# DEVELOPMENT OF DRUGS FROM THERAPEUTIC TARGETS IN COBRA VENOM USING STRUCTURE BASED DRUG DESIGN AND VIRTUAL SCREENING

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# A THESIS SUBMITTED IN PARTIAL FULFILLMENT OF THE REQUIREMENTS FOR THE DEGREE OF DOCTOR OF PHILOSOPHY (PHARMACEUTICAL CHEMISTRY AND PHYTOCHEMISTRY) FACULTY OF GRADUATE STUDIES MAHIDOL UNIVERSITY 2008

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Thesis Entitled

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## ACKNOWLEDGEMENTS

I would like to express my deep appreciation to the many people that have made this work possible. First of all, I would like to thank my advisor, Assoc. Prof. Opa Vajragupta, for her encouragement and inspiration throughout my study. She always give me invaluable advice.

To Assist. Prof. Wichet Leelamanit, Assoc. Prof. Penchome Peungvicha and Assoc. Prof. Yuvadee Wongkrajang, I am very grateful to express my gratitude for their helpful advices and cooperation. I would like to thank Dr. Atchara Kaewnoi for providing the rediocides. I wish to thank Assoc. Prof. Leena Suntornsuk and Assist. Prof. Pornchai Rojsitthisak for kindness in providing suggestion and who were examiner of the thesis defense.

I am deeply thank to Prof. Arthur J Olson and Prof. Palmer Taylor for their valuable advice and guidance in this research.

A big thanks goes to all staffs in Prof. Arthur J Olson's laboratory at The Scripps Research Institute and Prof. Palmer Taylor's laboratory at the University of California, San Diago, especially, Dr. Garrett Morris, Dr. Lindy Lindstrom, Dr. Ruth Huey, Dr. Alexandre Gillet, Dr. Rodney Harris, Mrs. Peggy Graber, Dr. Todd Talley and Ms. Joannie Ho for their helpful suggestion and kind help.

A special acknowledgement is extended to Department of Pharmaceutical Chemistry, Faculty of Pharmacy, Mahidol University, for providing research facilities, the Royal Golden Jubilee (RGJ) Project, Thailand Reseach fund, Thai Ministry of University Affairs, Commission on Higher Education, Mahidol University Research fund for providing the opportunity to study in Ph.D. program, supporting the research fund in USA and the presentation at Austria.

As well, my special appreciation to my friend and other persons who have not been mentioned here for their helpful, friendship and encouragement. Finally, I wish to express my deepest gratitude and infinite thankfulness to my family for their love, encouragement and support throughout my life.

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## DEVELOPMENT OF DRUGS FROM THERAPEUTIC TARGETS IN COBRA VENOM USING STRUCTURE BASED DRUG DESIGN AND VIRTUAL SCREENING.

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### ABSTRACT

Alpha-cobratoxin ( $\alpha$ -cbtx), the long chain neurotoxin from *Naja kaouthia* acts on the postsynaptic nicotinic acetylcholine receptors (nAChR) and causes paralysis by preventing acetylcholine binding to nAChR.  $\alpha$ -Cbtx has been proposed as a potential target for anticobratoxin drug design. Investigation of rediocides (A & G) binding mode to reveal the molecular mechanism of its antivenom action was achieved by means of molecular docking. The  $\alpha$ -cbtx/ acetylcholine binding protein (AChBP) complex (1YI5) was selected to construct  $\alpha$ -cbtx active binding site for docking. The constructed  $\alpha$ -cbtx template was validated with the control peptide (Ser182 – Tyr192 of AChBP) and the RMSD was less than 1.2 Å. Rediocides bind to the  $\alpha$ -cbtx in the same location that the  $\alpha$ -cbtx binds to AChBP at the Asp27, Phe29, Arg33, Gly34, Lys35 and Val37 residues. The binding energies were -14.17 and -14.14 kcal/mol. respectively. Thus,  $\alpha$ -cbtx cannot bind to AChBP because some of its binding sites are occupied with rediocides. The template was also employed in the virtual screening of over 1990 compounds in NCI diversity set and 39 hits were identified. The potential hits are NCI42258, NCI121865 and NCI134754 which competitively displace the antagonist (<sup>125</sup>I  $\alpha$ -bungarotoxin) and the agonist (<sup>3</sup>H epibatidine) from their mutually exclusive binding sites on Lymnaea stagnalis (Ls), Aplysia californica (Ac) and Bulinus truncatus (Bt) AChBPs. In particular, NCI121865 had  $K_d$  of 16.26 nM against Ac and 111 nM and 415 nM against Ls and Bt, respectively. The results from SDS-PAGE found that rediocides neutralized and diminished the  $\alpha$ -cbtx band intensity. NCI121865 bound the  $\alpha$ -cbtx in crude venom resulting in the increase of the  $\alpha$ -cbtx band intensity. NCI42258 and NCI134754 changed some characteristics of the  $\alpha$ -cbtx and broadened  $\alpha$ -cbtx band. In the presence of AChBP, it was apparent that three hits and rediocides bound to both  $\alpha$ -cbtx and AChBP. In vivo, three hits (5) mg/kg, i.v.) and rediocides (0.5 mg/kg, i.v.) can prolong the survival times of the mice when injected 30 minutes before injection of the  $\alpha$ -cbtx (3LD<sub>50</sub> dose). Only NCI121865 and NCI134754 demonstrated the antivenom action; they can prolong the survival time when injected immediately after injection of the  $\alpha$ -cbtx. In clinical applications, NCI121865 and NCI134754 would be a very useful potential lead for the treatment of snakebite victims.

# KEY WORDS : ALPHA-COBRATOXIN / ANTICOBRATOXIN / DOCKING / ACETYLCHOLINE / VIRTUAL SCREENING.

170 pp.

การพัฒนาขาจากบริเวณที่ออกฤทธิ์ในพิษงูโดยใช้เทคนิคการออกแบบขาจากโครงสร้างของโมเลกุล เป้าหมายและเทคนิคการคัดกรองประสิทธิภาพสูง (DEVELOPMENT OF DRUGS FROM THERAPEUTIC TARGETS IN COBRA VENOM USING STRUCTURE BASED DRUG DESIGN AND VIRTUAL SCREENING)

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# บทคัดย่อ

แอลฟาโคบราทีอกซินจากงูเห่าไทยเป็นนิวโรทอกซินชนิดสายโซ่ยาว ออกฤทธิ์โดยการจับที่โพสซิน
 แนบติกของอะซิติลโคลีนรีเซ็บเตอร์ทำให้อะซิติลโคลีนไม่สามารถจับกับรีเซ็บเตอร์ได้ เป็นสาเหตุของการเกิด
 อัมพาด การวิจัยนี้เป็นการออกแบบสารด้านพิษงูโดยใช้แอลฟาโคบราทีอกซินเป็นโมเลกุลเป้าหมายและสึกษา
 กลไกการด้านพิษงูของสารเรดิโอไซด์เอและบีจากสมุนไพรโลดทะนงโดยใช้เทคนิคโมเลกุลาร์ด็อกกิง เริ่มจากการ
 เตรียมแม่แบบของแอลฟาโคบราท็อกซินโดยใช้ข้อมูลจากการเอ็กซเรย์ผลึกแอลฟาโคบราท็อกซินที่จับกับอะซิติล
 โกลีนบายดิงโปรตีน (PDB: 1YI5) และใช้เปปไทด์ตำแหน่งที่ 182-192 ของอะซิติลโคลีนบายดิงโปรตีนในการ
 ตรวจสอบความถูกต้องของแม่แบบซึ่งผลการตรวจสอบให้ค่า RMSD น้อยกว่า 1.2 อังสตรอม กลไกการออกฤทธิ์
 ของสารเรดิโอไซด์เอและบีคือ จับกับแอลฟาโคบราทีอกซินที่ตำแหน่ง แอสพาราจีน27 ฟีนิลอะลานีน29 อาร์จินีน
 ใกลซีน34 ไลซีน35 และวาลีน37 ซึ่งเป็นบริเวณเดียวกันกับที่แอลฟาโคบราท์อกซินจับกับบายดิงโปรตีน

สามารถจับกับอะซิติลโกลีนบายดิงโปรตีนเนื่องจากกรดอะมิโนบางด้วของแอลฟาโกบราท็อกซินถูกเรดิโอไซด์ จับอยู่ นอกจากนี้ยังใช้แม่แบบแอลฟาโกบราท็อกซินมากัดกรองสารจำนวน 1990 สารจากสถาบันมะเร็งแห่งชาดิ ได้โกรงสร้างสารที่มีศักยภาพจำนวน 39 สาร และเมื่อนำสารทั้ง 39 สารมาทดสอบในหลอดทดลองได้สารค้นแบบ 3 สารที่มีความสามารถในการแข่งขันกับแอลฟาบังกาโลท็อกซินและสารอีพิบาติดีนที่ติดฉลากกัมมันตรังสีโดย แข่งจับกับอะซิติลโกลีนบายดิงโปรตีนทั้ง 3 ชนิด สาร NCI121865 มีฤทธิ์สูงสุดสามารถจับกับบายดิงโปรตีนจาก Aplysia Lymnaca และ Bolinus ที่ 16.26 111 และ 415 นาโนโมลาร์ ตามลำดับ ผลการศึกษาโดยใช้เจลอิเล็กโตร ฟอรีซิสพบว่า เรดิโอไซด์เอและบีสามารถลดความเข้มของแอลฟาโกบราท็อกซิน NCI121865 เพิ่มความเข้มของ แอลฟาโกบราท็อกซิน NCI42258 และ NCI134754 เปลี่ยนคุณสมบัติแอลฟาโกบราท็อกซิน ในกรณีที่มีทั้งบายดิง โปรตีนและพิษงูพบว่า สารด้นแบบทั้งสามสารและเรดิโอไซด์จับได้ทั้งแอลฟาโกบราท็อกซิน ในกรณีที่มีทั้งบายดิง โปรตีนและพิษงูพบว่า สารด้นแบบทั้งสามสารและเรดิโอไซด์จับได้ทั้งแอลฟาโกบราท็อกซิน ในกรณีที่มีกั้งบายดิง โปรตีนและพิษงูพบว่า สารด้นแบบทั้งสามสารและเรดิโอไซด์จับได้ทั้งแอลฟาโกบราทีอกซิน ในกรณีที่มีกิน เมื่อศึกษาต่อในหนูถีบจักรพบว่า สารด้นแบบและเรดิโอไซด์จับได้ทั้งแอลฟาโกบราทีอกซินและบายดิงโปรตีน เมื่อห้งหลังจากถืดสาร 30 นาที แต่มีสารต้นแบบสองสารก็อ NCI121865 และ NCI134754 เท่านั้นที่มีฤทธิ์ ด้านพิษงู โดยสามารถยึดระยะเวลาการตายของหนูเมื่อฉีดสารหลังจากแอลฟาโกบราท็อกซินทันที ในทางกลินิก NCI121865 และ NCI134754 จะมีประโยชน์ในการพัฒนาเป็นสารด้านพิษงูในการรักษผู้ถูกจูกัด

170 หน้า

# CONTENTS

Page

iii
iv
viii
х
xiii
xiv
1
4
4
24
27
33
36
38
38
38
38
42
42
43
49
49
49
53
54
68
87
92

# **CONTENTS (Cont.)**

V. CONCLUSION	94
REFERENCES	97
APPENDIX	107
Appendix A	108
Appendix B	115
Appendix C	119
Appendix D	121
BIOGRAPHY	170

vii

Page

# LIST OF TABLES

Tables	Page
1. Systematics of venomous snakes	6
2. Enzymes found in snake venoms	10
3. Main clinical features of snakebite	21
4. Medicinal plants in Thailand	25
5. 17 nACHR subunits	31
6. Notable variations of nAChR	32
7. The composition of separating gel and stacking gel	47
8. Critical values of Q	52
<ol> <li>Amino acid residues of α-cobratoxin interacted with the control peptide and the AChBP in the crystal pose (1YI5)</li> </ol>	59
10. Cluster analysis from docking $\alpha$ -cobratoxin with rediocides A and G	63
11. Amino acid residues of $\alpha$ -cobratoxin forming hydrogen bonds to rediocides	64
12. The top 175 compounds for each $\alpha$ -cobratoxin templates	66
13. Selected hits from virtual screening	67
14. Percent amino acid identity of AChBPs	68
15. NCI diversity set identified from virtual screening	76
16. $K_d$ of 4 hits on AChBPs	77
17. Amino acid residues of $\alpha$ -cobratoxin interacted with 3 hits	79
18. Toxicity test of rediocides and 3 hits (n=6)	88

# LIST OF TABLES (Cont.)

Tables	Page
19. Protective effect of rediocides (A & G) and 3 hits against $\alpha$ -cobratoxin	90
20. Anti-cobratoxin of rediocides against $\alpha$ -cobratoxin	91
21. Anti-cobratoxin of 3 hits against $\alpha$ -cobratoxin	92
B1. The top 175 compounds for each $\alpha$ -cobratoxin	115
C1. The screening test data from radioligand competition assay	119

# LIST OF FIGURES

Figures	Page
1. Structures of rediocide A and G	2
2. The morphology of a venomous snake	5
3. Three species of cobras living in Thailand	8
4. The structure of $\alpha$ -cobratoxin	15
5. Pathways of complement activation	16
6. The diagram of the alternative pathway of CVF	17
7. Schemeatic drawing of the degradation products of human C3	18
8. The amino acid sequence of pro-CVF	20
9. The major components of <i>Lotthanong</i>	26
10. Acetylcholine receptor (nicotinic) from electric torpedo rays	28
11. Acetylcholine receptor blocked by cobra venom (PDB code: 1YI5)	28
12. The 1YI5 bound crystal between $\alpha$ -cobratoxin (chain F-J) and	
AChBP (chain A-E); (PDB: 1YI5)	55
13. The control peptide for the $\alpha$ -cobratoxin template validation	56
14. The clusters of docked control peptide in the same 3D structure	
(RMSD of 2 Å), docking energy is in kcal/mol	57
15. Superposition of crystallographic Ser182 – Tyr 192 of chain C from	
1YI5 and docked orientation of control peptide (A-C)	58
16. Structures of rediocides A and G	60
17. Clusters of docked rediocides in the same 3D structure (RMSD $< 2$ Å),	
(A) rediocide A and (B) rediocide G, docking energy is in kcal/mol	61

