

REFERENCES

- [1] Euvananont, C., Junin, C., Inpor, K., Limthongkul, P., and Thanachayanont, C. TiO₂ optical coating layers for self-cleaning applications. Ceram. Inter. 34 (2008): 1067-1071.
- [2] Numpud, P., Charinpanitkul, T., and Tanthapanichakoon, W. Photoinduced hydrophilic property of zinc oxide thin films prepared by sol-gel dip coating method. J. Ceram. Soc. Japan. 116 (2008): 414-417.
- [3] Nimittrakoolchai, O-U., and Supothina, S. Deposition of organic-based superhydrophobic films for anti-adhesion and self-cleaning applications. J. Eur. Ceram. Soc. 28 (2008): 947-952.
- [4] Li, Y., et al. Superhydrophobicity of 2D ZnO ordered pore arrays formed by solution-dipping template method. J. Colloid Interf. Sci. 287 (2005): 634-639.
- [5] Ma, M., and Hill, R.M. Superhydrophobic surfaces. Curr. Opin. Colloid In. 11 (2006): 193-202.
- [6] Kenanakis, G., et al. Light-induced reversible hydrophilicity of ZnO structures grown by aqueous chemical growth. Appl. Surf. Sci. 254 (2008): 5695-5699.
- [7] Watanabe, T., et al. Photocatalytic activity and photoinduced hydrophilicity of titanium dioxide coated glass. Thin Solid Films. 351 (1999): 260-263.
- [8] Guo, M., Diao, P., and Cai, S. Highly hydrophilic and superhydrophobic ZnO nanorod array films. Thin Solid Films. 515 (2007): 7162-7166.
- [9] Ding, B., Ogawa, T., Kim, J., Fujimoto, K., and Shiratori, S. Fabrication of a superhydrophobic nanofibrous zinc oxide film surface by electrospinning. Thin Solid Films. 516 (2008): 2495-2501.

- [10] Wang, C., et al. Nucleation and growth of ZnO nanocrystals in polymer films. Thin Solid Films. 516 (2008): 6058-6062.
- [11] Kenanakis, G., Vernardou, D., Koudoumas, E., and Katsarakis, N. Growth of *c*-axis oriented ZnO nanowires from aqueous solution: The decisive role of a seed layer for controlling the wires' diameter. J. Cryst. Growth. 311 (2009): 4799-4804.
- [12] Zhang, J., Feng, H., Hao, W., and Wang, T. Blue-emitting ZnO sol and film obtained by sol-gel process. J. Sol-Gel Sci. Technol. 39 (2006): 37-39.
- [13] Li, P., Wei, Y., Liu, H., and Wang, X. Growth of well-defined ZnO microparticles with additives from aqueous solution. J. Solid State Chem. 178 (2005): 855-860.
- [14] Zhao, J., Jin, Z.-G., Liu, X.-X., and Jin, Z.-G. Growth and morphology of ZnO nanorods prepared from Zn(NO₃)₂/NaOH solutions. J. Eur. Ceram. Soc. 26 (2006): 3745-3752.
- [15] Vernardou, D., et al. The effect of growth time on the morphology of ZnO structures deposited on Si (100) by the aqueous chemical growth technique. J. Cryst. Growth. 308 (2007): 105-109.
- [16] Etienne, O., et al. Polyelectrolyte multilayer film coating and stability at the surfaces of oral prosthesis base polymers: an *in vitro* and *in vivo* study. J Dent Res. 85 (2006): 44-48.
- [17] Callewaert, M., Gohy, J.-F., Dupont-Gillain, C.C., Boulangé-Petermann, L., and Rouxhet, P.G. Surface morphology and wetting properties of surfaces coated with an amphiphilic diblock copolymer. Surf. Sci. 575 (2005): 125-135.
- [18] Silva, T.H., Barreira, S.V.P., Moura, C., and Silva, F. Electrochemical characterization of a self-assembled polyelectrolyte film. Portugaliae electrochemica acta. 21 (2003): 281-292.

