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ตำแหน่งทางวิชาการ: รองศาสตราจารย์ ระดับ 9

สถานที่ทำงาน: วิทยาลัยนาโนเทคโนโลยีพระจอมเกล้าลาดกระบัง

สถาบันเทคโนโลยีพระจอมเกล้าเจ้าคุณทหารลาดกระบัง

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ประวัติการรับราชการ

ได้รับแต่งตั้งให้ดำรงตำแหน่งอาจารย์

เมื่อวันที่ 19 พฤษภาคม พ.ศ. 2536

ได้รับแต่งตั้งให้ดำรงตำแหน่งผู้ช่วยศาสตราจารย์

เมื่อวันที่ 24 เมษายน พ.ศ. 2545

ได้รับแต่งตั้งให้ดำรงตำแหน่งรองศาสตราจารย์

เมื่อวันที่ 31 สิงหาคม พ.ศ. 2548

ทุนและรางวัล

- พ.ศ. 2531-35 ทุนโครงการพัฒนาและส่งเสริมผู้มีความสามารถพิเศษทางวิทยาศาสตร์และเทคโนโลยี (พสวท) ของ สวท ในระดับปริญญาตรี สาขาฟิสิกส์ คณะวิทยาศาสตร์ มหาวิทยาลัยเชียงใหม่
- พ.ศ. 2532 เหรียญผลการเรียนดีเด่น คณะวิทยาศาสตร์ มหาวิทยาลัยเชียงใหม่
- พ.ศ. 2536-40 ทุนกระทรวงวิทยาศาสตร์และเทคโนโลยี ในระดับปริญญาโท เพื่อศึกษาต่อในต่างประเทศ สาขาฟิสิกส์ Faculty of Art and Science, University of Central Florida, United State of America

ประสบการณ์การทำงาน:

- พ.ศ. 2536-52 อาจารย์ประจำ ภาควิชาฟิสิกส์ประยุกต์ คณะวิทยาศาสตร์ สถาบันเทคโนโลยีพระจอมเกล้าเจ้าคุณทหารลาดกระบัง
- พ.ศ. 2542-44 อาจารย์พิเศษ ภาควิชาฟิสิกส์ คณะวิทยาศาสตร์ มหาวิทยาลัยอุบลราชธานี
- พ.ศ. 2546 อาจารย์พิเศษ หลักสูตรปริญญาโท-เอก สาขาวิชาฟิสิกส์ คณะวิทยาศาสตร์ มหาวิทยาลัยมหิดล
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- พ.ศ. 2552 รักษาการ รองคณบดีฝ่ายวิชาการ คณะวิทยาศาสตร์ สถาบันเทคโนโลยีพระจอมเกล้าเจ้าคุณทหาร ลาดกระบัง
- พ.ศ. 2552 รักษาการผู้อำนวยการสำนักบริหารงานทั่วไป คณะวิทยาศาสตร์ สถาบันเทคโนโลยีพระจอมเกล้าเจ้าคุณทหารลาดกระบัง
- พ.ศ. 2552 รักษาการผู้อำนวยการศูนย์เครื่องมือวิทยาศาสตร์ คณะวิทยาศาสตร์ สถาบันเทคโนโลยีพระจอมเกล้าเจ้าคุณทหารลาดกระบัง
- พ.ศ. 2552-ปัจจุบัน อาจารย์ประจำ สาขานาโนวิทยาและนาโนเทคโนโลยี คณะวิทยาศาสตร์ สถาบันเทคโนโลยีพระจอมเกล้าเจ้าคุณทหารลาดกระบัง

หัวข้อวิจัยและพัฒนา :

ปี	ชื่องานวิจัย	แหล่งทุน
2554-2556	การสังเคราะห์สารประกอบออกไซด์เชิงซ้อนที่มีอนุภาคระดับนาโนเมตรด้วยกระบวนการใช้คลื่นเสียง	สถาบันเทคโนโลยีพระจอมเกล้าเจ้าคุณทหารลาดกระบัง
2553	เซ็นเซอร์จากวัสดุผสมไฮบริดนาโนสำหรับตรวจสอบสารพิษโลหะหนักในสิ่งแวดล้อม	ศูนย์ความเป็นเลิศทางฟิสิกส์
2552	การพัฒนาวัสดุคอมโพสิตของโลหะออกไซด์กับวัสดุโครงสร้างนาโนเพื่อใช้ในงานด้านอิเล็กทรอนิกส์	ศูนย์ความเป็นเลิศทางฟิสิกส์
2549-2551	การพัฒนาระบบการปลูกฟิล์มบางนำไฟฟ้าโปร่งใสและระบบวัดทางแสงเพื่อพิสูจน์เอกลักษณ์สมบัติฟิล์มบาง	สำนักวิจัยนาโนเทคโนโลยีลาดกระบัง
2548-2549	การพัฒนาระบบต้นแบบลอจิกเกตทางแสงโดยใช้ฟิล์มบางสารอินทรีย์	สจล.