# LIST OF FIGURES (Cont.)

Figures	Page
<ol> <li>The bound conformations of rediocides in the active site of α- cobratoxin</li> </ol>	62
<ol> <li>The bound conformations of rediocides in the active site of α- cobratoxin (A and B)</li> </ol>	65
20. The amino acid sequences of AChBPs ( <i>Ac</i> , <i>Ls</i> and <i>Bt</i> ) and human nAChR ( $\alpha$ 7 and $\alpha$ 1)	69
21. AChBPs bound to the bead by using the monoclonal anti-FLAG M2 antibody from mouse	70
22. Radioactive [ <sup>3</sup> H] epibatidine or [ <sup>125</sup> I] bungarotoxin bound to the binding protein and emitted the light	71
23. Screening test for the ability to displace the binding of [ <sup>3</sup> H] epibatidine on AChBP from <i>Lymnaea stagnalis</i> ( <i>Ls</i> ), <i>Aplysia</i> <i>californica</i> ( <i>Ac</i> ) and <i>Bulinus truncatus</i> ( <i>Bt</i> )	72
24. The ability to displace the binding of [ <sup>3</sup> H] epibatidine on AChBPs (~500 pM binding site) from <i>Lymnaea stagnalis</i> (A), <i>Aplysia</i> <i>californica</i> (B) and <i>Bulinus truncatus</i> (C)	73
25. The ability to displace the binding of [ <sup>125</sup> I] $\alpha$ -bungarotoxin on AChBPs from <i>Lymnaea stagnalis</i> (A) and <i>Aplysia californica</i> (B)	74
26. The dock orientations of NCI42258 (indigo blue), NCI121865 (light green) and NCI134754 (purple) (A)	78
<ul> <li>27. The amino acid residues of α-cobratoxin interacted with NCI42258</li> <li>(A), NCI121865 (B) and NCI134754 (C) with distance &lt; 4 Å</li> </ul>	79
<ul><li>28. SDS-PAGE of the mixtures between crude venom and rediocides (A &amp; G)</li></ul>	81

# LIST OF FIGURES (Cont.)

Figures	Page
29. SDS-PAGE of AChBP from <i>Aplysia stagnalis</i> in the presence of $\alpha$ -	
cobratoxin and rediocides (A & G)	82
30. SDS-PAGE of the mixtures between crude venom and 3 hits	84
31. SDS-PAGE of $\alpha$ -cobratoxin and AChBP from <i>Aplysia stagnalis</i> in the	
presence of 3 hits	86
32. LD <sub>50</sub> of $\alpha$ -cobratoxin (n=6)	87
33. Toxicity of rediocides (A & G). The mouse was injected with	
rediocides (A) and NCI134754 (B)	89

# LIST OF SCHEMES

Schemes	Page
A1. Virtual screening map	108
A2. Virtual screening data	109
A3. The virtual screening data to prepare the dpf files	112

# LIST OF ABBREVIATIONS

a-cbtx	alpha-cobratoxin
Å	angstrom
AChBP	acetylcholine binding protein
°C	degree celcius
σ	gram
b	hour
	50% inhibitory concentration
i	intramuscular
1.111.	intravenous
1.V.	liter
	mean lethal dose
$LD_{50}$	mole/liter, molar
M	milligram
mg	minute
min	milliliter
ml	millimole/liter millimolar
mM	millimole
mmol	malaaular waisht
MW	molecular weight
nAChR	nicotinic acetylcholine receptor
NCI	National Cancer Institute
nM	nanomole/liter, nanomolar
NMR	nuclear magnetic resonance
PDB	Protein Data Bank
RMSD	root mean square deviation
rnm	round per minute
SD	standard deviation
	microgram
$\mu g$	

# LIST OF ABBREVIATIONS (Cont.)

μl	microliter
V/V	volume by volume
w/v	weight by volume

# CHAPTER I

## **INTRODUCTION**

Snake bites are an occupational hazard, especially among tropical forest workers, farmers and agriculturists. In many cases, survivors are left with chronic functional disability from the necrotic effects of the venom, resulting in chronic ulceration, chronic renal failure and neurological sequelae from intracranial hemorrhages and thromboses or even amputation (1-4). Snake venoms contain complex mixtures that consist of several toxic proteins with diverse biological activities. Venom from the Thai cobra Naja kaouthia contains paralytic "neurotoxin" activity, which acts on nicotinic acetylcholine receptors (nAChR), cytolytic "cardiotoxin" activity, which acts on cell membranes, as well as phospholipase, and D-amino acid oxidase activities (5-7). Alpha-cobratoxin or neurotoxin 3 is the main component in cobra venom which accounts for 20-30% of the venom, dried weight. Antivenoms against many snake venoms have been produced and are being used for specific treatment of snake envenomations. Although polyvalent antivenom is an effective treatment, the recovery of the victim is slow and larger doses are needed. Furthermore, severe allergic reactions and serum sickness resulting from antivenom therapy are serious side effects. In Thailand, the only remedy for cobra bite is a monospecific antivenom produced by Queen Saovabha Memorial Institute, The Thai Red Cross Society. The horse antivenom against Naja kaouthia is very difficult to produce, expensive and in short supply. Only 20% of the horses immunized with Thai cobra venom produce adequate neutralizing activity for antivenom production. Furthermore, the neutralizing activity considered adequate is in fact quite low.

 $\alpha$ -Cobratoxin (Cbtx), a member of the long  $\alpha$ -neurotoxin family, is obtained from the venom of *Naja kaouthia* (previously called *Naja naja siamensis*). Cbtx has 71 amino acid residues and 5 disulfide bridges (8). It consists of three finger-like loops: loop I, II and III. It blocks nerve transmission by binding to the nAChR on the postsynaptic membranes of skeletal muscle and/or neurons, causing paralysis by preventing acetylcholine binding to the nicotinic acetylcholine receptor (9). nAChR mediates excitatory transmission at the neuromuscular junction and in the central and peripheral nervous systems. The soluble acetylcholine-binding protein (AChBP) from the freshwater snail *Lymnaea stagnalis* (*Ls*) is a structural homologue of the extracellular ligand-binding domain of muscle-type and neuronal nAChRs (10-11). Recently, the NMR and crystal structure of a Cbtx-AChBP complex (1YI5) was discovered (12-14). Since this initial discovery, AChBP have been identified in the molluskan species, *Bolinus truncatus* (*Bt*), and the marines-species, *Aplysia californica* (*Ac*) (15-16).

Application of medicinal plants with anti-snake venom activities might be useful as first aid treatment for victims of snake bite. There are about forty plants in Thailand reported to be effective in treating snake bites (17). One of the potential plants among these Thai plants was *Trigonostemon rediocides* (Kurz) Craib, which is known as *Lotthanong* in Thai belongs to the Euphorbiaceae family. The herb is particularly effective in treating snake bites especially against snake neurotoxins. According to the preliminary testing, it was found that the crude extract from Lotthanong's root can extend the survival time of the crude Thai cobra venom treated mice. In previous investigations of *T. rediocides*, the daphnanediterpenoids (rediocides A and G) (Figure 1) have been found to be the main components in the roots of this plant (18-21).



Figure 1. Structures of rediocide A (left) and G (right).

In search of an antidote of cobra venom by proventing the  $\alpha$ -cobratoxin from binding to the nAChR, the molecular docking was carried out to investigate the binding mode of rediocides to  $\alpha$ -cobratoxin. This study aimed to find the mechanism of rediocides against cobra venom at the molecular level to promote the utilization of medicinal plant in clinical use. In addition to the investigation of the molecular mechanism of rediocides, virtual screening was accomplished by using  $\alpha$ -cobratoxin as template and docked with 1990 compounds in National Cancer Institute diversity set. The objective of virtual screening is to discover the structure of new lead for drug design (22). The *in vitro* SDS-PAGE and binding assay as well as *in vivo* anti-toxin tests were performed to support the *in silico* results. The findings will be useful for patients who are bitten by snakes and will be useful for medical and pharmaceutical treatment in Thailand.

## **CHAPTER II**

## LITERATURE REVIEW

#### Part I. Snake

### 1. Snake and venomous snakes

Snake are cold-blooded vertebrates of class Reptilia, subclass Synaptosuaria, order Squamata, suborder Serpentes, which related to turtles, crocodiles and lizards. Anatomically, snakes have a long tube shape with narrow tail and they have heat reception pits as a heat-detecting organ. Their skins cover with scales, which are also used in classification (23). Although, they are deaf to air-borne sounds and have very poor vision, they have keen sense of smell, and are highly sensitive to ground conducted vibrations. Since snakes lack internal means of regulating body temperature, diurnal and seasonal temperature variations influence their degree of activity (24).

One of the anatomical characters of dangerous venomous snakes that distinguish them from harmless ones is "venom apparatus". The venom apparatus consists of venom glands, venom ducts that connect to venom fangs. The venom fangs are hollow and their function is like that of a hypodermic syringe (25). The morphology of a venomous snake is shown in Figure 2 (26). Generally, the function of the venom apparatus is to secure food and to aid in digestion of prey (27-29). Young snakes of poisonous families ar considered venomous at birth (24).

Mostly, snakes eat small vertebrates and some eat prey, such as rodents, which are capable of inflicting serious injury. Rapid dispatch of the prey through the injection of venom; therefore, greatly reduce the hazard to the snakes. It is a safe rule that venomous snakes do not attack people; they bite defensively when threatened and sometimes only after warning signal has fail to deter. Indeed, many poisonous snakes have warning signals, the effectiveness of which would seem to depend upon the surviving large animals learning to avoid the snakes concerned. Bitting of venomous snakes to large animals such as man is secondary defensive in nature (30).



Figure 2. The morphology of a venomous snake.

There are numerous species of venomous snake reorted worldwide. Based on the morphological characteristics, three major families are classified as shown in Table 1 (30). Thailand locates in the tropical area where appropriate for inhabitant of many species of venomous snakes. There are 19 kinds of deadly venomous snake, 11 kinds of venomous snake, 28 kinds of mildly venemous snake and 24 kinds of venomous sea snake (31). Among these, the most medically important terrestrial venomous snakes of Thailand are 6 species, which belong to the Elapidae and Viperidae families. The elapids include *Naja kaouthia*, also known as *Naja naja siamensis* (Thai cobra), *Ophiophagus hannach* (King cobra) and *Bungarus fasciatus* (banded krait). While the vipers include *Daboia russelli siamensis* (Russell's viper), *Calloselasma rhodostoma* (Malaya pit viper) and *Trimeresurus albolabris* (green pit viper) (32-34). The monocellate Thai cobra, *Naja kaouthia*, is a major cause of snakebite mortality and morbidity throughout Thailand. There is a high incidence of cobra bite amongst the fisherman and rice farmer in rural part of Thailand (35).

Family	Subfamily	Genera
Colubridae	Pseudoboinae	3 genara
	Aparallactinae	16 genera
	Atractaspindra	Atractaspis
	Boiginae	27 genera
	Homalopsinae	10 genera
	Xenodontinae	11 opisthoglyphous genera
	Natricinae	1 opisthoglyphous genera
	Colubrinae	Dispholidus, Thelotornis plus 9 other
		opisthoglyphous genera
Viperidae	Azemiopinae	Azemiops
	Viperinae	Adenorphinus, Atheris, Bitis, Causus, Cerastes, Echis, Eristicophis, Pseudocerastes, Vipera
	Crotarinae	Agkistrodon, Bothrops, Crotalus, Lachesis, Sistrurus, Trimeresurus
Elapidae	Elapidae	Acanthophis, Aspidelaps, Aspidomorphus, Austrelaps, Boulengerina, Bungarus, Cacaphis, Calliophis, Cryptophis, Demansia, Dendroaspis, Denisonia, Drysdali, Echiopsis, Elapognathus, Elapsoidea, Furina, Glyphodon, Hemachatus, Hemiaspis, Homoroselaps, Hoplocephalus, Leptomicrurus, Loveridgelaps, Maticora, Micropechis, Micruroides, Micrurus, Naja, Notechis, Ogmodon, Ophiophagus, Oxyuranus, Parademansia, Paranaja, Parapistocalamus, Pseudechis, Simoselaps, Solomorelaps, Suta, Toxicocalamus, Tropidechis, Vermicella, Walterinnesia
	Laticaudinae	Laticauda
	Hydrophiinae	
	Ephalophiini	Aipysurus, Emydocephalus, Ehpalohpis, Hydrelaps, Parahydrophis
	Hydrophiini	Acalyptophis, Enhydrina, Hydrophis, Kerilia, Kolpophis, Pelamis, Thalassophis

# Table 1. Systematics of venomous snakes (30)

### 2. Cobra

"Cobras" belong to family Elapidae, genus Naja. There are at least eight full species, which found throughout India and Southeast Asia (34). Cobras are easily distinguishable from other snakes by their neck can extent to be hood when there are in an erect position. Like all members of the Elapidae family, they have a pair of short, fixed fangs without covered sheath, which known as the Proteroglypha fangs, in front of the mouth (23-25).

There are three species of cobra presented in Thailand and are quite confusing in the identification. The first one is the monocellate non-spitting cobra or Thai cobra, (Naja Kaouthia). Naja kaouthia also known as Naja naja siamensis is the most dangerous cobra in Thailand. This specie is identified by an O-shaped marking on the back of the neck as shown in Figure 3a. Thai cobra is found in most parts of Thailand but mostly on central plain. Its length is around 755 mm (281-1680 mm). There are varies widely in its color and its hood-marking pattern. It is usually found in termite mounds, near habitation, forests and foothills up to 900 meters. It is also found in Myanmar, Laos, Vietnam and Malaysia. The other two species are spitting cobras, Naja siamensis and Naja sumatrana. These two species can spit venom to their enemies such as men and dogs to defense themselves, while this behavior is not found in Naja kaouthia. This could be due to the difference in physical characteristic of their fangs. Naja siamensis is found throughout the country except in the peninsular. In some area of Thailand, this specie variously known as Isan spitting cobra, the black spitting cobra and black and white spitting cobra. This form is also referred to as N. atra or N. sputratrix by some investigations. The V-shaped marking, U-shaped marking or H-shaped marking on the hood is used for define snakes in this species as shown in Figure 3b. The last one, Naja sumatrana (Golden spitting cobra) is occasionally found in southern part of Thailand particularly Surat Thani, Nakhon Sri Thamarat and Pattalung provinces. However, it is also found in Malaysia and Indonesia. This is the form referred to as N. naja sputratrix by some investigators (25, 28). Usually, there are not found any markig on the hood of snakes in this species as shown in Figure 3c (acellate) (23, 36-38).





Figure 3. Three species of cobras living in Thailand (23).

### 3. Snake venom

#### 3.1 Venom compositions

Poisonous snakes produce and store venom in bilateral venom glands that are similar in function to the human submaxillary glands. These glands are contracted by surrounding muscles that force the venom into venom ducts and subsequently into the fangs during the bite (39). Venom usually exhibits yellowish and transparent liquid due to the presence of L-amino acid oxidase, which contains riboflavin as a prosthetic group (40). The primary biologic purpose of snake venom is to facilitate the incapacitation and death of prey; the second is to initiate digestion before ingestion (28). Therefore, various enzymes in the venom start the digestion of the tissue after the venom is injected into the prey. The death of prey is due to respiratory or circulatory failure cause by various substances within venom acting alone or synergistically (40). Fac. of Grad. Studies, Mahidol Univ.

The venoms of snakes are usually composed of a complex mixture of organic and inorganic components. Insoluble tissue debris is also often noted in the venom from milk snakes. The inorganic constituents of the venoms include: Ca, Cu, Fe, K, Mg, Mn, Na. P, Co and Zn. Not all of these metals are found in every type of venom, and amount of each metal varies with the species of snake. The biological role of each of the metals is not clear; however, it is likely that some of them, such as Ca, Mg, and Mn, are quite important for the stabilization of certain venom proteins. While others, in particular, Zn, Cu, Fe and Co, may be actual participants in the catalytic mechanisms of certain venom enzyme components, such as the metalloproteinases. Although, it is continent to divide the organic compounds of the venom into proteinaceous and non-proteinaceous material; the majority of the crude venom is composed of proteinaceous components. The other compounds include carbohydrates, nucleosides, biofenic amines, amino acids and lipids. Carbohydrates are mainly present in the form of glycoprotein (41).