- [19] Zhai, L., Cebeci, F.Ç, Cohen, R.E., and Rubner, M.F. Stable superhydrophobic coatings from polyelectrolyte multilayers. Nano Lett. 4 (2004): 1349-1353.
- [20] Tang, Z., Donohoe, S.T., Robinson, J.M., Chiarelli, P.A., and Wang, H.L. Film formation, surface character, and relative density for electrochromic PEI/(PSS:PEDOT) Multilayered thin films. Polymer. 46 (2005): 9043-9052.
- [21] Zhu, Y., Shi, M., Wu, X., and Yang, S. Amphiphilic copolymer grafted “smart surface” enhanced by surface roughness. J. Colloid Interf. Sci. 315 (2007): 580-587.
- [22] Liu, Z., Li, J., Ya, J., Xin, Y., and Jin, Z. Mechanism and characteristic of porous ZnO films by sol-gel method with PEG template. Mater. Lett. 62 (2008): 1190-1193.
- [23] Dai, X., Zhang, Y., Guan, Y., Yang, S., and Xu, J. Mechanical properties of polyelectrolyte multilayer self-assembled films. Thin Solid Films. 474 (2005): 159-164.
- [24] Shao, L., and Lutkenhaus, J.L. Thermochemical properties of free-standing electrostatic layer-by-layer assemblies containing poly(allylamine hydrochloride) and poly(acrylic acid). Soft matter. 6 (2010): 3363-3369.
- [25] Iliescu, C., Chen, B., and Miao, J. On the wet etching of Pyrex glass. Sens. Actuator A. 143 (2008): 154-161.
- [26] Houn, B., Huang, C.-L., and Tsai, S.-Y. Effect of the pH on the growth and properties of sol-gel derived boron-doped ZnO transparent conducting thin film. J. Cryst. Growth. 307 (2007): 328-333.
- [27] Ayouchi, R., Martin, F., Leinen, D., and Ramos-Barrado, J.R. Growth of pure ZnO thin films prepared by chemical spray pyrolysis on silicon. J. Cryst. Growth. 247 (2003): 497-504.

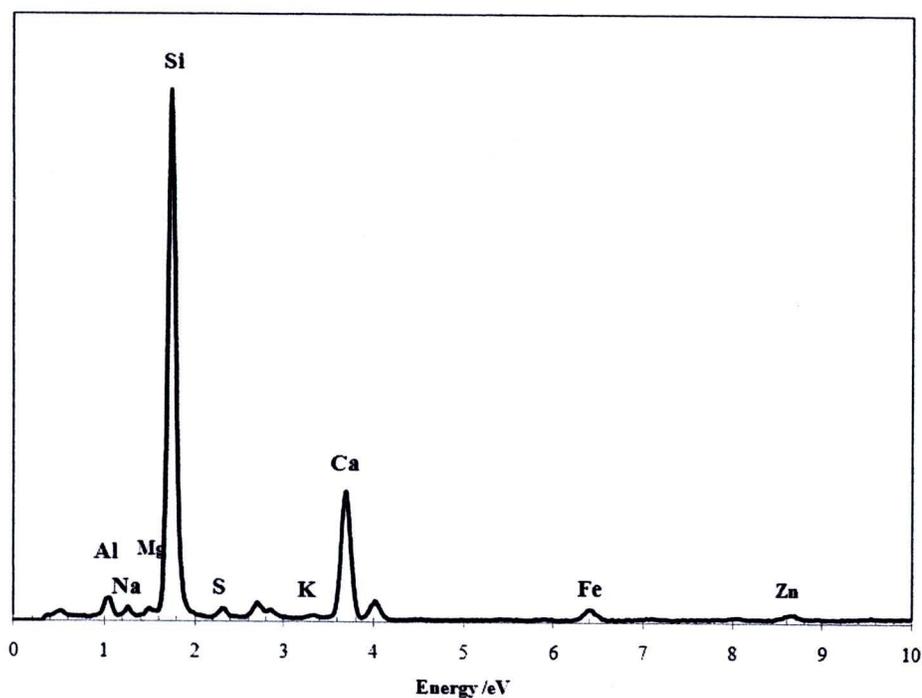
- [28] Li, Y., Xu, L., Li, X., Shen, X., and Wang, A. Effect of aging time of ZnO sol on the structural and optical properties of ZnO thin films prepared by sol-gel method. Appl. Surf. Sci. 256 (2010): 4543-4547.
- [29] Karuppuchamy, S., and Ito, S. Cathodic electrodeposition of nanoporous ZnO thin films from new electrochemical bath and their photoinduced hydrophilic properties. Vacuum. 82 (2008): 547-550.
- [30] ZnO [Online]. Available from: http://en.wikipedia.org/wiki/Zinc_oxide. [2008, September 30].
- [31] Rani, S., Suri, P., Shishodia, P.K., and Mehra, R.M. Synthesis of nanocrystalline ZnO powder via sol-gel route for dye-sensitized solar cells. Solar Energy Mater Solar Cells. 92 (2008): 1639-1645.
- [32] Zhou, X., Guo, X., Ding, W., and Chen, Y. Superhydrophobic or superhydrophilic surfaces regulated by micro-nano structured ZnO powders. Appl. Surf. Sci. 255 (2008): 3371-3374.
- [33] Berg, J.C. Wettability. Washington: Washington University Press. 1993.
- [34] Numpud, P. Photoinduced Hydrophilic property of zinc oxide thin films prepared by sol-gel dip coating method. Master's Thesis. Department of Chemical Engineering, Faculty of Engineering, Chulalongkorn University, 2008.
- [35] Yimsiri, P., and Mackley, M.R. Spin and dip coating of light-emitting polymer solutions: matching experiment with modelling. Chem. Eng. Sci. 61 (2006): 3496-3505.
- [36] Rheometer [Online]. Available from: <http://en.wikipedia.org/wiki/Rheometer>. [2010, August 18].