- **หัวข้องานวิจัยและพัฒนาที่สนใจ:** วัสดุพลังงาน, วัสดุทางแสง นาโนเทคโนโลยี เทคโนโลยีฟิล์มบาง

ผลงานวิจัย (2005-2012)

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8. R. Noonuruk, W. Techitdheera, W. Pecharapa, "Characterization and ozone-induced coloration of Zn_xNi_{1-x}O thin films prepared by sol-gel method", *Thin Solid Films*, 520, (2012), pp. 2769-2775, (IF: 1.909)
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10. R. Noonuruk, N. Wongpisutpaisan, P. Mukdacharoenchai, W. Techitdheera and W. Pecharapa, "Ozone-induced Optical Density Change of NiO Thin Films and Their Applicability as Neutral Optical Density Filter", *Procedia Engineering*, 8 (2011), pp. 212-216.
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18. W. Mekprasart, N. Vittayakorn and W. Pecharapa, "Ball-milled CuPc/ TiO_2 hybrid nanocomposite and its photocatalytic degradation of aqueous Rhodamine B", *Materials Research Bulletin*, 47, Issue 11, (2012), pp. 3114-3119 (IF: 2.105)
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1. W. Techittheera, P. Thanomngam, W. Pecharapa, and J. Nukeaw, “ $\text{Al}_{0.3}\text{Ga}_{0.7}\text{As}/\text{GaAs}$ single quantum well transition energy calculation”, *KMITL SCIENCE JOURNAL*, Vol. 5, No. 1, pp. 445-449, 2005. (TJIF:0.045)

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ผลงานวิชาการอื่น ๆ (เช่น **Proceeding** ตำรา สัทธิบัตร ฯลฯ) ประชุมวิชาการ

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ภาคผนวก

งานประชุมวิชาการวิทยาศาสตร์และเทคโนโลยีแห่งประเทศไทย ครั้งที่ 38 (วทท. 38)

ระหว่างวันที่ 17-19 ตุลาคม 2555

ณ จังหวัดเชียงใหม่



Facile Synthesis of CuO Nanoparticles by Precipitation Method

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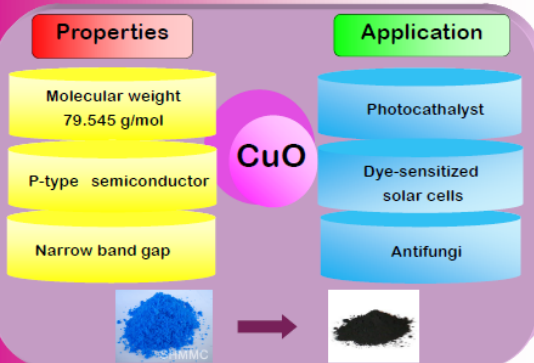
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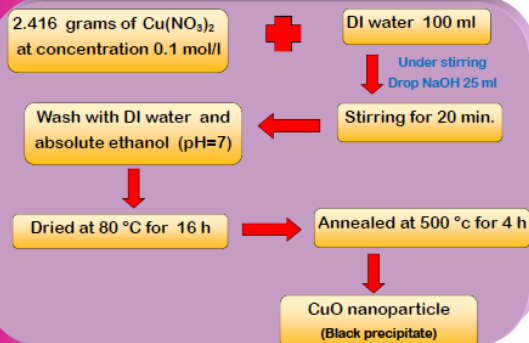
Abstract

CuO nanoparticles were synthesized by precipitation method using copper nitrate $\text{Cu}(\text{NO}_3)_2$ as starting precursor combination with post-heating. Relevant properties of as-synthesized nanoparticles were investigated by X-ray diffraction, Scanning electron microscope and Fourier transform infrared spectroscopy. XRD result suggests that CuO nanoparticles in cubic structure are obtained by this synthesized process. Meanwhile, SEM result reveals that surface morphologies and structure of as-prepared CuO are in short nanorod-like structure with average size of few hundred nanometer range. FTIR results suggest that the formation of CuO nanostructure and Cu-O bonding were obtained by this simple process.