Proteinaceous material, proteins and peptides, comprise approximately 90-95% of the dry weight of the venom (42). Venom proteins have many diverse enzymatic activities including, but not limited to, phospholipase A<sub>2</sub>, phosphodiesterase, phophomonoesterase, L-amino acid oxidase, acetylcholinesterase, proteolytic activity (enzymes of the serine proteinase and metalloproteinase classes), arginine esterase, 5<sup>'</sup>-nucleotidase, hyaluronidase and nicotinamide adenine dinucleotide nucleosidase as shown in Table 2. Peptides found in snake venoms may comprise of presynaptic and postsynaptic neurotoxins, potassium channel-binding neurotoxins, cytotoxins, myotoxins, cardiotoxins and platelet aggregation inhibitors (disintegrins) (40, 43). Not all of these enzymes and peptides are present in any one kind of snake venom. The proportions of different components in snake venoms vary with species. For example, the venoms of Viperidae and Crotalidae. The hydrophid venoms are abundant in myotoxins and neurotoxins (39). However, the closer the phylogenetic relationship of snakes, the more similar venom properties and compositions (44).

Venom	Enzyme
all snake venoms	Phospholipase A <sub>2</sub> , L-amino acid oxidase,
	Phosphodiesterase, 5 <sup>'</sup> -Nucleotidase,
	Phosphomonoesterase, Deoxyribonuclease,
	Ribonuclease, Adenosine triphosphatase,
	Hyaluronidase, NAD-nucleosidase, Arylamidase,
	Peptidase
crotalid and viperid venoms	Endopeptidase, Arginine ester hydrolase, Kininogenase,
	Thrombin-like enzyme, Factor X activator,
	Prothrombin activator
elapid venoms	Acetylcholinesterase, Phospholipase B,
	Glycerophosphatase
some venoms	Glutamic-pyruvic transaminase, Catalase, Amylase, B-
	Glucosaminidase, Lactate dehydrogenase, Heparinase-
	like enzyme
cobra venom	Mettalloproteinase, Serine proteinase,
	Acetylcholinesterase, neurotoxin, cobra venom factor

**Table 2.** Enzymes found in snake venoms and composition of cobra venom (40)

### 3.2 Pharmacology of snake venom

The specific pharmacology of snake venom has been difficult to delineate because a various venom components may have one or several receptor sites, and these are not necessary organ-specific or system-specific. When venom is dispensed subcutaneously, it is absorbed through the lymphatic system as well as the capillary bed and other cell membranes. Venom also may be injected directly into an artery or vein and distributed throughout the blood-vascular system to other tissue (27).

Basically, the venom is an adaptation to obtain food. Because the snake's prey is usually large and swallowed whole, digestion time in the snake's gut may be prolonged up to 14 days. Putrefaction may force the snake to regurgitate its meal before digestion takes place. Injection and circulation of digestive enzymes before ingestion of the prey reduces digestive time to 2 to 5 days. The venom therefore contains a lethal protein component designed to immobilize the prey for a short period during which the second component, the digestion enzymes, may be distributed throughout the prey's tissues. Thereafter, the prey dies before it can retreat out of the snake's range of activity (45).

The lethal proteins and peptides consist of 20 to 80 amino acid chains and range in molecular weight between 4,800 and 100,000 Da. Electronmicroscopy shows that these proteins damage endothelial cells of vascular walls, causing blebs in the endothelium, dilating the perinuclear space, and breaking down the plasma membrane. Blocking neuromuscular transmission results in flaccid paralysis and impaired respiration. Plasma and erythrocytes leak into the tissues, resulting in often massive accumulation of fluid in intracellular spaces; this is manifested as edema. Plasma loss reduces the ciculating blood volume and leads to hypovolemic shock, hemoconcentration, and lactic acidosis. The capillary bed of the lungs may also be affected by the proteins, which led to fluid loss into the alveoli, pulmonary edema and hemorrhage (45).

#### 3.2.1 Enzymes in snake venom

Digestive enzymes containing within snake venom cause both local and systemic damage to human tissue. At least 26 enzymes have been identified as shown in Table 2. The major enzymes and their pharmacological effects are the following (40, 44-45).

### 3.2.1.1 Phospholipase A<sub>2</sub>

One of the most common enzymes is found in snake venoms. All venoms of Hydrophiidae (sea snakes), Elapidae, Viperidae and Crotalidae so far investigated contain this enzyme. They catalyze the hydrolysis of the ester bond at  $C_2$  position of lecithin, releasing a molecule of fatty acid and isolecithin. The enzymes thus damage the fatty molecules of all membranes. Damage to muscle fiber membrane results in local necrosis. Permeability of erythrocyte membranes is altered, allowing water to enter the cells sell, eventually resulting

in haemolysis. Precipitation of free hemoglobin and erythrocyte ghosts in kidney tubules leads to renal shutdown (7, 46).

3.2.1.2 Hyaluronidase

This enzyme cleaves internal glucoside bonds of various collagens, decreasing the viscosity of connective tissue and allowing venom fractions to spread beyond the site of the bite. Although venom hyaluronidase is not toxic per se, it enhance the effect of the toxic by increasing their spread. This enzyme is also referred to as the "spreading factor".

3.2.1.3 Amino acid esterase and other thrombin-like enzymes

These enzymes promote fibrin formation by incompletely splitting ether fibrinopeptide A or B from the fibrinogen molecule. Factor XII is not activated, however. Instead, an unstable fibrin clot is formed, trapping platelets. This clot is readily lysed by protein spitting enzymes in the snake venom. The result is disseminated intravascular coagulation syndrome followed by often massively hemorrhage due to consumption of platelets and fibrinogen. This bleeding decreases the already depleted volume of blood circulation.

3.2.1.4 Proteolytic enzymes and phosphomonoesterase

These enzymes include R Nase, D Nase, and 5' nucleotidase, which damage muscle fiber proteins, resulting in local necrosis if the venom is injected deeply enough to penetrate a muscle sheath. Up to five of these enzymes in this group may occur in the same type of venom.

3.2.1.5 Acetylcholinesterase

Commonly, this enzyme occurs in nerves, erythrocytes, and the retina. It is also commonly present in the venom of Elapidae and Hydrophiidae but not in those of Viperidae and Crotalidae. Since acetylcholinesterase is involved in nerve transmission, it was thought at one time that venom acetylcholinesterase was responsible for the neurotoxic action of Elapidae venom. Fac. of Grad. Studies, Mahidol Univ. Ph.D. (Pharmaceutical Chemistry and Phytochemistry) / 13

#### 3.2.1.6 Serine proteinase

Snake venom glands synthesize a variety of serine proteinases capable of affecting the heamostatic system. They act on macromolecular substrates of the coagulation, fibrinolytic, and killikrein-kinin systems, and on platelets to cause an imbalance of the haemostatic system of the prey (47).

#### 3.2.2 Toxin in snake venom

As known that, snake venom usually contains more than one toxic principle, and these tend to act in combination in actual poisoning. The overall toxicity is due to enzymes as well as to non-enzymatic proteins. However, the main lethal action, especially elapid and hydrophid venoms, can be attributed to neurotoxins. Postsynaptic neurotoxin or alpha-neurotoxin is the most lethal among the toxic components in elapid venoms. The type of toxins found in venoms are described below:

## 3.2.2.1 Snake neurotoxin

Snake neurotoxin are peptides, which can cause skeleton muscle paralysis by inhibiting neuromuscular transmission. They are divided into two pharmacological classes based on their sites of action.

**Presynaptic neurotoxins:** These peptide toxins have been found in the venom of Elapidae, Crotalidae and Viperidae snakes with phopholipase  $A_2$  activity. These toxins interact with a target that is specifically present on the synaptic plasma membrane of motor nerve terminals. They cause blockade of neuromuscular transmission by inhibiting the release of acetylcholine from motor nerve (40, 45-48).

**Postsynaptic neurotoxins:** These peptide, toxins found predominantly in elapid venoms, are also called curaremimetic toxins (alpha-cobratoxin or neurotoxin3) based on their mimicing action of curare on poison arrow of South American Indians. The postsynaptic neurotoxins are low molecular weight basic proteins found mostly in Elapidae or hydrophidae snakes (24). Two types of postsynaptic neurotoxins can be distinguished on the basis of their sizes and disulfide linkages. These are short chain toxins containing 60-62 amino acid residues with four disulfide bonds and long chain

toxins containing 71-74 residues and four or five disulfide bonds. Both types show extensive homology in amino acid sequence and act in similar manner: blocking neuromuscular transmission by binding speifically and with high affinity to the nicotinic acetylcholine receptor. However, short chain toxin shows faster dissociate rate than that of long chain ones. They are immunochemically distinct from each other; hence antibodies produced against them do not cross-react. Moreover, they differ in circular dichroism spectra and stability including heat resistance, lyophilization and chemical modification (40, 44, 49-51).

One of the best known snake postsynaptic neurotoxins in Naja kaouthia (alpha-cobratoxin or neurotoxin3), which accounts for 20-30% of the venom, dried weight. This is the highest content of an individual neurotoxin found in venom of any terrestrial snakes in Thailand. It is the basic polypeptide containing 71 amino acid residues with 5 disulfide bridges. The structure was shown in Figure 4. The molecular weight of this toxin is 7821 daltons (39, 50-52). It consists of three fingerlike loops: loop I, II and III. It blocks nerve transmission by binding to the nicotinic acetylcholine receptor (nAChR) on the postsynaptic membranes of skeletal muscle and/or neurons, causing paralysis by preventing acetylcholine binding to the nicotinic acetylcholine receptor (nAChR) (9). This postsynaptic neurotoxin is largely responsible for the toxicity and lethality of Thai cobra venom. nAChR mediates excitatory transmission at the neuromuscular junction and/or in the central and peripheral nervous systems. The soluble acetylcholine-binding protein (AChBP) from the freshwater snail Lymnaea stagnalis is a structural homolog of the extracellular ligand-binding domain of muscle-type and neuronal nAChRs (10-11). Its pentamers composed of either identical or homologous transmembrane subunits. Recently, the NMR and crystal structure of a Cbtx-AChBP complex (1YI5) was discovered. The amino acid residues from Cbtx that bind AChBP are Thr6, Pro7, Ile9 in loop I, Trp25, Cys26, Cys30, Asp27, Ala28, Phe29, Ser31, Ile32, Arg33, Gly34, Lys35, Arg36, Val37 in loop II and Phe65, Arg68 for C-terminus (12-14). Since this initial discovery, AChBP have been identified in the molluskan species, Bolinus truncatus (Bt), and the marines-species, Aplysia californica (Ac) (15-16).



**Figure 4**. The structure of  $\alpha$ -cobratoxin. Disulfide bridges are in yellow line (8).

### 3.2.2.2 Cardiotoxin

Cardiotoxins are other low molecular weight (around 6000-7000 kDa) basis peptide toxins found in elapid venoms. They are single-chain polypeptides chemically and structurally related to the postsynaptic neurotoxins. They contain approximately 60 amino acid residues and 4 disulfide bridges. They have lytic effect on a wide range of cells so they were named as "Direct Lytic Factor", cytotoxins or membrane toxins. However, the name cardiotoxins were extensively used because the heart appears to be the main target of these toxins and that the toxins are much more selective in their actions than other names above.

Cardiotoxins cause death "shock" follows quickly by fastened  $K^+$  released from lysed RBC. The lethality of cardiotoxins is greatly increased in the presence of phopholipase A<sub>2</sub>, which also has a synergistic effect on other activities displayed by this toxin.

Apart from lytic effect on erythocytes and various cell types, they also induce depolarization resulting in blockade of axonal conduction in peripheral nerve, ventricular fibrillation, systolic arrest of heart muscle and prolonged muscular contractures with paralysis. It was shown that cardiotoxins affect the membrane and not the myofibril itself. The effect is irreversible. However, addition of  $Ca^{++}$  can prevent this effect due to inhibition of toxin binding (40, 44, 53).

#### 3.2.2.3 Cobra venom factor

In addition to the hemolytic pathways, which involve the venom phspholipases and cardiotoxin, red cell lysis can be produced through an entirely different mechanism by one component in cobra venoms called "Cobra Venom Factor" (CVF). CVF is active on the alternative pathway complement system. Cobra venom factor (CVF) is the complement-activating protein from venom, causing complement consumption or complement depleting in human and mammalian serum (54). The complement system is an integral component of the immune system and plays important roles in immune response and host defense. The system is comprised of approximately 20 plasma proteins and a number of cell surface receptors. In addition to its physiological role, the complement system is also activated in many disease states and contributes importantly to the pathogenesis of many diseases and conditions of major medical importance including, rheumatoid arthritis and other immune complex diseases, certain types of glomerulonephritis, autoimmune hemolytic anemia, myasthenia gravis and other autoimmune diseases, and reperfusion injury (55). The pathway of complement activation was shown in Figure 5.



Figure 5. Pathways of complement activation.

The latter refers to the tissue damage that occurs after reopening of a blocked vessel after a heart attack, stroke, or reperfusion with a recipient's blood after organ transplantation. The molecular interactions of CVF with human complement were elucidated experimentally, simultaneously revealing the reaction sequence of the

alternative complement pathway and the functional analogy of CVF and C3b, the activated form of C3. The diagram of the alternative pathway was shown in Figure 6. Like C3b, CVF binds to factor B in human and mammalian serum to form the complex CVF-B (56), which is cleaved by factor D into the bimolecular enzyme CVF-Bb and Ba. The bimolecular complex CVF-Bb is a C3/C5 convertase that activates C3 and C5 analogously to the C3/C5 convertase formed with C3b (57-61).



Figure 6. The diagram of the alternative pathway of CVF.

CVF is a three-chain 149 kDa glycoprotein (58-59). The functional similarity of CVF and C3b correlates with many structural similarities, including immunological cross reactivity, secondary structure, and N-terminal amino acid sequences (60-61). From these data, it was concluded that CVF structurally resembles C3c, a physiological degradation product of C3 which lacks an intramolecular thioester and a free sulfhydryl group. The degradation product of C3 was shown in Figure 7. There are several reasons why there is significant interest in CVF. For one, CVF can serve as a tool to investigate the multifunctionality of C3, a protein that interacts specifically with more than ten different plasma proteins or cell surface receptors. The elucidation of the structural differences between C3 and CVF, two closely related molecules that share some properties e.g., formation of a C3/C5 convertase but differ in others e.g.,

the susceptibility to regulation by factors H and I, can be expected to help identify functionally important regions of the C3 molecule. Both C3b-Bb and CVF-Bb exhibit spontaneous decay dissociation at 37 °C into the two respective subunits. However, the half-life of this decay dissociation is 7 h. for CVF-Bb and only 1.5 min for C3b-C3b-Bb is effectively disassembled by factor H and C3b is subsequently Bb. inactivated by factor I (62). In contrast, CVF-Bb and CVF are resistant to these two regulatory proteins, factor H and factor I (58, 63-64). Because of its stability and resistance to regulation, the CVF-Bb will continuously hydrolyze C3 and C5, eventually resulting in complement depletion. According to the potential of CVF in clinical relevance, CVF has been tested and shown that CVF to be safely administered to laboratory animals (65). The CVF has been used in numerous studies to deplete the plasma complement activity in order to investigate the role of complement in host CVF is a promising therapeutic agent for defense or pathogenesis of disease. complement depletion in clinical conditions where activation is involved in the pathogenesis of disease. This includes such diverse clinical conditions as reperfusion injury (66), xenograft rejection (67), rheumatoid arthritis and retroviral gene therapy. Another potential application of CVF is its use in antibody conjugates for targeted complement activation to induce tumor-cell killing (68).