- [37] Djouadi, D., Chelouche, A., Aksas, A., and Sebais, M. Optical properties of ZnO/silica nanocomposites prepared by sol-gel method and deposited by dip-coating technique. Physics procedia. 2 (2009): 701-705.
- [38] Confocal laser scanning microscopy (CLSM) [Online]. Available from: http://en.wikipedia.org/wiki/Confocal_laser_scanning_microscope. [2010, June 7].
- [39] Evans Analytical Group (EAG). Fourier transform infrared spectroscopy (FTIR) [Online]. Available from: http://www.eaglabs.com/techniques/analytical_techniques/ftir.php. [2010, June 7].
- [40] Scanning electron microscope (SEM) [Online]. Available from: http://en.wikipedia.org/wiki/Scanning_electron_microscope. [2010, June 7].
- [41] Energy-dispersive X-ray spectroscopy (EDX) [Online]. Available from: http://en.wikipedia.org/wiki/Energy-dispersive_X-ray_spectroscopy. [2010, June 7].
- [42] Shiratori, S.S., and Rubner, M.F. pH-Dependent thickness behavior of sequentially adsorbed layers of weak polyelectrolytes. Macromolecules. 33 (2000): 4213-4219.
- [43] Landau, L., and Levich, B. Dragging of a liquid by a moving plate. Acta Physicochimica U.R.S.S. 17 (1-2) (1942): 42-54.
- [44] Dorrer, C., and Ruhe, J. Drops on Microstructured surfaces coated with hydrophilic polymer: Wenzel's model and Beyond. Langmuir. 24 (2008): 1959-1964.
- [45] Wu, H.-S., Jone, H.-C., and Hwang, J.-W. Reaction of polyacrylic acid and metal oxides: Infrared spectroscopic kinetic study and solvent effect. J. Appl. Polym. Sci. 63(1997): 89-101.

