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MODIFIED CARBON NANOTUBES FOR DRUG DELIVERY APPLICATIONS

MISS CHULARAT IAMSAMA

A DISSERTATION SUBMITTED IN PARTIAL FULFILLMENT OF THE REQUIREMENTS
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(INTERDISCIPLINARY PROGRAM)
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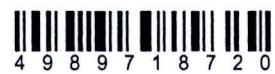
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ที่มาขอรับอนุญาตในดัดแปลงเพื่อการประยุกต์ใช้ในการนำเสนอ

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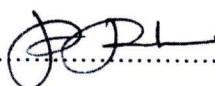
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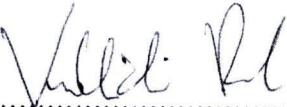
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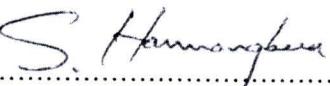
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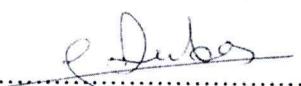
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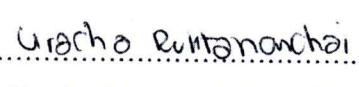
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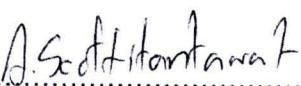
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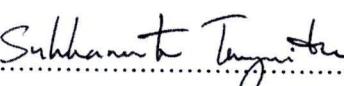
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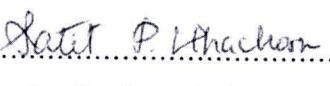
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การดัดแปรพื้นผิวของท่อคาร์บอนนาโนดั้งหลาวยั่น (MWCNTs) แบบอนโนโคลาเลนต์ โดยการเคลือบพื้นผิวแบบขั้นเดียวด้วยไฮโดรเจนไดออกไซด์ในงานวิจัยนี้ ระดับการทำจัดหมู่อะซิทิก (%DD) ของไฮโดรเจน (61%, 71%, 78%, 84%, 90% และ 93%) และความเข้มข้นของไฮโดรเจนที่แตกต่างกัน มีผลต่อความสามารถในการกระจายตัวและความเสถียรของท่อคาร์บอนนาโนดั้งหลาวยั่น ผลการทดลองแสดงว่า ไฮโดรเจนที่มีระดับการทำจัดหมู่อะซิทิกต่ำที่สุด (61%) สามารถปั๊บปุ่งการกระจายตัวของท่อคาร์บอนนาโนดั้งหลาวยั่น จากการวัดค่า zeta potential ยืนยันได้ว่า ไฮโดรเจนที่มีระดับการทำจัดหมู่อะซิทิก 61% สามารถปักอยู่พื้นผิวท่อคาร์บอนนาโนดั้งหลาวยั่นได้สูงกว่าสองเท่าเมื่อเทียบกับไฮโดรเจนที่มีระดับการทำจัดหมู่อะซิทิก 93% เพื่อศึกษาไฮโดรเจนซึ่งเป็นพอลิแซคไคร์ดว่าช่วยเพิ่มความสามารถในการกระจายตัวและความเสถียรให้กับท่อคาร์บอนนาโนได้อย่างไร ดังนั้น Molecular Dynamics Simulation ถูกนำมาใช้เพื่อทำการศึกษา 3 ระบบ ได้แก่ i) ท่อคาร์บอนนาโนที่ไม่ผ่านการทำดัดแปรทั้งคู่ (*p*CNT-*p*CNT), ii) ท่อคาร์บอนนาโนที่ไม่ผ่านการทำดัดแปรกับไฮโดรเจนที่มีระดับการทำจัดหมู่อะซิทิก 60% (*p*CNT-*cw*CNT) และ iii) ท่อคาร์บอนนาโนที่ผ่านการทำดัดแปรด้วยไฮโดรเจนที่มีระดับการทำจัดหมู่อะซิทิก 60% ทั้งคู่ (*cw*CNT-*cw*CNT) การรวมตัวกันของท่อคาร์บอนนาโนเกิดขึ้นในกรณีที่เป็น *p*CNT-*p*CNT และ *p*CNT-*cw*CNT เนื่องจากแรง van der Waals ที่เกิดขึ้นระหว่างห่วงแหวนอะโนมาติกของท่อคาร์บอนนาโนและการเกาะกันระหว่างท่อคาร์บอนนาโนด้วยไฮโดรเจน ตามลำดับ ส่วนในกรณี *cw*CNT-*cw*CNT พบว่า เกิดการผลักกันระหว่างปะจุส่งผลให้ท่อทั้งสองแยกออกจากกันและกระจายตัวได้ในสารละลาย แม้ว่าท่อคาร์บอนนาโนดั้งหลาวยั่นที่ผ่านการทำดัดแปรโดยการเคลือบพื้นผิวแบบขั้นเดียวด้วยไฮโดรเจนที่มีระดับการทำจัดหมู่อะซิทิกต่ำ สามารถปั๊บปุ่งความสามารถในการกระจายตัวของท่อคาร์บอนนาโนดั้งหลาวยั่นได้ แต่ความเสถียรของท่อคาร์บอนนาโนดั้งหลาวยั่นไม่เทียบเท่าพ่อการนำมายังท่อที่มีความเสถียรของท่อคาร์บอนนาโนดั้งหลาวยั่นได้ แต่ความเสถียรของท่อคาร์บอนนาโนดั้งหลาวยั่นได้ด้วยวิธี layer-by-layer deposition เป็นวิธีที่มีศักยภาพที่ถูกเลือกใช้เพื่อเตรียมพิล์มบางหลาวยั่นระหว่างพอลิ(ไดอะลิลไดเมทิลแอมโมเนียม คลอไรด์) (PDADMAC) และพอลิ(โซเดียม 4-สไตรีน ชัลฟีเนต) (PSS) เพื่อเคลือบบนท่อคาร์บอนนาโนที่ผ่านการทำดัดแปรทางเคมี เป็นที่น่าสนใจว่า การเตรียมพิล์มบางชั้นที่ 1, 2 และ 3 บนท่อคาร์บอนนาโนที่ผ่านการทำดัดแปรทางเคมี สามารถเตรียมได้ด้วยวิธีที่ง่ายโดยใช้ “just enough polyelectrolyte” ซึ่งปราศจากการผ่านชั้นของการบีบเนื้อง ทำให้ท่อคาร์บอนนาโนดั้งหลาวยั่นที่มีความเสถียรอยู่ในสารละลายสูง ท่อคาร์บอนนาโนดั้งหลาวยั่นที่มีความเสถียรของท่อคาร์บอนนาโนดั้งหลาวยั่นได้โดยการใช้ load ยานโนเดลที่ละเอียด ได้แก่ ยาเจนเรียนไวโอล็อกและยาไดโคลฟีแนค พนวณว่า ยาเจนเรียนไวโอล็อกสามารถ load ลงบนท่อคาร์บอนนาโนดั้งหลาวยั่นที่มีพื้นผิวเป็นปะจุสบได้ดี ในขณะที่ยาไดโคลฟีแนคไม่สามารถ load ลงบนท่อคาร์บอนนาโนดั้งหลาวยั่นได้เลย ทั้งนี้ท่อคาร์บอนนาโนที่ผ่านการทำดัดแปรด้วยหมู่ดั้ดแปรที่แตกต่างกันถูกนำมาทดสอบเพื่อประเมินค่าความเป็นพิษต่อเซลล์ L929 ด้วย MTT assay พบว่า ท่อคาร์บอนนาโนที่ผ่านการทำดัดแปรทางเคมี ประมาณ 25 ไมโครกรัม/มิลลิลิตร มีความเป็นพิษต่อเซลล์ L929 ในขณะที่ท่อคาร์บอนนาโนดั้งหลาวยั่นที่มี PDADMAC มีความเป็นพิษต่อเซลล์ L929 ที่ปริมาณ 12.5 ไมโครกรัม/มิลลิลิตร