Figure 7. Schemeatic drawing of the degradation products of human C3.

The molecular cloning and derived primary structure of CVF were reported. CVF mRNA is more than 5924 nucleotides in length (69). It contains a single open reading frame of 4926 nucleotides, coding for a pre-pro-protein of 1642 amino acids. The single-chain pre-pro-CVF consists of a 22-amino acid signal sequence, a 627amino acid  $\alpha$ -chain, and a 989-amino acid precursor chain for the CVF  $\gamma$ - and  $\beta$ chains. The 627 amino acid  $\alpha$ -chain has three potential N-glycosylation sites at residues 131, 136, and 187. Immediately following the C terminus of the  $\alpha$ -chain, there are 4 arginine residues and a 79 amino acid peptide resembling a C3a anaphylatoxin. There is a single potential glycosylation site at position 640 in the "C3a" domain, though this site is not present in the mature protein. The  $\gamma$ -chain begins at position 711. The exact position of the C terminus of the  $\gamma$ -chain is still unknown and is apparently heterogeneous. Based on the size of the y-chain on SDS/PAGE and the lack of a free sulfhydryl group, the y-chain must terminate in the immediate proximity to the C terminus of what would be C3g in C3. Assuming the C terminus of the  $\gamma$ -chain to correspond to the C terminus of the C3g domain in human C3, the  $\gamma$ chain would be 252 residues long with its C terminus being at position 962. The  $\gamma$ chain contains no glycosylation sites. The  $\gamma$ -chain is followed by a 279-residue-long "C3d" domain that is not present in the mature CVF protein. The  $\beta$ -chain of CVF begins at position 1242 and extends for 379 residues to the end of the open reading frame. The  $\beta$ -chain contains a single glycosylation site at position 1324 (69). The post-translational proteolytic processing of pro-C3 is limited to the removal of four arginine residues, resulting in the mature two-chain C3 molecule consisting of a 115kDa  $\alpha$ -chain and a 70-kDa  $\beta$ -chain, the proteolytic processing of pro-CVF in the venom gland is more complex. It involves in addition to the removal of four arginine residues, the proteolytic removal of the C3a-like and C3d-like domains, resulting in the mature three-chain form of CVF consisting of the 70-kDa  $\alpha$ -chain, 48-kDa  $\beta$ chain, and the 32 kDa y-chain (70). Pro-CVF contains five potential glycosylation sites, of which only three can be expected to be glycosylated in mature CVF. Like C3, pro-CVF contains 27 cysteine residues and a homologous thioester site in the C3d-like region. The amino acid sequence of pro-CVF was shown in Figure 8.



Figure 8. The amino acid sequence of pro-CVF.

### 4. Symtom and pathology of cobra bite

The common symptoms and pathology of snakebite patients can be classified in three groups, neurotoxic effects, myotoxic effects and haematotoxic effects, according to the mechanism of action of the toxic substances containing in snake venoms (23). The three major families of venomous snakes, which led to cause of neurotoxic, heamotoxic and myotoxic effects, are the elapids, vipers and sea snakes, respectively. The clinical features of snakebite are summarized in Table 3 (71). Furthermore, the ultimate severity of any venomous snakebite depends on the quantity of venom injected and the comparative toxicity of the venom.

Factors that affect the amount of venom to be injected included the extent or the anger motivates the attack, the length of time the snake hold on, the site of injection, the number of bite, the bite depth, interference by clothing and the age, size, and species of the snake (72).
P Snake c	Percentage	Effects of poisoning		Approximate	Average	
	poisoning	Local	Systemic	natural mortality (%)	death (h)	
Elapids	50	Slow swelling, then necrosis with Asian cobras, African spitting cobras, Usually no local efects with other elapids	Neurotoxic effects: ptosis, glossopharyngeal palsy, respiratory paresis, cardiac effects	10	5-20	
Sea snakes	80	None	Myotoxic effects: myalgia on moving, paresis, myoglobinuria, hyperlalaemia	10	15	
Vipers	30	Rapid swelling, necrosis in 5-10% (some vipers only)		1-15	48	

**Table 3**. Main clinical features of snakebite (71)

#### 4.1 Local symptoms

Pain, then variable swelling and later necrosis, were the outstanding features of local poisoning. The pain commences in the bitten area and frequently radiate up the limb and may last from less than 1 to over 10 days depending on the extent of the necrosis. Swelling is one of the most common features of cobra bite. It commences 2-3 hours after the bite and reaches its maximum extent 24-48 hours after the bite and may persist for up to 18 days. A bite on finger may lead to swelling of the hand and forearm (73-74).

Local necrosis is now accepted as the most common sequelae to an effective cobra bite but once was classified as peculiar to cobra bite. Approximately fifty percent of the victims bitten by the cobra face the problem of local tissue necrosis, which is dificult to treat (75). After bite, a dusky discoloration around the bite marks extent in area and deepening in colour each day. After 3-4 days, a grayish black area

becomes encircled by a red raised rim and sanguineous blisters may occur. Then fluctuation was often evident: incision released red-yellow material and revealed necrosis of skin and subcutaneous tissue. Local tissue necrosis is detected between 36 hours and ten days, usually by the fifth day (76). The area of skin necrosis may be found vary from a few to 600 cm<sup>2</sup>. The extensive skin loss may take several months to heal (73).

## 4.2 Systemic symptoms

Before going to the stage of paralysis, many preparalytic symptoms occur. These include: vomitting, loss of conciousness, headache, vasomotor sign such as pallor, sweating, weak to absent pulse and hypotension, pain in regional lymph nodes including tenderness and enlargement. Normally, the earliest systemic poisoning is drowsiness occures 1-5 hours after the bite. Also observed are spitting, vomitting and coughing of blood, allergic reaction and cobra venom conjunctivitis in case of spitting cobra venom gains access to the eyes (74-75).

The recognition of these preparalytic symptoms and signs of envenomation assumes great importance because, if they are recognized for what they are and antivenin is given at this stage, it may effectively prevent paralysis and necrosis developing or may limit its extent. However the paralytic latent period of cobra bite can be vary short, so these symptoms may occur in the same time of muscle paralysis.

The muscle paralysis, however, is not common sequelae to a cobra bite, but if it occurs, it tends to occur early in the course of envenomation and its progress is frequently rapid (76). The paralysis of several muscle groups are usually involved and are reported as difficulty in seeing or double vision, difficulty in opening the mouth, swallowing, speaking and breathing, difficulty in lifting limbs, eyelids and head. The muscles supplied by the cranial nerves are always affected first. The susceptibility of various muscles to neurotoxins varies considerably. The most susceptible are extrinsic eye muscles and elevator of eyelids while superficial facial muscle and diaphram are resistant.

In some cases, neurotoxic symptoms resolve rapidly, usually within a week. Drowsiness, ptosis, facial paresis, and difficulty in swallowing ensured. Limb and neck paresis were followed by depression then loss of tendon reflexes and these neurotoxic symptoms did not resolve completely until two weeks after the bite (72-74, 77).

## 5. Treatment of snake envenomation

The antivenom is the most recommended and the mainstay of treatment for snake venom poisoning. However, antivenom is a time proven and medically accepted standard in virtually every country. It should be given early after envenomation, because its effectiveness is both dose and time related (78). It was found that almost half of the patients receiving horse sera showed signs of hypersensitivity reactions (40, 45). It is therefore not only safe, but it is also highly desirable to wait for clear clinical evidence of systemic poisoning before giving antivenom. It should not be given routinely in all cases of snakebite. A patient with a mild envenomation can be managed without antivenom but the patient must be under close observation.

The only remedy for cobra bite is a monospecific antivenom produced by Queen Saovabha Memorial Institute, The Thai Red Cross Society. The horse antivenom against *Naja kaouthia* is very difficult to produce, expensive and in short supply. Only 20% of the horses immunized with Thai cobra venom produce adequate neutralizing activity for antivenom production. Furthermore, the neutralizing activity considered adequate is in fact quite low. Administration of antivenom to patient may cause hypersensitivity. Hypersensitivity reactions may be either type I (anaphylactic) or type III (complex mediated hypersensitivity). It was reported that approximately 25% of patients develop anaphylaxis in response to horse serum-derived antivenin and that 50% of those reactions could be life threathening. Skin testing is necessary before antivenin is administered. However, both fasle-positive and false-negative test results occur. Moreover, antivenin does not provide enough protection against venom, which need a long schedule, is expensive and requires ideal storage condition (39, 79).

Over the years many attempts have been made for the development of chemical antagonists which may provide protection against snakebite. These chemicals mostly receive from the medicinal plants due to the folk medicinal used of people in rural part of different countries. Not many compounds are known, even today, which is proved to be effective to counter venom action.

#### Part II. Medicinal plant against snake venom

Plant have long been used for treatment envenomation by snakebite. Many examples could be cited of plants, whose use as snake-venom anidotes can be traced back to antiquity. Ethnobotanical reports from every corner of the world abound with such citation (80). However, from several points of view, an overview of such plants would be useful. From a scientific viewpoint, it is important to know the truth of "anti-snakebite treatmants" preceding the advent of serum therapy less than 100 year ago. Another is more practically oriented: if certain substances from plants are really active against snakebite, they could be used as first-aid measures or to support any other treatment. Thirdly, industrial companies world wide are now interested in plants which have been used previously in folk medicine; quite a number of such substances have been successfully intoduced into modern therapy against various diseases.

There are about thirty eight plants in Thailand reported to be effective in treating snake bites. Most of them are the mixture more than 2 types of plants for treatment and it was shown in Table 4 (17). One of the potential plants among these Thai plants was *Trigonostemon rediocides* (Kurz) Craib, which is known as *Lotthanong* in Thai and belongs to the Euphorbiaceae family. The roots, ground with water, have been used in traditional medicine as an emetic for food and shell, and also used a laxative and anti-asthmatic. The herb has also been used in the treatment of bloody and mucous sputum or stool. Topically, it was applied to reduce abcesses and to alleviate sprains, swelling and bruises. The herb is particularly effective in treating snake bites especially against snake neurotoxins (81). According to the research report, it was found that the crude extract from Lotthanong's root can extend the survival time of mice after being injected by Thai cobra venom (82). The major components of *Lotthanong* were shown in Figure 9.

Aloe vera Linn.	Gynurapseudochina DC. Var.
Andrographis paniculata (Burm.) Wall.	Lygodium flexuosum Kinn. SW.
Antidesma leucopodum Mig.	Melastomr normale
Typhonium trilobatum Schott.	Micromelum sp.
<i>Barleria lupulina</i> Lindl.	Mimosa pudica Linn. Var. hispida Bren.
Boesenbergia pandurata (Roxb.) Schltr.	Mitragyna speciosa Korth.
Brachiaria distachya Linn. Stapt.	Musa balbisiana colla
Centella asiatica (Linn.) Urban.	Pandanus momotheca Martelli
<i>Cleome viscosa</i> Linn.	Phyllanthus glollrayi Beille
Clinacanthus nutans (Burm.f.)	Phyllanthus oxyphyllus Mig.
Cocos nucifera Linn.	Plumcia Rubra
Curcuma longa Linn. Roxb.	<i>Psidium guajava</i> Linn.
Cyathostemma micranthum (A.DC.)	Scirpus mucronatus Linn. ssp.
J.Sincl.	mucronatus
Cyathula prostrate (Linn.) Bl.	Tamarindus indica Linn.
Cyclea sp.	Tetracera loureiri Pierre
Dischidia imbricate (Blume) Steud.	Thunbergia laurifolia Linn.
Drynaria quercifolia (L.) J.Sm.	Tiliacora triandra Diels.
Glycosmis pentaphylla Corr.	Tithomsnospora crispa F.
Grammatophyllum speciosum BL.	Typhonium trilobatum Schott.

# Table 4. Medicinal plants in Thailand

Maleeruk Utsintong

Literature Review / 26



Figure 9. The major components of *Lotthanong*.

#### Part III. Nicotinic Acetylcholine Receptor

Nicotinic acetylcholine receptors, or nAChRs, are cholinergic receptors that form ligand-gated ion channels in the plasma membrane of certain neurons. Being ionotropic (i.e. ligand-gated) receptors, nAChRs are directly linked to an ion channel and do not make use of a second messenger like metabotropic receptors do. Like the other type of acetylcholine receptors, muscarinic acetylcholine receptors mAChRs), the opening of the ion pore of an nAChR is triggered by the binding of the neurotransmitter acetylcholine (ACh) and also by nicotine. The name "nicotinic" based on the ability of the nicotine to mimic the effects of ACh as a neurotransmistter (83). Nicotinic acetylcholine receptors are present in many tissues in the body, and are the best-studied of the ionotropic receptors. The neuronal receptors are found in the central nervous system and the peripheral nervous system. The neuromuscular receptors are found in the neuromuscular junctions of somatic muscles; stimulation of these receptors causes muscular contraction.





Acetylcholine

Nicotine

#### 1. Structure

Nicotinic receptors, with a molecular mass of 290 kDa (84), are made up of five subunits, arranged symmetrically around the central pore. They share similarities with GABA<sub>A</sub> receptors, glycine receptors, and the type 3 serotonin receptors. They also called ionotropic receptors or the signature Cys-loop proteins (85). Twelve types of nicotinic receptor subunits,  $\alpha 2$  through  $\alpha 10$  and  $\beta 2$  through  $\beta 4$ , combine to form pentamers. The subunits are somewhat similar to one another, especially in the hydrophobic regions. At the neuromuscular junction, the two  $\alpha$  subunits of the nAChR are combined with up to four other subunits ( $\beta$ , $\gamma$ , $\delta$ , $\epsilon$ ) in the ratio 2 $\alpha$ : $\beta$ : $\epsilon$ : $\delta$  (83, 86). The neuronal forms are much more homogeneous, and are made up of only receptor subunit types ( $\alpha$  and  $\beta$ ) present in a ratio of  $3\alpha$ : $2\beta$ . The neuronal forms also

differ from the muscle forms in that they are not sensitive to  $\alpha$ -bungarotoxin. The acetylcholine binding site is on the outside of the  $\alpha$  subunit near the N terminus. When an agonist binds to the site, the  $\alpha$  subunits become more similar to the other subunits, the channel becomes more symmetrical (87), and a pore with a diameter of about 0.65 nm opens. Acetylcholine receptor (nicotinic) from electric torpedo rays and *Lymnaea stagnalis* were shown on Figure 10 and 11, respectively.



Figure 10. Acetylcholine receptor (nicotinic) from electric torpedo rays (very similar to human receptor) is made of 5 subunits, 2 of which (shown in orange) bind ACh (red). Structure was determined by electron crystallography at 4 Å resolution (PDB code: 2bg9).



Figure 11. Acetylcholine receptor blocked by cobra venom (PDB code: 1YI5). A similar effect can be achieved by high doses of curare or nicotine.