- [46] Liu, H., Feng, L., Zhai, J., Jiang, L., and Zhu, D. Reversible wettability of a chemical vapor deposition prepared ZnO film between superhydrophobicity and superhydrophilicity. Langmuir. 20 (2004): 5659-5661.
- [47] Miyauchi, M., et al. Reversible wettability control TiO₂ surface by light irradiation. Surf. Sci. 511 (2002): 401-407.
- [48] Liu, Z., Jin, Z., Li, W., and Qiu, J. Preparation of ZnO porous thin films by sol-gel method using PEG template. Mater. Lett. 59 (2005): 3620-3625.
- [49] Oyama, M., Kozuka, H. and Yoko, T. Sol-gel preparation of ZnO films with extremely preferred orientation along (002) plane from zinc acetate solution. Thin Solid Films. 306 (1997): 78-85.
- [50] Znaidi, L., et al. Oriented ZnO thin films synthesis by sol-gel process for laser application. Thin Solid Films. 428 (2003): 257-262.
- [51] Sun, R.-D., Nakajima, A., Fujishima, A., Watanabe, T., and Hashimoto, K. Photoinduced surface wettability conversion of ZnO and TiO₂ thin films. J. Phys. Chem. B. 105 (2001): 1984-1990.
- [52] Brinker, C.J., Frye, G.C., Hurd, A.J., and Ashley, C.S. Fundamentals of sol-gel dip coating. Thin Solid Films. 201 (1991): 97-108.
- [53] Terrier, C., Chatelon, J.P., and Roger, J.A. Electrical and optical properties of Sb: SnO₂ thin films obtained by the sol-gel method. Thin Solid Films. 295 (1997): 95-100.

APPENDIX

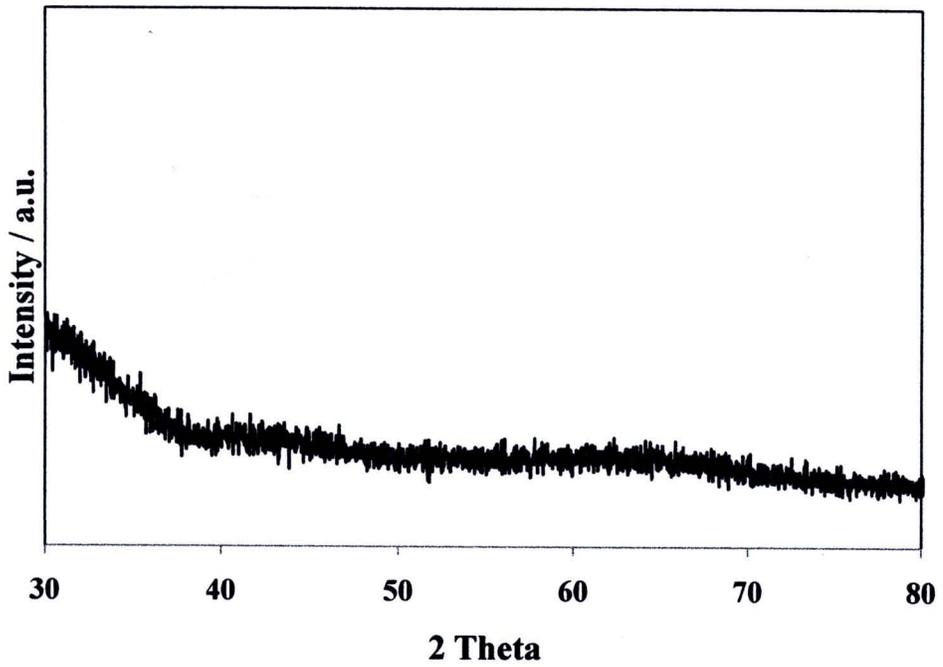
Appendix 1 EDX analysis of ZnO film prepared by 0.1 M $\text{Zn}(\text{NO}_3)_2$ with 3 precipitation cycles coated at withdrawal speed of 3.0 cm/min and drying temperature of 180°C for 2 h