Introduction



Experiment



Acknowledgements

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Results & Discussion

XRD, SEM and FTIR Characterization

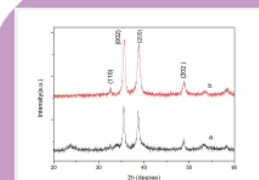


Fig. 1. XRD pattern of CuO nanostructure from $\text{Cu}(\text{NO}_3)_2$. (a) as-synthesized and (b) after calcination of CuO.

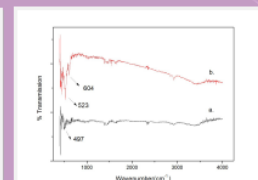


Fig. 2. FTIR spectra of CuO nanostructure from $\text{Cu}(\text{NO}_3)_2$. (a) as-synthesized and (b) after calcination of CuO.

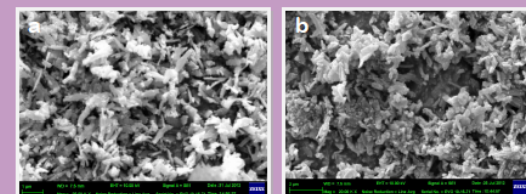


Fig. 3. SEM images of CuO nanostructure from $\text{Cu}(\text{NO}_3)_2$. (a) as-synthesized and (b) after calcination of CuO.

Conclusion

- CuO nanostructures were successfully synthesized by precipitation method
- XRD peaks confirm that the formation of CuO from $\text{Cu}(\text{NO}_3)_2$ is in monoclinic phase
- SEM results show that good dispersion in short nanorod-like structure are found in as-synthesized CuO whilst the aggregation of CuO is formed by annealing process.
- FTIR result suggests that the formation of copper oxide compound is obtained after annealing process in this method accompanying the presence of Cu-O bonds in stretch-

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ณ จังหวัดอุบลราชธานี

ได้รับการตีพิมพ์ในวารสารวิชาการ Energy Procedia (Special issue; 10th Eco-Energy and
Materials Science and Engineering Symposium

Synthesis of CuO nanoparticles by precipitation method using different precursors

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Abstract

CuO nanoparticles were synthesized by precipitation method using different precursors as copper nitrate (Cu(NO₃)₂) and copper chloride (CuCl₂) with post-heating comparing between as-synthesized and after calcinations. Relevant properties of as-synthesized nanoparticles were investigated by X-ray diffraction (XRD), scanning electron microscope (SEM) and Fourier Transform Infrared (FTIR) Spectroscopy. Overall results suggest that the formation of CuO nanostructures with different shape, size and morphology can be achieved using different precursors via this process. The improvement in their crystallinity and purification can be further attained by post calcinations process.

Introduction

Copper oxide (CuO) is one of potential p-type semiconductors and gains considerable attentions due to its excellent optical, electrical, physical, and magnetic properties. CuO with narrow band gap of 1.2 eV is extensively used in various applications such as catalysis [1], solar energy conversion [2], gas sensor [3] and field emission [4]. However, these novel properties can be improved by synthesis in CuO nanostructures that shown excellent performance comparing to bulk counterpart. Different nanostructures of CuO are synthesized in form of nanowire, nanorod, nanoneedle, nano-flower and nanoparticle. In the past decades, various methods have been proposed to produce CuO nanoparticles with different sizes and shapes such as thermal oxidation [5], sonochemical [6], combustion [7] and quick-precipitation [8-9]. Among these processes, precipitation method is a facile way which attracts considerable interest in industries because of low energy and temperature, inexpensive and cost-effective approach for large scale production and good yield.

In the present work, the main objective is to investigate the effect of starting precursors on structural properties of CuO nanostructures synthesized via precipitation method and annealing process. Copper nitrate and copper chloride was chosen as starting precursors. The as-prepared precipitates were analyzed by scanning electron microscopy, X-ray diffractometer and Fourier Transform Infrared Spectroscopy.