#### 2. Opening the Channel

Nicotinic AChRs may exist in different interconvertible conformational states. Binding of an agonist stabilizes the open and desensitised states. Opening of the channel allows positively charged ions to move across it, in particular, sodium and potassium, to enter the cell. The nAChR is a non-selective cation channel, meaning that several different positively charged ions can cross through. It is permeable to Na<sup>+</sup> and K<sup>+</sup>, with some subunit combinations that are also permeable to Ca<sup>2+</sup>. The amount of sodium and potassium the channels allow through their pores (their conductance) varies from 50-110 pS, with the conductance depending on the specific subunit composition as well as the permeant ion (88). Interestingly, because some neuronal nAChRs are permeable to Ca<sup>2+</sup>, they can affect the release of other neurotransmitters (83). The channel usually opens rapidly and tends to remain open until the agonist diffuses away, which usually takes about 1 millisecond. However, AChRs can sometimes open with only one agonist bound and in rare cases with no agonist bound, and they can close spontaneously even when ACh is bound. Therefore, ACh binding only creates a probability of pore opening, which increases as more ACh binds (87).

## 3. Effects

The activation of receptors by nicotine modifies the state of neurons through two main mechanisms. On one hand, the movements of cations cause a depolarization of the plasma membrane (which results in an excitatory postsynaptic potential in neurons), but also by the activation of other voltage-gated ion channels. On the other hand, the entry of calcium acts, either directly or indirectly, on different intracellular cascades leading, for example, to the regulation of the activity of some genes or the release of neurotransmitters.

#### 4. Receptor Regulation and Roles

Ligand-bound desensitization of receptors was first characterized by Katz and Thesleff in the nicotinic acetylcholine receptor (89). Prolonged or repeat exposure to a stimulus often results in decreased responsiveness of that receptor for a stimulus. nAChR function can be modulated by phosphorylation and the activation of second messenger-dependent protein kinases (90). Phosphorylation of the nAChR by protein kinase A and protein kinase C results in its desensitization (91-92). It has been reported that after prolonged receptor exposure to the agonist, the agonist itself causes an agonist-induced conformational change in the receptor, resulting in receptor desensitization.

The subunits of the nicotinic receptors belong to a multigene family (17 members in human) and the assembly of combinations of subunits results in a large These receptors, with highly variable kinetic, number of different receptors. electrophysiological and pharmacological properties, respond differently to nicotine, at very different effective concentrations. This functional diversity allows them to take part in two major types of neurotransmission. Classical synaptic transmission (wiring transmission) involves the release of high concentrations of neurotransmitter, acting on immediately neighbouring receptors. In contrast, paracrine transmission (volume transmission) involves neurotransmitters released by synaptic buttons, which then diffuse through the extra-cellular medium until they reach their receptors, which may be distant. Nicotinic receptors can also be found in different synaptic locations, for example the muscle nicotinic receptor always functions post-synaptically. The neuronal forms of the receptor can be found both post-synaptically (involved in classical neurotransmission) and pre-synaptically where they can influence the release of multiple neurotransmitters (93).

To date 17 nAChR subunits have been identified, these are divided into muscle-type and neuronal-type subunits (Table 5). Of these 17 subunits,  $\alpha 2$ - $\alpha 7$  and  $\beta 2$ - $\beta 4$  have been cloned in humans, the remaining genes identified in chick and rat genomes (94). The nAChR subunits have been divided into 4 subfamilies (I-IV) based on similarities in protein sequence (95). In addition, subfamily III has been further divided into 3 tribes.

Nicotinic receptors are pentamers of these subunits, i.e. each receptor contains five subunits. Thus, there is an immense potential of variation of the aforementioned subunits. However, some of them are more notable than others, specifically  $(\alpha 1)_2\beta 1\delta\epsilon$  (muscle type),  $(\alpha 3)_2(\beta 4)_3$  (ganglion type),  $(\alpha 4)_2(\beta 2)_3$  (CNS type) and  $(\alpha 7)_5$  (another CNS type) (96). A comparison was shown in Table 6.

Туре	Subtype	Subunit
Neuronal I		α9, α10
Neuronal II		α7, α8
Neuronal III	1	α2, α3, α4, α6
	2	β2, β4
	3	β3, α5
Muscle IV		α1, β1, γ, δ, ε

**Table 5.**17 nAChR subunits (95)

Receptor type (location)	Effect	Agonists	Antagonists
Muscle type:	EPSP, mainly by	acetylcholine	α-bungarotoxin
$(\alpha 1)_2\beta 1\delta \epsilon$	increased $Na^+$ and $K^+$	carbachol	α-conotoxin
or	permeability	suxamethonium	tubocurarine
α1β1δγε			pancuronium
(neuromucular			
junction)			
	EPSP, mainly by	acetylcholine	α-bungarotoxin
Ganglion type:	increased $Na^+$ and $K^+$	carbachol	mecamylamine
$(\alpha 3)_2(\beta 4)_3$	permeability	nicotine	trimetaphan
(autonomic		epibatidine	hexamethonium
ganglia)		dimethylphenylpiperaz inium varenicline	bupropion
			dextromethorphan
			ibogaine
			18-methoxycoronaridine
CNS type:	Post- and presynaptic	nicotine	mecamylamine
$(\alpha 4)_{2}(\beta 2)_{3}$	excitation, mainly by	epibatidine	methylcaconitine
(brain)	increased $Na^+$ and $K^+$	acetylcholine	α-conotoxin
	permeability	cytisine	
(another) CNS	Post- and presynaptic	epibatidine	mecamylamine
type:	excitation, mainly by	dimethylphenylpiperaz	α-bungarotoxin
(α7) <sub>5</sub>	increased Ca <sup>2+</sup>	mum	α-cobratoxin
(brain)	permeability		

# Table 6. Notable variations of nAChR (96)

#### Part IV. Automated Docking

Molecular recognition of small molecules by enzymes is a fundamental step in most biological processes, and predicting binding conformation and affinities of small molecules is a challenge in computational chemistry. Computaional techniques including rapid database searching, automated docking, molecular mechanics, free energy perturbation, and lambda dynamics have been integrated with structural information from x-ray crystallography and/or NMR in an effort to understand the physical forces that drive molecular recognition. One application of these affords has been structure-based drug design. The design of inhibitors for HIV-1 protease (97) and inhibitors of malaria cysteine protease (98) are considered major successes of structure-based drug design. The number of protein targets for structure-based drug design is rapidly increasing due to advances in crystallography, homology modeling, genomics, and proteomics. It is important to develop, evaluate, and improve computational methods in parallel with structural determination.

Automated docking is one strategy for predicting ligand-protien interactions and estimating interaction energies. Important features that distinguish automated docking methods from one another are the scoring (or energy) function and the strategy used to search for the lowest score. Ideally, one would want to computationally determine protein-ligand interactions by finding the global energy minimum within a complex energy binding landscape that contains local minima. The energy landscape would accurately account for hydrophobic packing, electrostatic interactions, hydrogen bonding, solvation / desolvation, and entropic terms. However, there is a trade-off between speed and accuracy that is inherent in any computational approach attempting to model molecular recognition.

AutoDock is an automated docking method that uses a grid-based force field to rapidly dock flexible ligands to a protein target (99). In grid-based methods, a threedimentional lattice of regularly spaced points is created around the binding site. A grid map is computed for each atom type in the ligand that is to be docked. Most gridbased methods allow the user to define the size and placemet of a grid, thereby allowing te extent of grid maps to range from a know protein active site to the entire macromolecule. Grids used during AutoDock runs are calculated prior to docking using the AutoGrid module. Within the AutoGrid parameter file, the user specifies coordinates of the grid center, the number of points in each dimension, and the spacing between points. Ligands are treated as sets of atoms. At each grid point, the interaction energy for a probe atom of the ligand with the surrounding macromolecule is computed using the energy function. During docking, the interaction energy of each ligand atom with the protein is computed by interpolating between the energy values stored at the grid points that are closest to the docked position of that atom.

The primary advantage of grid-based force fields is that pairwise calculations are performed prior to the docking simulation. During docking, the interaction energy between the ligand and protein is calculated, essentially, by summing atomic affinities of each ligand atom at its location with the grid map. There may be additional calculations, such as penalties for restricting rotatable bonds, internal interaction energy of the ligand, and interpolation between grid points, but none of these terms depend upon the number of atoms in the protein. Therefore, in grid-based methods, the interaction energy calculation for a ligand position is function of approximately order O(n), where n is the nuber of atoms in the ligand. The calculation done during docking using a grid-based method depends solely on the ligand, which generally contains significantly fewer atoms than the protein.

In AutoDock, the binding free energy is estimated as the sum of the energy of complex formation in a vacuum and the energy required to desolvate the surfaces on the ligand and the macromolecule that are buried when the complex formed. The AutoDock energy function uses a simplified molecular mechanics force field that is represented by a linear combination of van der Waals, hydrogen bonding, electrostatics, torsional restraint, and solvation energy terms:

$$\Delta G = \Delta G_{vdW} + \Delta G_{H-bond} + \Delta G_{elec} + \Delta G_{tor} + \Delta G_{solvation}$$
 Equation 1

This equation neglects conformational changes in the protein and covalent geometric distortion, but includes estimates for desolvation ( $\Delta G_{solvation}$ ) and entropic loss due to tosional restraint in the ligand ( $\Delta G_{tor}$ ).

The first three terms in the equation 1 are typical molecular mechanics interactions: van der Waals (steric fit), hydrogen bonding, and electrostatics. Van der Waals interaction energy between two atoms,  $\Delta G_{vdW}$ , is 12-6 Leonard-Jones potential.

In this model, when the distance between the two atoms is small, the atom repel each other. At longer distances, there is attractive force between the atoms. The Lennard-Jones energy function is at its minimum when two atoms are separated by their equilibrium internuclear separation.

Hydrogen bonding interaction energy,  $\Delta G_{H-bond}$ , is a model with a 12-10 Lennard-Jones potential. The 12-10 Lennard-Jones potential creates a deeper energy potential well for two atoms than the 12-6 potential. Thus, hydrogen bond reactions are shorter and stronger than van der Waals interactions.

Electrostatic interaction energy,  $\Delta G_{elec}$ , is a modeled as a distance-dependent Coulombic potential.

The fourth term,  $\Delta G_{tor}$ , is a measure of the unfavorable entropy due to restriction of ligand conformation and is proportional to the number of rotatable bonds. These four terms,  $\Delta G_{vdW}$ ,  $\Delta G_{H-bond}$ ,  $\Delta G_{elec}$  and  $\Delta G_{tor}$ , constitute the in vacuo ligand-binding energy.

The fifth term,  $\Delta G_{solvation}$ , is the most challenging to estimate in AutoDock. Ligand binding requires at least partial desolvation of the surfaces that become buried when the ligand-macromolecule complex forms. The energy required to desolvate these surfaces makes significant contributions to the overall binding energy. In AutoDock, the desolvation energy is computed for hydrophobic atoms. It is an approximation of the energy required when hydrophobic ligand atoms (e.g. carbon and aromatic carbon) desolvate protein atoms in a binding site.

For search strategies, the most common searching strategies for grid-based automated docking are Monte Carlo simulated annealing and the genetic algorithm. Both methods have been implemented in Autodock. For the genetic algorithm is initiated by creating a randomized population of ligand binding orientations. The binding orientations are doubled by crossover of the genes of the individuals, a process analogous to mating. Random point mutations are also introduced. The fitness, or interaction energy, of each ligand binding orientation is computed and is used as selection-criteria for survival to the next generation. Favorable binding orientations are identified through iterative generations of population doubling and survival selection. The Lamarkian genetic algorithm, a variant of the genetic algorithm, was the most efficient method of the search methods tested in AutoDock 3.0. The Lamarkian genetic algorithm was inspired by Jean Baptiste de Lamark's theory that phenotype changes occurring during an individual's lifetime can be passed on as genetic changes to its offspring. In the Lamarkian genetic algorithm, a global genetic algorithm search is combined with adaptive local minimization searching capability, and thus an individual can improve its binding orientation through local search and minimization and then pass on the improved binding orientation information.

## Part V. Virtual Screening

In medicine, biotechnology and pharmacology, drug discovery is the process by which drugs are discovered and/or designed. In the past, most drugs have been discovered either by identifying the active ingredients from traditional remedies or by serendipitous discovery. A new approach has been to understand how diseases and infections are controlled at the molecular and physiological level and to target specific entities based on this knowledge. The process of drug discovery involves the identification of candidates, synthesis, characterization, screening and assays for therapeutic efficacy. Once a compound has shown its value in these tests, it will begin the process of drug development prior to clinical trials.

Virtual screening (VS) has become an integral part of the drug discovery process in recent years. Related to the more general and long pursued concept of database searching, the term "virtual screening" is relatively new. Walters, *et al.* define virtual screening as "automatically evaluating very large libraries of compounds" using computer programs (100). As this definition suggests, VS has largely been a numbers game focusing on questions like how can we filter down the enormous chemical space of over  $10^{60}$  conceivable compounds to a manageable number that can be synthesized, purchased and tested. Although filtering the entire chemical universe might be a fascinating question, more practical VS scenarios focus on designing and optimizing targeted combinatorial libraries or vendor offerings. There are two broad categories of screening techniques: ligand-based and structure-based.

#### 1. Ligand-based Drug Design

Given a set of structurally diverse ligands that binds to a receptor, a model of the receptor can be build based on what binds to it. These are known as pharmacophore models. A candidate ligand can then be compared to the pharmacophore model to determine whether it is compatible with it and therefore likely to bind (101).

#### 2. Structure-based Drug Design

Structure-based virtual screening involves docking of candidate ligands into a protein target followed by applying a scoring function to estimate the likelihood that the ligand will bind to the protein with high affinity (102-103).

The main goal of a virtual screen is to come up with hits of novel chemical structures that yield a unique pharmacological profile. Thus, success of a virtual screen is defined in terms of finding interesting new scaffolds rather than many hits. Interpretations of VS accuracy should therefore be considered with caution. Low hit rates of interesting scaffolds are clearly preferable over high hit rates of already known scaffolds. Once the hit has been established with sufficient target potency and selectivity and favourable drug-like properties, one or two hits will then be proposed for drug development. The best of these is generally called the lead compound, while the other will be designated as the "backup".

## **CHAPTER III**

## **EXPERIMENTAL**

This chapter was separated in to 3 parts which are (i) molecular modeling (*in silico*) experiment, (ii) *in vitro* binding assay i.e. radioligand competition assay and sodium dodecyl sulfate polyacrylamide gel electrophoresis (SDS-PAGE) experiment and (iii) *in vivo* experiment.