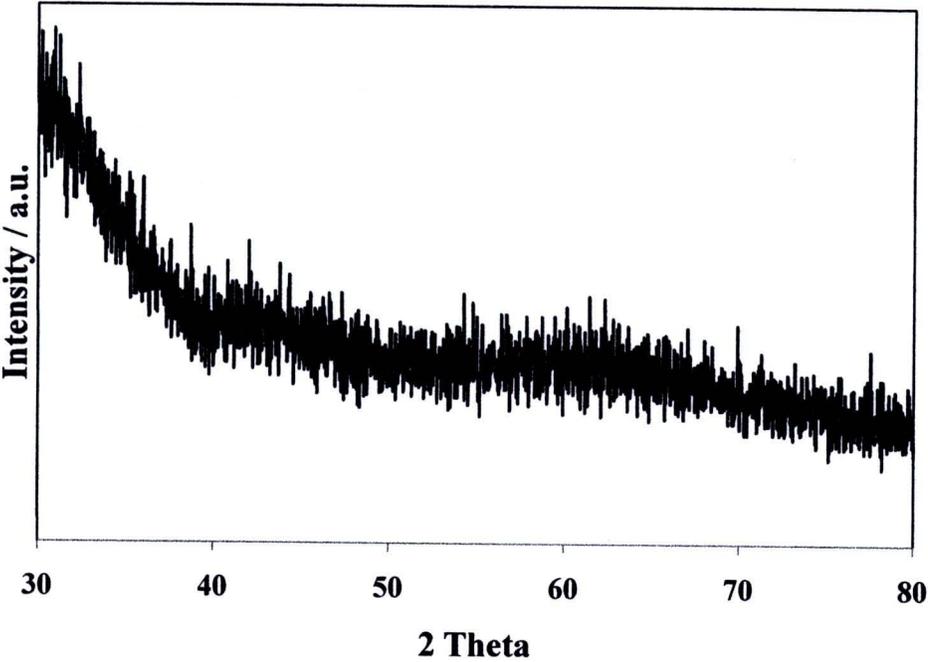
Appendix 2 Summary of the oxide compounds on the ZnO film (prepared by 0.1 M $\text{Zn}(\text{NO}_3)_2$ with 3 precipitation cycles coated at withdrawal speed of 3.0 cm/min and drying temperature at 180°C for 2 h) observed by EDX

Oxide compound	Weight percent
SiO_2	71.21
SO_3	0.80
K_2O	0.20
CaO	7.11
Na_2O	14.84
MgO	2.28
ZnO	0.02

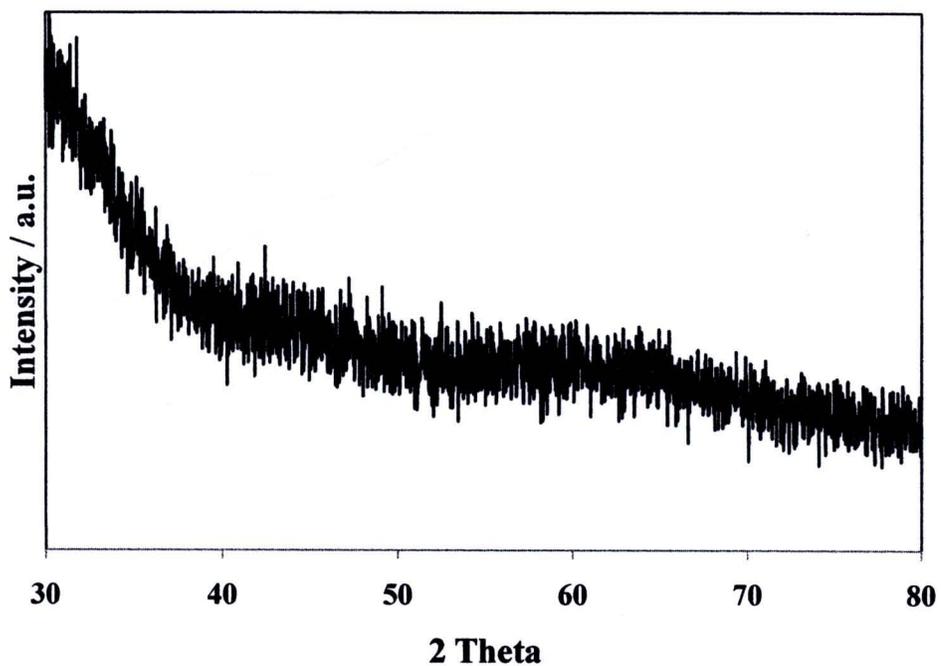
Appendix 3 X-ray diffractograms of ZnO incorporated with polyelectrolyte film prepared by precipitation method with 3 precipitation cycles and drying temperature of 180°C