Material & Method

copper chloride (CuCl₂·2H₂O) or copper nitrate (Cu(NO₃)₂·6H₂O)

Dispersed with DI water 100 ml

Mixed solution under stirring for 20 min

Group filter and filtrate

Wash with DI water and absolute ethanol (pH=7)

Dried at 80 °C for 16 h and Annealed at 500 °C for 4 h

CuO nanostructures (after precipitation)

SEM Characterization (continue)

Fig.3 SEM images of CuO nanostructure prepared from Cu(NO₃)₂ and CuCl₂
(a) as-synthesized of CuO from Cu(NO₃)₂ (b) after calcination of CuO from Cu(NO₃)₂
(c) as-synthesized of CuO from CuCl₂ (d) after calcination of CuO from CuCl₂

Conclusion

In summary, CuO nanostructures were successfully synthesized by precipitation method using different precursors including copper nitrate (Cu(NO₃)₂) and copper chloride (CuCl₂) with post-heating treatment. XRD and FTIR results suggest that the better formation of CuO nanostructures can be attained by single-step precipitation of Cu(NO₃)₂ precursor without annealing treatment. Calcinations process can effectively remove residue and lead to the better crystallization of CuO. Furthermore, it is notified that different precursors used in the precipitation process has strong influence on shape, size and morphology of CuO-nanostructures.

Results & Discussion

XRD, FTIR and SEM Characterization

Fig.1 XRD pattern of CuO nanostructure from Cu(NO₃)₂ and CuCl₂

Fig.2 FTIR spectra of CuO nanostructure from Cu(NO₃)₂ and CuCl₂

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**Study of Antifungal Activities of CuO/ZnO Nanocomposites Synthesized
by Co-precipitation Method**

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Keywords: CuO/ZnO, nanocomposites, co-precipitation method, antifungal activities

Abstract. In this work, series of CuO/ZnO functional nanocomposites were synthesized through co-precipitation method using CuCl₂·H₂O and ZnCl₂ as starting materials with various molar ratio of copper:zinc, followed by annealing process at 600 °C for 2 hours to obtain CuO/ZnO nanocomposites. The structures of the composites were analyzed using X-ray diffraction and field emission scanning electron microscopy. For XRD result, diffraction peaks of the composites reveal the well-crystalline characteristic indicating the mixture phase of CuO and ZnO. SEM results show different morphologies of CuO, ZnO and Cu-Zn oxide nanocomposites appearing in quasi spherical structure. The composites were used for antifungal activity via agar disk diffusion method. It is found that the composite with certain ratio of Cu:Zn exhibits superiority in partial inhibition of strain *Aspergillus flavus* *Trichoderma* comparing to either pure CuO or ZnO.

Introduction

At the present time, nanoparticles inward to the size range of 100 nm have attracted great interest due to their rather high surface to volume ratio with unique morphologies, leading to practical utilizations in various potential applications of science and technology. Among them, metal oxide nanoparticles such as CuO, ZnO, TiO₂ and MgO are well recognized because of their good physical properties suitable for widespread applications including optoelectronic, semiconductor technology and fungal inhibition [1-4]. Copper oxide has been extensively utilized as antifungal agent accompanying low cost, abundant resource, simple preparation [5]. However, some fungi strain cannot be inhibited by CuO. Therefore, composite form with compatible elements or compounds is one of effective method for improving CuO functionality. It is known that ZnO has been comprehensively advertent owing to its fascinating properties such as high exciton binding energy of 60 meV, strong chemical-thermal stability and most of all, its ability for resisting microbial activity [6]. In consequence, the advantage of CuO coupled with ZnO is easy incorporation, improvement of antifungal properties and similar physical-chemical properties comparing to other metal oxide materials [7]. Moreover, both material are rather cheap and environmental friendly. CuO/ZnO composites with different sizes and shapes can be synthesized various methods such as electrodeposition [8], hydrothermal process [9], electrospinning [10] and co-precipitation [11]. Among these mentioned processes, precipitation method is a facile way which attracts considerable interest because of low temperature process, inexpensive and cost-effective approach for large scale production. The practical applications of CuO-ZnO composite in solar energy conversion applications, photocatalysis and gas-sensing have been recently reported [12-14]. Nevertheless, few literature has been so far focused on the study of antifungal properties of CuO-ZnO nanocomposite. In this study, we reports antifungal property (*Aspergillus flavus* *Trichoderma* agents) of CuO-ZnO nanocomposite synthesized by co-precipitation method with various molar ratio.

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