#### A. Molecular Modeling (In Silico) Experiment

#### 1. Materials

- 1.1 Bluefish (computer for linux cluster at central computing resources of The Scripps Research Institute).
- 1.2 Software ChemOffice 8.0 for Windows (CambridgeSoft Corporation, USA)
- 1.3 Software AutoGrid and AutoDock 3.0.5 (The Scripps Research Institute)
- 1.4 National Cancer Institute (NCI) diversity set
- 1.5 Alpha-cobratoxin crystal structures (α-cbtx) 1CTX, 2CTX, 1YI5 (Protein Data Bank) (104)

#### 2. Methods

#### 2.1 Protein Template Preparation

A crystallographic structure of  $\alpha$ -cobratoxin (PDB entry code: 1CTX (8) and 2CTX (105) and  $\alpha$ -cobratoxin bound to acetylcholine binding protein (AChBP, PDB entry code: 1YI5 (13)), a soluble homolog of the extracellular binding domain of nicotinic acetylcholine receptor, were selected to prepare  $\alpha$ -cobratoxin active binding site template for docking. It was solved by X-ray diffraction techniques with a resolution of 2.8, 2.4 and 4.2 Å for 1CTX, 2CTX and 1YI5, respectively. The  $\alpha$ -cobratoxin (1CTX and 2CTX) and  $\alpha$ -cobratoxin/AChBP complex (1YI5) were used in

the virtual screening whereas only 1YI5 was selected for studying the rediocide binding mode. Prior to docking studies,  $\alpha$ -cobratoxin was read in and repeated using SwissPDBViewer before using AutoDock (ADTools version 1.3). In order to prepare the target protein as a template, the AChBP and crystallographic water were removed. Polar hydrogen atoms were added and Kollman charges were assigned to all atoms. The grid maps representing the  $\alpha$ -cobratoxin were calculated with AutoGrid. The dimentions of the grid were 60 x 88 x 86 grid points for 1YI5 and 86 x 60 x 88 for 1CTX and 2CTX, with a spacing of 0.375 Å between the grid points. The grid box was centered on the coordinates 118.056 x 160.484 x 64.991 for 1YI5 and 29.652 x 49.388 x 4.481 for 1CTX and 2CTX. The grip maps were computed using AutoGrid 3.0 for each protein model.

#### 2.1.1 Template Validation

The constructed  $\alpha$ -cobratoxin template from 1YI5 was validated. The Ser182 – Tyr 192 from AChBP, the binding site of  $\alpha$ -cobratoxin, was selected to be the control peptide. It was designed to have 9 rotatable bonds from the single bonds of the residues except amide bond. In order to validate the prepared  $\alpha$ -cobratoxin template, the Ser182 – Tyr 192 was docked back into the prepared template. If the result showed the largest cluster of docked conformations and the docked conformations were close to that of the crystal pose with root-mean-square deviation (RMSD) < 2 Å, the re-docking result indicated that the prepared  $\alpha$ -cobratoxin protein is a good model for docking studies.

## 2.2 Ligand Preparation

#### 2.2.1 NCI Diversity Set

The NCI diversity set is a reduced set of 1990 compounds selected from the original NCI-3D structural database for their unique pharmacophores (106). The diversity set was derived by others from the almost 140,000 compounds available on plates. The 71,756 compounds meeting this criterion were then reduced to the final set using the program Chem-X (Oxford Molecular Group). Chem-X uses defined centers (hydrogen bond acceptor, hydrogen bond donor, positive charge, aromatic, hydrophobic, acid, base) and defined distance intervals to create a particular finite set

of pharmacophores. The 3-point pharmacophores were used with the default settings, resulting in almost 1,000,000 possible pharmacophores. The Chem-X diverse subset generating function reads through a set of structures and for every structure, determines the acceptable conformations of that structure. For each acceptable conformation, it determines all the pharmacophores for that conformation. The pharmacophores for the current structure are compared to the set of all pharmacophores found in structures already accepted into the diverse subset. If the current structure has more than a preset number of new pharmacophores, it is added to the diverse subset. The requirements were set as 5 new pharmacophores and, additionally, 5 or fewer rotatable bonds. Because the selection procedure is order dependent, the order in which the structures were considered was randomized. This procedure resulted in the selection of 1990 compounds. For ligand preparation, All hydrogen were added and Gasteiger charges were assigned for each compound (107). Then the non-polar hydrogens were merged, aromatic carbons identified, and lastly, the rigid root and rotatable bonds were assigned via AutoTors (85).

#### 2.2.2 Rediocides (A & G)

The three-dimensional structures of rediocides were generated and cleaned up by the ChemDraw Ultra version 8.0.7 program. After cleaned up structure, the structure was minimized the energy by Chem3D Ultra version 8.0.3 using the MM2 method. The molecular dynamics was performed and the run parameters used were as follows: the step interval was 1.0 fs; the frame interval was 100 fs; the termination criterion was 10,000 steps; the target temperature was 300 kelvin; the heating/cooling rate was 1.000 kcal/atom/ps; and the job type was record every iteration. Every iteration was changed to pdb file. All hydrogens were added and Gasteiger charges were assigned to each pdb file. Then the non-polar hydrogens were merged, aromatic carbons were identified, and lastly, the rigid root and rotatable bonds were assigned via AutoTors.

#### 2.3 Molecular Docking

The program AutoDock3.0.5 (85) from The Scripps Research Institute, USA that used Larmarkian genetic algorithm for docking flexible ligands into protein

binding sites to explore the full range of ligand conformational flexibility with the rigid protein was used for docking study. The AutoDock run parameters used were as follows: the number of GA runs were 100; the population size was 150; the maximum number of energy evaluations was increased to 10,000,000 per run; and the maximum number of generations was 27,000. The crossover rate was set to 80% and the mutation rate to 2% with Cauchy distribution parameters a=0 and b=1. The jobs were run on Bluefish at The Scripps Research Institute. Final docked conformations were clustered using a tolerance of 2 Å root-mean-square deviation (RMSD). The final docked structure, RMSD from the bound crystal structure and docked energy were all used to analyse its interaction with the active site.

## 2.3.1 Binding Mode of Rediocides (A & G)

Each conformation of rediocides from the molecular dynamic was docked with  $\alpha$ -cobratoxin template prepared from 1YI5. After docking, the lowest docking energy of each conformation (101 conformations) was ranked from lowest docking energy to highest docking energy. The conformation that bound in the binding site of  $\alpha$ -cobratoxin with lowest energy in the highest custer was selected. The amino acid residues of AChBP in the sphere radius at 5 Å from rediocide were shown. The distance of hydrogen bond between rediocide and interacted amino acid residues was measured.

#### 2.3.2 Virtual Screening

The  $\alpha$ -cobratoxin template derived from 1CTX, 2CTX and 1YI5 were used for molecular docking against 1990 ligands in NCI diersity set. The virtual screening was performed on Bluefish by the script (a 64-bit 576 processor LINUX cluster). After docking, the lowest docking energy of each ligand binding to each  $\alpha$ -cobratoxin template (1YI5, 1CTX and 2CTX) was ranked from lowest docking energy to highest docking energy. The top 175 compounds from 1990 compounds with the best docking energies from each protein template were selected. The selected compounds (3 x 175 compounds) that matched 3 proteins were selected for the experiment.

## B. In Vitro Binding Assay

# 1. Materials

## Equipment

Analytical balance

Micropipettes 0.5-10  $\mu$ l

Micropipettes 20-200  $\mu$ l

Micropipettes 100-1000  $\mu$ l

Micropipettes 2-20  $\mu$ l

Mildmixer SI-36

MINI protein<sup>®</sup> 3

Power supply

Vortex mixer

Water bath

Precisa 240A and Precisa 4000C (Oerlikon, Switzerland) LS 6500 liquid scintillation counter Beckman Scientific (California, USA) High Tech Lab (Warsaw, Poland) High Tech Lab (Warsaw, Poland) High Tech Lab (Warsaw, Poland) Genex Beta, UK Taitec Corporation (Saitama, Japan) BIO-RAD (California, USA) BIO-RAD and e-power 100 (California, USA) Vortex-genie (New York, USA) Neslab Instrument (New Hampshire, USA)

## Chemicals

AChBP from Aplysia californica	University of California, San Diego, USA
AChBP from Bolinus truncatus	University of California, San Diego, USA
AChBP from Lymnaea stagnalis	University of California, San Diego, USA
Acrylamide	BIO-RAD (California, USA)
Alpha-cobratoxin (C6903)	Sigma (Missouri, USA)
Ammonium persulfate	BDH Laboratory Supplies
	(Poole, England)
Anti-mouse SPA	Amersham Biosciences
	(New Jersey, USA)
Bromophenol blue	Sigma (Missouri, USA)
Coomassie Blue R-250	USB Corporation (Ohio, USA)
Crude cobra venom	Thai Red Cross Society
	(Bangkok, Thailand)

Dimethyl sulfoxide (DMSO)	E. Merck (Darmstadt, Germany)		
Glacial acetic acid	Lab-Scan (Bangkok, Thailand)		
Glycerol	Carlo Erba Reagenti (Milan, Italy)		
Glycine	Fisher Scientific (Leicestershire, UK)		
[ <sup>3</sup> H] epibatidine	University of California, San Diego, USA		
Hits from NCI diversity set	National Cancer Institute, USA		
Hydrochloric acid (HCl)	Lab-Scan (Bangkok, Thailand)		
$[^{125}I] \alpha$ -bungarotoxin	University of California, San Diego, USA		
Low range protein standard	BIO-RAD (California, USA)		
2-mercaptoethanol	BIO-RAD (California, USA)		
Methanol	Lab-Scan (Bangkok, Thailand)		
Methyllycaconitine (MLA)	Sigma (Missouri, USA)		
Monoclonal anti-FLAG M2	Sigma (Missouri, USA)		
N,N,N',N'-tetramethylene-ethylened	liamine BIO-RAD (California, USA)		
N, N'-methylene-bisacrylamide	Serva (Heidelberg, Germany)		
Precision plus protein standard #161	0374 BIO-RAD (California, USA)		
Rediocides (A & G)	Bansomdejchaopraya Rajabhat Univ.		
	(Bangkok, Thailand)		
Sodium lauryl sulfate	Scharlau (Barcelona, Spain)		
Tris (hydroxymethyl)-aminomethane	Fisher Scientific (Leicestershire, UK)		
Tween 80	Sigma (Missouri, USA)		

## 2. Methods

## 2.1 Radioligand Competition Assay

## 2.1.1 Screening Test

The assay was performed to determine the fraction of  $[{}^{3}\text{H}]$  epibatidine. An adaptation of a scintillation proximity assay was used to determine the apparent  $K_{d}$  value (108). The test compound 10  $\mu$ M, methyllycaconitine 15  $\mu$ M (positive control) and phosphate buffer 0.1 M, pH 7.0 (negative control) were added to AChBP (~500 pM binding sites), polyvinyltoluene anti-mouse SPA scintillation beads (0.1 mg/ml), monoclonal anti-FLAG M2 antibody from mouse. Methyllycaconitine is a non-specific binding ligand to AChBP. The mixture was incubated at room temperature

for 2 hours. After that, agonist [<sup>3</sup>H] epibatidine (5 nM for Lymnaea stagnalis and Bolinus truncatus, 20 nM for Aplysia californica) were added and allowed to equilibrate at room temperature for a minimum of 2 h.  $[^{3}H]$  epibatidine that bound on the bead was measured in a LS 6500 liquid scintillation counter. The result was calculated in the fraction of [<sup>3</sup>H] epibatidine between sample and the control (total substracts with the binding from negative control background from methyllycaconitine). The ligand that has the fraction of  $[{}^{3}H]$  epibatidine lower than 0.6 was selected for the determination of  $K_d$  values in section 2.1.2.

## 2.1.2 Determination of *K*<sub>d</sub>

## 2.1.2.1 [<sup>3</sup>H] epibatidine

The test compound, methyllycaconitine 15  $\mu$ M (positive control) and phosphate buffer 0.1 M, pH 7.0 (negative control) were added to AChBP (~500 pM binding sites), polyvinyltoluene anti-mouse SPA scintillation beads (0.1 mg/ml), monoclonal anti-FLAG M2 antibody from mouse. The mixture was incubated at room temperature for 2 hours. After that, agonist [<sup>3</sup>H] epibatidine (5 nM for *Lymnaea stagnalis* and *Bolinus truncatus*, 20 nM for *Aplysia californica*) were added and allowed to equilibrate at room temperature for a minimum of 2 h and measured the [<sup>3</sup>H] epibatidine that bound on the bead in a LS 6500 liquid scintillation counter. The result was calculated in fraction of [<sup>3</sup>H] epibatidine. The result was calculated in fraction of [<sup>3</sup>H] epibatidine. The data obtained were normalized, fit to a sigmoidal dose-response curve (variable slope), and the *K<sub>d</sub>* calculated from the observed EC<sub>50</sub> value using GraphPad Prism version 4.02 for Windows (San Diego, CA).

## 2.1.2.2 [<sup>125</sup>I] $\alpha$ -bungarotoxin

The test compound, methyllycaconitine 15  $\mu$ M (positive control) and phosphate buffer 0.1 M, pH 7.0 (negative control) were added to AChBP (~500 pM binding sites), polyvinyltoluene anti-mouse SPA scintillation beads (0.1 mg/ml), monoclonal anti-FLAG M2 antibody from mouse. The mixture was incubated at room temperature for 2 hours. After that, antagonist [<sup>125</sup>I]  $\alpha$ -bungarotoxin (10 nM for *Lymnaea stagnalis* and 20 nM for *Aplysia californica*) were added and allowed to equilibrate at room temperature for a minimum of 2 h and measured the [<sup>125</sup>I]  $\alpha$ - bungarotoxin that bound on the bead in a LS 6500 liquid scintillation counter. The result was calculated in fraction of  $[^{125}I] \alpha$ -bungarotoxin. The data obtained were normalized, fit to a sigmoidal dose-response curve (variable slope), and the  $K_d$  calculated from the observed EC<sub>50</sub> value using GraphPad Prism version 4.02 for Windows (San Diego, CA). The effective concentration (EC<sub>50</sub>) is defined as the concentration of a compound which bound to AChBP halfway between the baseline and maximum.

## 2.2 Sodium Dodecyl Sulfate Polyacrylamide Gel Electrophoresis (SDS-PAGE)

## 2.2.1 Solutions

2.2.1.1 Stock Solution

**2** M Tris-HCl (pH 8.8). Distilled water (50 ml) was added into 24.2 g of Tris base. Afterthat, the concentrated HCl was added slowly and adjusted to pH 8.8. The volume was adjusted with distilled water to 100 ml.

**1 M Tris-HCl (pH 6.8).** Distilled water (50 ml) was added into 12.1 g of Tris base. Afterthat, the concentrated HCl was added slowly to pH 6.8. The volume was adjusted with distilled water to 100 ml.

10% SDS (w/v). Distilled water was added in 10 g of SDS to make 100 ml.

**50% Glycerol.** Distilled water (50 ml) was added into 50 ml 100% glycerol.

**1% Bromophenol Blue, (w/v).** Distilled water (10 ml) was added into 100 mg of bromophenol blue. Afterthat, the mixture was stirred until dissolved and filtered to remove the aggregated dye.

**Coomassie Gel Stain.** One liter of Coomassie gel stain was prepared by adding 1g of Coomassie Blue R-250, 100 ml of glacial acetic acid and 450 ml of methanol into 450 ml of distilled water.

**Coomassie Gel Destain** One liter of Coomassie gel destain was prepared by adding 100 ml of glacial acetic acid and 100 ml of methanol into 800 ml of distilled water.

#### 2.2.1.2 Working Solution

**Solution A (Acrylamide Stock Solution).** Distilled water was added in 29.2 g of acrylamide and 0.8 g of N, N'-methylene-bisacrylamide to make 100 ml. The solution was covered with parafilm and stirred under hood until acrylamide powder is completely dissolved. The solution can be stored for months in the refrigerator.