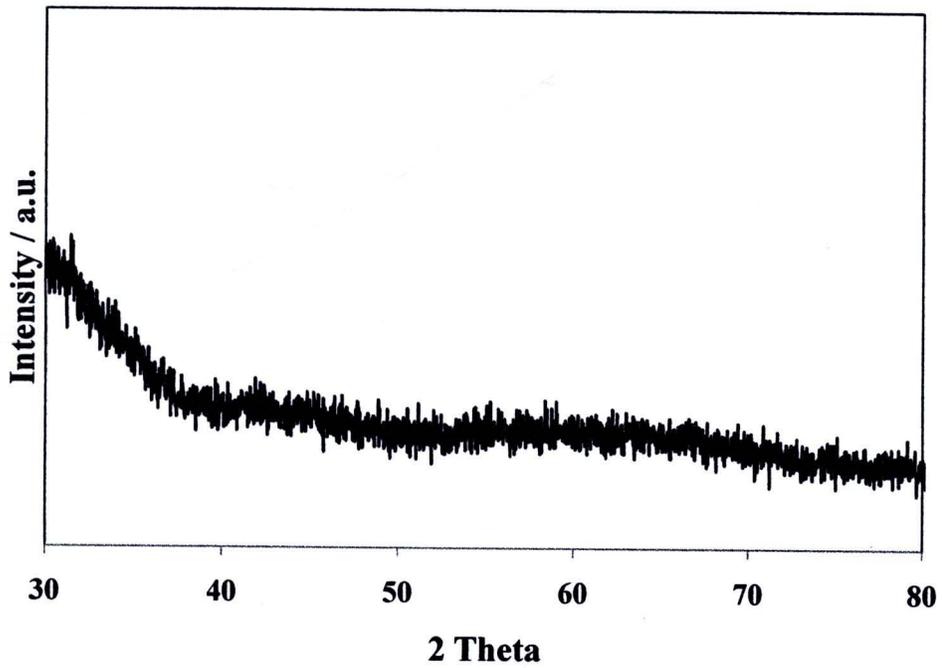
Appendix 4 X-ray diffractograms of ZnO incorporated with polyelectrolyte film prepared by 0.1 wt% ZnO in poly(acrylic acid) (PAA) solution with drying temperature of 180°C



Appendix 5 X-ray diffractograms of ZnO incorporated with polyelectrolyte film prepared by 0.2 wt% ZnO in poly(acrylic acid) (PAA) solution with drying temperature of 180°C



Appendix 6 X-ray diffractograms of ZnO incorporated with polyelectrolyte film prepared by sol-gel method as prepared by pH of precursor sol of 1 with calcination temperature of 500°C





VITA

Miss Walaiporn Suthabanditpong was born on April 23, 1984 in Bangkok, Thailand. She graduated from a secondary school of Mahidolwittayanusorn School in 2001. In 2006, she received a Bachelor degree in Chemical Engineering from King Mongkut's University of Technology Thonburi and continued her study in Chemical Engineering for the Master degree at the Department of Chemical Engineering, Faculty of Engineering, Chulalongkorn University. During the master program, she conducted a research entitled "Deposition of nano-scaled zinc oxide on polyelectrolyte film by layer-by-layer technique" under supervision of Assoc. Prof. Tawatchai Charinpanitkul, the head of Center of Excellence in Particle Technology (CEPT). As a part of her master program, she received financial support from Japan Student Service Organization (JASSO, Japan) to work as a research student at Nagoya Institute of Technology (NIT, Japan) for 10 months. She received the Master degree of Chemical Engineering in 2010.

International Proceeding

1. W. Suthabanditpong, K. Faungnawakij, N. Viriya-empikul, A. Sootitantawat and T. Charinpanitkul. "Deposition of nano-scaled zinc oxide on polyelectrolyte film by layer-by-layer technique" *Proceedings of International Conference on 6th Asian Aerosol Conference (AAC09)*, November 24-27, 2009, Bangkok, Thailand, p.69.
2. W. Suthabanditpong, H. Watanabe, T. Shirai, K. Faungnawakij, N. Viriya-empikul, A. Sootitantawat, T. Charinpanitkul and M. Fuji. "Deposition of nano-scaled zinc oxide on polyelectrolyte film by layer-by-layer technique" *Proceedings of International Conference on 3rd Thailand-Japan international Conference (TJIA2010)*, November 19, 2010, Nagoya, Japan, p.72-73.