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Noncovalent surface modification of multiwall carbon nanotubes (MWCNTs) using chitosan coating as a monolayer is presented. The different degree of deacetylation (%DD) (61%, 71%, 78%, 84%, 90% and 93%) of chitosan and chitosan concentration affected to carbon nanotubes's dispersion efficiency and their stability. Results showed that the dispersion of MWCNTs could be improved when using chitosan with the lowest degree of deacetylation (61%DD). Zeta potential measurements confirmed that the chitosan surface coverage on the MWCNTs was twice as high when modifying the nanotubes surface with the 61%DD than when using the 93%DD chitosan. To study how a chitosan-polysaccharide increases the dispersion efficiency and stability of CNTs, molecular dynamics simulation done on the three models: i) two pristine CNTs (*p*CNT-*p*CNT), ii) a pristine CNT-a chitosan wrapped CNT (*p*CNT-*cw*CNT) and iii) two chitosan wrapped CNTs (*cw*CNT-*cw*CNT). As a result, the CNT aggregation was found in *p*CNT-*p*CNT and *p*CNT-*cw*CNT due to van der Waals interaction between the tube-tube aromatic rings, and intertube bridging by chitosan, respectively. In case of *cw*CNT-*cw*CNT, charge-charge repulsion was found to separate the two tubes and well disperse in aqueous solution. Although the monolayer coating MWCNTs with low %DD chitosan was improved the dispersion of MWCNTs, their stability was insufficient to be used as drug carrier. Therefore, the layer-by-layer deposition technique was selected as a potential method for preparing multilayers between poly(diallyldimethylammonium chloride) (PDADMAC) and poly(sodium 4-styrenesulfonate) (PSS) coating on treated MWCNTs surface. Interestingly, we found the simple method to prepare primary, secondary and tertiary layers on treated MWCNT with "just enough polyelectrolyte" without centrifugation process. Multilayer coating on MWCNT was provided high stability in aqueous solution. Gentian violet and diclofenac were used as hydrophilic model drugs for loading on modified MWCNTs. Gentian violet selectively loaded on negatively charged surface of MWCNT while diclofenac cannot be achieved to load in any kind of MWCNTs. The cytotoxicity of modified MWCNTs with different functional groups was evaluated by MTT assay. Treated MWCNTs were toxic to L929 cells when the concentration reached 25 μ g/ml while primary coating MWCNTs with PDADMAC was toxic at concentration 12.5 μ g/ml.

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LIST OF ABBREVIATIONS

CNT(s)	:	Carbon nanotube(s)
MWCNT(s)	:	Multiwall carbon nanotube(s)
SWCNT(s)	:	Singlewall carbon nanotube(s)
DD	:	Degree of deacetylation
PEL	:	Polyelectrolyte
PEM	:	Polyelectrolyte Multilayer Thin Films
LbL	:	Layer-by-Layer technique
ESA	:	Electrostatic self - assembly
PDADMAC	:	Poly(diallyldimethylammonium chloride)
PSS	:	Poly(sodium 4-styrene sulfonate)
RDF	:	Radial Distribution Function
τ	:	Torsion angle
Sg	:	Distance between CNTs surface
Cg	:	Distance between center of mass of CNTs
Cw	:	Chitosan wrapped
p	:	Pristine
mM	:	milli Molar
M	:	Molar
nm	:	nanometer
A°	:	Angstrom
min	:	minute
ps	:	picosecond
fs	:	femtosecond
<i>et al.</i>	:	<i>et alii</i>