็บษา 2551 ถึงเห็บษา 2551 L00255589



การสังเคราะห์และประยุกต์ควอเทอร์นารีแอมโมเนียมไคโทซานเป็นสารเติมแต่งในผลิตภัณฑ์บำรุงเส้นผม

นายธีระชัย เกิดชลเพ็ชร



วิทยานิพนธ์นี้เป็นส่วนหนึ่งของการศึกษาตามหลักสูตรปริญญาวิทยาศาสตรมหาบัณฑิต สาขาวิชาปิโตรเคมีและวิทยาศาสตร์พอลิเมอร์ คณะวิทยาศาสตร์ จุฬาลงกรณ์มหาวิทยาลัย ปีการศึกษา 2551 ลิขสิทธิ์ของจุฬาลงกรณ์มหาวิทยาลัย



SYNTHESIS AND APPLICATIONS OF QUATERNARY AMMONIUM CHITOSAN AS ADDITIVES IN HAIR-CARE PRODUCTS

Mr. Teerachai Kerdcholpetch

A Thesis Submitted in Partial Fulfillment of the Requirements

for the Degree of Master of Science Program in Petrochemistry and Polymer Science

Faculty of Science

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Ву	Mr. Teerachai Kerdcholpetch
Field of Study	Petrochemistry and Polymer Science
Advisor	Assistant Professor Varawut Tangpasuthadol, Ph.D.
Co-Advisor	Assistant Professor Voravee P. Hoven, Ph.D.
	ted by the Faculty of Science, Chulalongkorn University in Partial Fulfillment
of the Requirements f	for the Master's Degree
	S. Harmongheea Dean of the Faculty of Science
	Dean of the Faculty of Science
	(Professor Supot Hannongbua, Dr.rer.nat.)
THESIS COMMITTI	EE .
	Gy- Z. Chairman
	(Associate Professor Supawan Tantayanon, Ph.D.)
	AA2
	Advisor
	(Assistant Professor Varawut Tangpasuthadol, Ph.D.)
	1/2 1/4
	Vp. Haven. Co-Advisor
	(Assistant Professor Voravee P. Hoven, Ph.D.)
	Number Chartagaig Examiner
	(Associate Professor Nuanphun Chantarasiri, Ph.D.)
	Radahul kar External Examiner
	(Assistant Professor Radchada Buntem, Ph.D.)

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เส้นผมของมนุษย์มีประจุลบ ณ สภาวะที่มีพีเอชสูงกว่าค่าไอโซอิเล็กทริก (~3.67) ดังนั้น ส่วนผสมที่มีประจุบวกสามารถเคลือบติดเส้นผมได้ง่าย ในงานนี้ได้สังเคราะห์อนุพันธ์ไคโทซานที่มี ประจุบวกสองชนิดคือ เอ็น,เอ็น,เอ็น-ไทรเมทิลแอมโมเนียมไคโทซานคลอไรด์ (TMC) และเอ็น-[(2-ไฮดรอกซิล-3-ไทรเมทิลแอมโมเนียม)โพรพิล]ไคโทซานคลอไรด์ (HTACC) ที่มีระดับควอเทอร์ไนเซ ขัน (DQ) ต่างๆ กันเพื่อใช้ในน้ำยาปรับสภาพเล้นผมแบบซะโลมทิ้งไว้ ได้ยืนยันโครงสร้างทางเคมี ของอนุพันธ์ที่สังเคราะห์ได้ด้วยโปรตอนเอ็นเอ็มอาร์และเอฟทีไออาร์ จากการทดสอบความเป็นพิษ ต่อเซลล์ในหลอดทดลองโดยวิธีเอ็มที่ที่ พบว่า ความเป็นพิษของ HTACC ต่อเซลล์ชนิด HaCaT ขึ้นอยู่กับ %DQ ได้เลือกใช้ TMC (22%DQ) และ HTACC (24%DQ) ในสารปรับสภาพเล้นผม เพราะสารทั้งสองมีความเป็นพิษต่ำที่สุดและยังมีต้นทุนการสังเคราะห์ต่ำ น้ำยาปรับสภาพเล้นผม ที่เตรียมได้นี้สามารถเคลือบติดบนเล้นผมได้ดังผลการวิเคราะห์จากกล้องจุลทรรศน์อิเล็กตรอน แบบส่องกราดและเอทีอาร์เอฟทีไออาร์ไมโครสเปกโทรสโกปี นอกจากนี้ ยังได้วิเคราะห์สมบัติการ ทนต่อแรงดึง, เนื้อสัมผัส และการหวีแบบเปียกของเล้นผมพบว่าน้ำยาปรับสภาพเล้นผมที่เตรียมได้ นี้สามารถปรับสภาพเล้นผมที่ถูกดัดและยืดให้มีความแข็งแรงเพิ่มขึ้นและให้ความนุ่มลี่นในการ ล้มผัสมากกว่าผลิตภัณฑ์ที่ขายตามท้องตลาด ผลการศึกษานี้สมควรที่จะนำไปสู่การประยุกต์ใช้ เชิงพาณิชย์ของสารประกอบไคโทซานที่มีประจุบวกในอุตสาหกรรมถนอมเล้นผม