**Solution B (4x Separating Gel Buffer).** Distilled water (21 ml) and 4 ml of 10% SDS were added into 75 ml of 2 M Tris-HCl (pH 8.8). The solution can be stored for months in the refrigerator.

**Solution C (4x Stacking Gel Buffer).** Distilled water (46 ml) and 4 ml of 10% SDS were added into 50 ml of 1 M Tris-HCl (pH 6.8). The solution can be stored for months in the refrigerator.

**10% Ammonium Persulfate.** Distilled water (1 ml) was added into 0.1 g of ammonium persulfate. The solution can be stored for months in a capped tube in the refrigerator.

**Electrophoresis Buffer.** Distilled water was added in Tris (hydroxymethyl)aminomethane (3 g), glycine (14.4 g) and SDS (1 g) to make 1 liter. The pH should be approximately 8.3 and the solution can be stored at room temperature.

**5x Sample Buffer.** Ten milliliter of 5x Sample Buffer was prepared by adding 0.6 ml of 1 M Tris-HCl (pH 6.8), 5 ml of 50% glycerol, 2 ml of 10% SDS, 0.5 ml of 2-mercaptoethanol and 1 ml of 1% bromophenol blue into 0.9 ml of distilled water. The solution can be stored for weeks in the refrigerator or for months at -20 °C.

## 2.2.2 Effect of Rediocides (A & G) on *a*-Cobratoxin and AChBP

2.2.2.1 Effect of Rediocides on  $\alpha$ -Cobratoxin

SDS-PAGE (109) was performed with 15% acrylamide gels using MINI protein<sup>®</sup> 3 as marker. The separating gel was made by mixing acrylamide solution, 4x Separating Gel Buffer, de-ionized water, ammonium persulfate and TEMED (N,N,N',N'-tetramethylene-ethylenediamine) in the particular ratio as shown in Table 5. The separating gel was subjected to space between the sandwich glasses, overlayed with de-ionized water and allowed to polymerize. De-ionized water was removed

after the separating gel was polymerized and stacking gel was set by mixing of acrylamide solution, 4x stacking gel buffer, de-ionized water, 10% ammonium persulphate and TEMED (Table 7). Mixed stacking gel was poured on top of the separating gel and the comb was inserted. After the stacking gel was completely polymerized, the comb was removed, and the whole unit of the gel and the glass-plate sandwich were attached to the electrophoresis chamber. The tank was filled with electrophoresis buffer both top and bottom of the tank.

The rediocides (A & G) samples were prepared by mixing 5x sample buffer with venom solution to bring the final concentration of sample buffer to 1x and then the samples were boiled for 5 minutes and centrifuged. The supernatant were loaded into each well and electrophoresis was performed at 150 volts until the dye front reached to the end of the gels. After electrophoresis, the gels were stained with Coomassie blue R-250 (0.001% in glacial acetic acid: methanol: water, 10:45:45, v/v) and then destained with 10% methanol and 10% glacial acetic solution until the gel was clear.

Ingradiants	Separating gel		Staalving gol	
ingreutents	15% gel	20% gel	Stacking gei	
Solution A (acrylamide stock solution)	5 ml	6.66 ml	0.67 ml	
Solution B (4x separating gel buffer)	2.5 ml	2.5 ml	-	
Solution C (4x stacking gel buffer)	-	-	1 ml	
De-ionized water	2.5 ml	0.84 ml	2.3 ml	
10% Ammonium persulfate	50 µl	50 µl	25 µl	
TEMED	5 µl	5 µl	10 µl	

Table 7. The composition of separating gel and stacking gel

## 2.2.2.2 Effect of Rediocides on $\alpha$ -Cobratoxin and AChBP

SDS-PAGE was performed again with AChBP,  $\alpha$ -cobratoxin and rediocides but the samples were dissolved in 0.1% Tween 80 and was not boiled before loading to each well. nAChBP and  $\alpha$ -cobratoxin were mixed for 10 minutes before adding the sample. The sample mixtures were loaded into each gel and electrophoresis was performed at 100 volts until the marker's dye front reached to the end of the gels. After electrophoresis, the gels were stained with Coomassie blue R-250 (0.001% in glacial acetic acid: methanol: water, 10:45:45, v/v) and then destained with 10% methanol and 10% glacial acetic solution until the gel was clear.

## 2.2.3 Effect of 3 Hits on *a*-Cobratoxin and AChBP

## 2.2.3.1 Effect of 3 Hits on $\alpha$ -Cobratoxin

SDS-PAGE was performed with 20% acrylamide gels using MINI protein<sup>®</sup> 3 (BIO-RAD). The separating gel was made by mixing acrylamide solution, 4x Separating Gel Buffer, SDS, de-ionized (DI) water, ammonium persulfate and TEMED (N,N,N',N'-tetramethylene-ethylenediamine) in the particular ratio as shown in Table 5.

The samples were dissolved in 0.1% Tween 80, boiled for 5 minutes and centrifuged. The samples were loaded into each well and electrophoresis was performed at 100 volts until the dye front reached to the end of the gels. After electrophoresis, the gels were stained with Coomassie blue R-250 (0.001% in glacial acetic acid: methanol: water, 10:45:45, v/v) and then destained with 10% methanol and 10% glacial acetic solution until the gel was clear.

2.2.3.2 Effect of 3 Hits on  $\alpha$ -Cobratoxin and AChBP

SDS-PAGE was performed again with nAChBP,  $\alpha$ -cobratoxin and 3 hits but the samples were not boiled before loading to each well. The sample mixtures were loaded into each gel and electrophoresis was performed at 100 volts until the marker's dye front reached to the end of the gels. After electrophoresis, the gels were stained with Coomassie blue R-250 (0.001% in glacial acetic acid: methanol: water, 10:45:45, Fac. of Grad. Studies, Mahidol Univ.

v/v) and then destained with 10% methanol and 10% glacial acetic solution until the gel was clear.

## C. In Vivo Experiment

## 1. Materials

## Equipment

Analytical balance Micropipettes  $0.5-10 \mu l$ Micropipettes  $2-20 \mu l$ Micropipettes  $20-200 \mu l$ Micropipettes  $100-1000 \mu l$ Triple beam balance Ultrasonic bath Vortex mixer

## Chemicals

Alpha-cobratoxin (C6903) NSC 42258 NSC 121865 NSC 134754 Rediocides (A & G) Precisa 240A (Oerlikon, Switzerland) High Tech Lab (Warsaw, Poland) High Tech Lab (Warsaw, Poland) High Tech Lab (Warsaw, Poland) Genex Beta, UK OHAUS (Boston, USA) Brason 2210 (Colorado, USA) Vortex-genie (New York, USA)

Sigma (Missouri, USA) National Cancer Institute, USA National Cancer Institute, USA National Cancer Institute, USA Bansomdejchaopraya Rajabhat Univ. (Bangkok, Thailand) Sigma (Missouri, USA)

National Laboratory Animal Center, Mahidol University

## 2. Methods

Tween 80

Animals

ICR mice

## 2.1 Determination of Median Lethal Dose

The median lethal dose (LD<sub>50</sub>) is defined as the least amount of  $\alpha$ -cobratoxin which injected intramuscularly through the leg muscle of the mice that resulted in the death of half amount of mice within 24 h. ICR mice weighing 20-30 g were kept in a

temperature-controlled room on a 12/12-h cycle. The mice were divided into 10 groups of 6 animals in each. One group served as a control group receiving vehicle (sterile water) only. The other 9 groups were injected with  $\alpha$ -cobratoxin. Nine concentrations of the  $\alpha$ -cobratoxin solutions were prepared and diluted for intramuscular injection (i.m.) at 0.01, 0.1, 0.14, 0.17, 0.18, 0.2, 0.3, 0.4 and 0.5 mg/kg in 0.1 ml of sterile water. The number of deaths occurring within 24 h were recorded, LD<sub>50</sub> was calculated from the plot between the concentration of  $\alpha$ -cobratoxin and % death.

## 2.2 Effect of Rediocides and 3 Hits on Mice

## 2.2.1 The Protection Activity

In the protocol using rediocides or hits as protecting agent, rediocides or hits were injected to the mice before the injection of  $\alpha$ -cobratoxin. The sample solution, in 0.05 ml of 0.3% Tween 80, were prepared in varying concentrations. Each dose was injected to the mice intravenously (i.v.) at tail vein. The control group was injected with only the vehicle, 0.3% Tween 80. After 30 minutes,  $\alpha$ -cobratoxin at 3LD<sub>50</sub> dose in 0.1 ml of sterile water was injected (i.m.) into the mice. The survival time and number of deaths occurring within 24 hours was recorded. The results were expressed as the mean ± SD. Statistical comparisons were calculated using Student's unpaired *t*-*test*, with a *p* value < 0.05 indicating significance.

## 2.2.2 The Inhibition Activity

In the protocol using rediocides or hits as antitoxin treatment, rediocides or hits were injected after the injection of  $\alpha$ -cobratoxin.  $3LD_{50}$  of  $\alpha$ -cobratoxin in 0.1 ml of sterile water was injected (i.m) before the injection of sample in 0.05 ml of 0.3% Tween 80 (i.v). The survival time and number of deaths occurring within 24 hours was recorded. The results were expressed as the mean  $\pm$  SD. Statistical comparisons were calculated using Student's unpaired *t-test*, with a *p* value < 0.05 indicating significance.

#### 2.3 Dixon's Q Test (110)

In a set of replicate measurements of a physical or chemical quantity, one or more of the obtained values may differ considerably from the majority of the rest. In this case there is always a strong motivation to eliminate those deviant values and not to include them in any subsequent calculation (e.g. of the mean value and/or of the standard deviation). This is permitted only if the suspect values can be "legitimately" characterized as outliers. Usually, an outlier is defined as an observation that is generated from a different model or a different distribution than was the main "body" of data. Although this definition implies that an outlier may be found anywhere within the range of observations, it is natural to suspect and examine as possible outliers only the extreme values. The rejection of suspect observations must be based exclusively on an objective criterion and not on subjective or intuitive grounds. This can be achieved by using statistically sound tests for "the detection of outliers".

The Dixon's Q-test is the simpler test of this type and it is usually the only one described in textbooks of Analytical Chemistry in the chapters of data treatment. This test allows us to examine if one (and only one) observation from a small set of replicate observations (typically 3 to 10) can be "legitimately" rejected or not. Q-test is based on the statistical distribution of "subrange ratios" of ordered data samples, drawn from the same normal population. Hence, a normal (Gaussian) distribution of data is assumed whenever this test is applied. In case of the detection and rejection of an outier, Q-test cannot be reapplied on the set of the remaining observations.

The N values comprising the set of observations under examination are arranged in ascending order:  $x_1 < x_2 < ... < x_N$ . The statistic experimental Q-value  $(Q_{exp})$  is calculated. This is a ratio defined as the difference of the suspect value from its nearest one divided by the range of the values (Q: rejection quotient). Thus, for testing  $x_1$  or  $x_N$  (as possible outliers), use the following  $Q_{exp}$  values:

$$Q_{exp} = \frac{X_2 - X_1}{X_n - X_{n-1}}$$
  $Q_{exp} = \frac{X_n - X_{n-1}}{X_n - X_1}$ 

The obtained  $Q_{exp}$  value is compared to a critical Q-value ( $Q_{crit}$ ) found in Table 8. This critical value should correspond to the confidence level (CL) we have decided

to run the test (usually: CL=95%). If  $Q_{exp} > Q_{crit}$ , then the suspect value can be characterized as an outlier and it can be rejected, if not, the suspect value must be retained and used in all subsequent calculations.

Ν	Q <sub>crit</sub> (CL : 90%)	Q <sub>crit</sub> (CL : 95%)	Q <sub>crit</sub> (CL : 99%)
3	0.941	0.970	0.994
4	0.765	0.829	0.926
5	0.642	0.710	0.821
6	0.560	0.625	0.740
7	0.507	0.568	0.680
8	0.468	0.526	0.634
9	0.437	0.493	0.598
10	0.412	0.465	0.568

Table 8.	Critical	values of Q	
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## CHAPTER IV

## **RESULTS AND DISCUSSION**

Alpha-cobratoxin ( $\alpha$ -cbtx), a member of the long  $\alpha$ -neurotoxin family, is obtained from the venom of *Naja kaouthia* (previously called *Naja naja siamensis*). Cbtx has 71 amino acid residues and 5 disulfide bridges. It consists of three fingerlike loops: loop I, II and III. Alpha-cobratoxin blocks nerve transmission by binding to the nicotinic acetylcholine receptor (nAChR) on the postsynaptic membranes of skeletal muscle and/or neurons and causes paralysis by preventing acetylcholine binding to nAChR. The only remedy for cobra bite is a monospecific antivenom produced by Queen Saovabha Memorial Institute, The Thai Red Cross Society. The horse antivenom against *Naja kaouthia* is very difficult to produce, expensive and in short supply. Only 20% of the horses immunized with Thai cobra venom produce adequate neutralizing activity for antivenom production.

Application of medicinal plants with anti-snake venom activities might be useful as first aid treatment for victims of snake bite. There are about forty plants in Thailand reported to be effective in treating snake bites. One of the potential plants among these Thai plants was *Trigonostemon rediocides* (Kurz) Craib belonging to the Euphorbiaceae family, which is known as "Lotthanong" in Thai. The herb is particularly effective in treating snake bites especially against snake neurotoxins. The daphnanediterpenoids (rediocides A and G) have been found as the major components in the roots of this plant.

In search of an antidote of cobra venom by preventing the  $\alpha$ -cobratoxin from binding to the nicotinic acetylcholine receptor, the molecular docking was carried out to investigate the binding mode of rediocides to  $\alpha$ -cobratoxin. The aim is to find the mechanism of rediocides against cobra venom at the molecular level to promote the utilization of medicinal plant in clinical use. In addition to the investigation of the molecular mechanism of rediocides, virtual screening was accomplished by using  $\alpha$ - cobratoxin as template and docked with 1990 compounds in NCI diversity set. The objective of virtual screening is to discover the novel scaffold of new lead for drug design. The *in vitro* SDS-PAGE and binding assays as well as *in vivo* anti-toxin tests were performed to support the *in silico* results. The findings will be useful for patients who are bitten by snakes and will be useful for medical treatment in Thailand.

## A. Molecular Modeling (In Silico) Experiment

Molecular docking was carried out to investigate the docking energy and binding mode of rediocides bound to neurotoxin or  $\alpha$ -cobratoxin ( $\alpha$ -cbtx) by using AutoDock program version 3.0.5 (22). Among three reported crystal structures of the  $\alpha$ -cobratoxin from the *Naja naja siamensis* or *Naja kaothia*, 1CTX (8) and 2CTX (92) were unbound  $\alpha$ -cobratoxin crystals and 1YI5 (13) was  $\alpha$ -cobratoxin co-crystallized with pentameric acetylcholine binding protein (AChBP) from snail (Lymnaea stagnalis, Ls). The AChBP from Ls mimics the binding domain of nicotinic acetylcholine receptor (nAChR) which mediates excitatory transmission at the neuromuscular junction and in the central and peripheral nervous systems. The binding mode of  $\alpha$ -cobratoxin at AChBP was displayed in Figure 12. The unbound  $\alpha$ cobratoxin crystal structures (1CTX and 2CTX) have the structures of 71 amino acid residues while the  $\alpha$ -cobratoxin in the bound crystal structure (1YI5) has 68 amino acid residues with missing Arg68 side chain. The difference in number of amino acid residues is due to disorder of these three residues. The missing residues are at the tip of  $\alpha$ -cobratoxin loop III and was found to contribute weakly to  $\alpha$ -cobratoxin binding (13). The closest distance between the missing amino acids and AChBP was 8 Å.