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At any conditions with pH above the isoelectric point (~3.67), the surface of hair bears a net negative charge. Therefore, positively charged ingredients can be easily coated on it. In this work, two positively charged derivatives of chitosan, N,N,N-trimethylammonium chitosan chloride (TMC) and N-[(2-hydroxyl-3trimethylammonium)propyl]chitosan chloride (HTACC), with varying degrees of quaternization (DQ) were synthesized for use in "leave-on" conditioners. The chemical structures of the charged derivatives were verified by ¹H NMR and FTIR. From in vitro cytotoxicity study by MTT assay, %DQ dependent on toxicity against HaCaT cells line was observed for HTACC. TMC (22%DQ) and HTACC (24%DQ) were used as cationic ingredients in the conditioner because of their lowest toxicity as well as synthesis costs. The prepared leave-on conditioners were able to coat on hair as evidenced by scanning electron microscopy and ATR-FTIR microspectroscopy. In addition, tests for hair's tensile properties, texture, and wet combing were also evaluated. The prepared conditioners could increase the strength and the smooth feel of waved and straightened hairs more than did the commercial product. This study can in fact pave the way for commercial application of the positively charged chitosan compounds in hair-care industry.

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Academic Year : 2008	Advisor's Signature :	and the
	Co-Advisor's Signature	: Vp. Horem.

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LIST OF ABBREVIATIONS

Å : Angstroms

ATR : Attenuated total reflectance

DD : Degree of deacetylation

DQ : Degree of quaternization

eq : Equivalent

FBS : Fetal bovine serum

FTIR : Fourier transform infrared

Ge : Germanium

GTMAC : Glycidyltrimethylammonium chloride

HaCaT : Human keratinocyte cells line

IRE : Internal reflection element

CH₃I : Iodomethane

kV : Kilovolt (10^3V)

LSM : Laser scanning micrometer

LO : Leave-on without cationic polymers

LO+1%quat : Leave-on with 1%w/w polyquaternium-10

LO+1%chitosan : Leave-on with 1%w/w chitosan

LO+1%TMC : Leave-on with 1%w/w TMC

LO+1%HTACC : Leave-on with 1%w/w HTACC

MCT : Mercury-cadmium-telluride

MTT : Miniature tensile tester

 μm : Micrometer (10⁻⁶ m)

nm

: Nanometer (10⁻⁹ m)

HTACC

 $: N\hbox{-}[(2\hbox{-hydroxyl-3-trimethylammonium}) propyl] chitosan \ chloride$

HEPES

: N-(2-hydroxyethyl)piperazine-N-(2-ethanosulphonic acid)

TMC

: N,N,N-trimethylammonium chitosan chloride

NMR

: Nuclear magnetic resonance spectroscopy

ppm

: Part per million

SEM

: Scanning electron microscopy

W/W

: Weight per weight

ZnSe

: Zinc selenide

MTT

: 3-(4,5-dimethylthiazol-2-yl)-2,5-diphenyltetrazolium bromide