Figure 12. The 1YI5 bound crystal between α-cobratoxin (chain F-J) and AChBP (chain A-E); (PDB: 1YI5). The binding mode of α-cobratoxin at AChBP showing amino acid residues from α-cobratoxin (chain H) bound the binding site of AChBP: chain C (yellow); chain D (indigo blue) (12-14).

In the experiment, 1YI5, 1CTX and 2CTX were constructed and used as templates for virtual screening. The area on AChBP that bound to  $\alpha$ -cobratoxin from 1YI5 (Figure 13) was used as a control peptide to validate the  $\alpha$ -cobratoxin template. Chain H was selected to be the  $\alpha$ -cobratoxin template, while Ser182 – Tyr 192 of chain C of AChBP was brought for the preparation of the control peptide in the validation. Nine rotatable bonds in the control peptide were allowed to rotate in the docking and these eleven residues were re-docked to validate the prepared  $\alpha$ -cobratoxin active binding site. The missing Arg68 side chain of chain H in the crystal structure of 1YI5 was reconstructed by SwissPDBViewer, the missing atoms in the side chain were a reasonable conformation. The conformation of Arg68 side chain was in the same as in the unbound crystal structures (1CTX and 2CTX). Chain H was selected from 1YI5 because it was only one chain that has complete amino acid residues comparing to other chains of  $\alpha$ -cobratoxin.



**Figure 13**. The control peptide for the  $\alpha$ -cobratoxin template validation. It composed of eleven amino acid residues from chain C of AChBP (1YI5). The arrows show 9 rotatable bonds.

The validation result of redocking the control peptide to the  $\alpha$ -cobratoxin template shows that 100% of the docked conformations grouped into a single cluster of 2.0 Å RMSD clustering tolerance (Figure 14). The docked orientation in this single cluster is close to that of the crystal pose with RMSD of 1.1020 Å. The residues interacted with the control peptide were the same amino acids in the crystal pose (Figure 15). The same interacted amino acids of the AChBP from the crystal structure and the control peptide are Ile9, Asp27, Arg33, Gly34, Lys35, Val37 and Arg68 (Table 9) (13). The validation by re-docking indicated that the prepared  $\alpha$ -cobratoxin
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template is a good model for docking studies of rediocide binding modes and the diversity set.



Figure 14. The clusters of docked control peptide in the same 3D structure (RMSD of 2 Å), docking energy is in kcal/mol.





**Figure 15**. Superposition of crystallographic Ser182 – Tyr 192 of chain C from 1YI5 (purple) and docked orientation of control peptide (orange) (A-C). The amino acid residues of  $\alpha$ -cobratoxin interacted with control peptide (D) and crystal pose (E) with distance < 4 Å.

Crystal pose	(1YI5)	<b>Control peptide (validation)</b>		
AChBP - Cbtx	Distance (Å)	Control <sup>a</sup> - Cbtx	Distance (Å)	
CO – HN (Ile9)	3.594	CO –HN (Ile9)	3.330	
CO –HN (Ile9)	3.787			
OH – OC (Asp27)	1.565	OH – OC (Asp27)	1.993	
OH – OC (Asp27)	2.578	OH – OC (Asp27)	2.220	
NH – OC (Arg33)	3.484	NH – OC (Arg33)	3.433	
NH – OC (Gly34)	3.666	NH – OC (Gly34)	3.941	
NH – OC (Lys35)	2.033	NH – OC (Lys35)	2.096	
OH – OC (Lys35)	2.838			
CO – HN (Val37)	2.476	CO – HN (Val37)	1.821	
		OH – OC (Val37)	3.400	
		NH – OC (Val37)	3.702	
		CO – HO (Thr67)	3.520	
CO – HN (Arg68)	3.008	CO – HN (Arg68)	1.903	
OH – OC (Arg68)	3.526	OH – OC (Arg68)	3.474	

**Table 9**. Amino acid residues of  $\alpha$ -cobratoxin interacted with the control peptide and<br/>the AChBP in the crystal pose (1YI5)

<sup>a</sup>Eleven residues with 9 rotatable bonds from AChBP that bind to  $\alpha$ -cobratoxin.

## 1. Binding Mode of Rediocides (A & G)

Rediocides A and G are daphnanediterpenoids from *Trigonostemon rediocides*. The differences between the structures of rediocides A and G are (i) a 5-membered ring with a methyl side chain for rediocide A and a 6-membered ring for rediocide G on the backbone structure and (ii) the isobutyl ester side chain for rediocide A and phenyl ring for rediocide G (Figure 16). The docking study on rediocides and  $\alpha$ - cobratoxin interaction was performed to investigate the binding modes of the rediocides. The docking simulation techniques have been carried out using AutoDock program. Because of the complex structures of rediocides, a molecular dynamics was assigned to generate conformations. Each of the 101 conformations of rediocide A and rediocide G was designed to have 11 and 10 rotatable bonds respectively and docked to the  $\alpha$ -cobratoxin template with the AutoDock3 program. The 100 independent dockings were run for each ligand, the docked ligand generally converged to a small number of different positions or conformers or poses. The pose in top cluster with greatest frequency of occurrence is the most probable orientation, or conformer that bound to the active binding site with most favorable docking energy. After each of 100 docking runs, the conformations that have the lowest docked energy with the same 3D structure (RMSD < 2 Å) were clustered as shown in Figure 17 and 18. There are 40 and 38 conformers in the first cluster of rediocide A and rediocide G, respectively (Table 10).



Figure 16. Structures of rediocides A (left) and G (right). The rectangular boxes show the differences in structures between rediocide A and rediocide G and the arrows show the assigned rotatable bonds.



Figure 17. Clusters of docked rediocides in the same 3D structure (RMSD < 2Å),</li>(A) rediocide A and (B) rediocide G, docking energy is in kcal/mol.

Results and Discussion / 62



Figure 18. The bound conformations of rediocides in the active site of α-cobratoxin. The structures of rediocide A (blue) and rediocide G (pink) before docking (A) and after docking (B). The msms models of α-cobratoxin molecule template prepared from 1YI5 bind with rediocide A (C) and rediocide G (D).

Ligand	Cluster number	%Member in cluster	Docking energy (Kcal/mol)
Rediocide A	$1^{a}$	40	-14.17
Rediocide G	$1^{a}$	38	-14.14
Control peptide <sup>b</sup>	$1^a$	100	-12.20

**Table 10**. Cluster analysis from docking  $\alpha$ -cobratoxin with rediocides A and G

<sup>a</sup>The cluster with lowest docking energy of a total 100 runs.

<sup>b</sup>Alpha-cbtx binding site at AChBP (Ser182-Tyr192 from chain C of 1YI5).

The binding sites of rediocide A and rediocide G at  $\alpha$ -cobratoxin were in the same general location and preferably shifted to loop II of the  $\alpha$ -cobratoxin which provided ample space to better accommodate this larger molecule. Table 11 and Figure 19 show the interacted amino acid residues between functional groups of rediocides and amino acid residues surrounding their binding sites to  $\alpha$ -cobratoxin. Hydrogen bonds were formed directly between the rediocides and amino acid residues at the binding site of  $\alpha$ -cbtx. The significant positions in rediocide A were Asp27, Phe29, Arg33, Gly34, Lys35 and Val37 with the distance of 1.8-4.0 Å and those of rediocide G were Asp27, Phe29, Arg33, Lys35 and Val37 with 1.9-3.9 Å. The Phe29 and Arg33 side chains extending at the tip of  $\alpha$ -cobratoxin are well positioned for establishing hydrophobic and aromatic interactions. The amino acid residues from  $\alpha$ cobratoxin that bind AChBP representing the acetylcholine receptor were reported to be Thr6, Pro7, Ile9 in loop I, Trp25, Cys26, Asp27, Ala28, Phe29, Cys30, Ser31, Ile32, Arg33, Gly34, Lys35, Arg36, Val37 in loop II and Phe65, Arg68 for C-terminus as shown in Figure 12 (12-14). The rediocide bound location was in the same area that the  $\alpha$ -cobratoxin bound with nicotinic acetylcholine receptor. Thus, the active binding site of  $\alpha$ -cobratoxin is occupied in the presence of rediocides and consequently  $\alpha$ -cobratoxin cannot bind with the acetylcholine receptor. Therefore, the in silico experiment can reveal the detoxification mechanism of cobra venom at the molecular level.

Rediocide A - Cbtx	Distance (Å)	Rediocide G - Cbtx	Distance (Å)
OH – OC (Asp27)	2.634	OH – OC (Asp27)	2.589
OH – OC (Phe29)	2.757	OH – OC (Phe29)	3.258
OH – OC (Phe29)	3.924	CO – HN (Arg33)	2.805
CO – HN (Arg33)	2.556	O – HN (Gly34)	3.804
OH – OC (Lys35)	3.894	OH – OC (Gly34)	2.239
O – HN (Val37)	2.730	O – HN (Val37)	2.697
OH – OC (Val37)	1.878	OH – OC (Val37)	1.946
OH – OC (Val37)	2.284	OH – OC (Val37)	1.907

Table 11. Amino acid residues of  $\alpha$ -cobratoxin interacted with rediocides



**Figure 19**. The bound conformations of rediocides in the active site of  $\alpha$ -cobratoxin (A and B). The ribbon model shows the backbone of  $\alpha$ -cobratoxin catalytic domain (1YI5) with all interacting amino acid residues shown as stick and ball models, rediocide A (blue), rediocide G (pink). The amino acid residues of  $\alpha$ -cobratoxin interacted with rediocide A (C) and rediocide G (D) with distance less than 4 Å.

#### 2. Virtual Screening

Virtual screening was conducted by docking the  $\alpha$ -cobratoxin template derived from 3 crystal structures (1CTX, 2CTX and 1YI5) against 1990 ligands in NCI diversity set. The virtual screening was performed on Bluefish (a 64-bit 576 processor LINUX cluster) by the unix script command for virtual screening (110). After docking, the docking energy of each ligand that bind to each  $\alpha$ -cobratoxin template (1YI5, 1CTX and 2CTX) was ranked from lowest docking energy to highest docking energy. The top 175 compounds from each  $\alpha$ -cobratoxin templates were in Table 12. The lowest docking energy for 1YI5, 1CTX and 2CTX were -13.86, -13.59 and -12.74 kcal/mol, respectively and the highest docking energy were -10.11, -10.12 and -9.52 kcal/mol for 1YI5, 1CTX and 2CTX, respectively. There are seventy seven compounds that matched 3 templates and only nineteen compounds have the "drug like" properties (molecular weight < 500, Log P < 5 and number of rotatable bonds <5). For experimental testing, 20 more compounds were included in the in vitro evaluation on the basis of low docking energy and high %member in the cluster of lowest energy. The total number of compounds in the binding study was thirty nine. The docking energies from 3 templates of the 39 selected compounds were listed in Table 13.

	1YI5		20	СТХ	1CTX		
	NCI number	Docking energy (kcal/mol)	NCI number	Docking energy (kcal/mol)	NCI number	Docking energy (kcal/mol)	
1	371884	-13.86	49487	-12.74	11241	-13.59	
	49487	-13.54	11241	-11.93	371884	-13.30	
I	11241	-13.32	371878	-11.91	23128	-13.10	
175	371878	-13.19	372280	-11.77	134755	-12.95	
Cpds	-				-		
I						•	
			•		•		
↓ ↓	105687	-10.11	136469	-9.52	14143	-10.12	

Table 12. The top 175 compounds for each  $\alpha$ -cobratoxin templates

Compound	NCI	MW	Log	Rotatable	E <sub>doc</sub> cluste	<sub>king</sub> in hi ring (kc	ghest al/mol)	% hig	membe hest clu	r in ster
-	number	(Da)	r	Donus	1YI5	2CTX	1CTX	1YI5	2CTX	1CTX
1	10458	482	3.57	7	-9.64	-9.78	-10.20	28	21	57
2	12181	458	5.22	5	-10.50	-10.10	-11.41	66	16	74
3	14410	444	0.35	4	-12.11	-10.81	-11.31	69	43	55
4	17245	459	2.17	7	-9.74	-9.71	-10.20	26	24	11
5	18877	406	4.62	4	-10.89	-9.56	-10.5	29	55	26
6	23217	405	6.50	4	-11.54	-10.11	-9.64	56	26	27
7	23904	524	2.85	7	-11.00	-6.68	-10.60	28	57	18
8	36387	696	3.18	4	-11.10	-9.34	-9.64	16	8	25
9	37245	467	4.33	4	-11.45	-9.19	-11.55	54	20	53
10	42258	417	3.27	4	-11.23	-8.27	-10.69	65	38	58
11	45583	504	4.18	7	-10.10	-9.78	-10.5	23	35	43
12	56452	378	5.02	6	-10.80	-10.65	-11.30	13	37	25
13	60043	317	3.56	4	-11.12	-9.70	-10.10	49	49	23
14	74702	549	5.95	2	-11.00	-11.12	-9.80	21	17	27
15	81509	352	1.16	5	-10.80	-9.20	-11.74	33	31	16
16	95090	582	2.02	5	-9.85	-8.69	-1.46	13	12	64
17	95926	447	2.71	5	-11.83	-10.73	-10.10	61	24	34
18	99671	401	1.06	4	-10.64	-9.30	-10.10	60	34	12
19	105687	379	6.01	5	-9.27	-10.16	-11.29	14	36	26
20	113491	470	-1.43	4	-12.07	-10.30	-10.00	74	34	21
21	116654	349	2.80	6	-10.50	-9.85	-10.2	29	22	25
22	121855	435	1.10	5	-10.50	-9.40	-8.19	50	29	60
23	121865	412	0.30	5	-11.09	-9.59	-9.91	72	53	20
24	127917	299	0.56	5	-9.71	-9.45	-10.96	48	32	29
25	128437	477	3.90	4	-11.83	-11.67	-10.20	39	33	50
26	131547	476	4.19	6	-11.41	-9.43	-10.10	16	16	22
27	134754	580	3.10	5	-10.20	-9.84	-11.00	44	71	22
28	134755	580	3.10	5	-10.80	-9.20	-9.87	41	24	7
29	163443	394	6.26	5	-11.41	-9.51	-10.10	28	17	13
30	205511	335	3.49	6	-9.91	-10.75	-10.62	31	54	24
31	282027	320	2.00	5	-11.01	-9.66	-10.20	29	36	21
32	339601	415	2.62	5	-10.75	-8.97	-9.56	23	12	26
33	371884	447	3.62	5	-13.20	-9.24	-12.10	17	25	28
34	372046	493	6.89	5	-11.70	-10.70	-12.48	50	58	20
35	372280	438	5.64	5	-10.90	-11.77	-9.86	54	16	36
36	372294	454	4.54	6	-10.50	-11.51	-9.73	37	16	21
37	600067	324	1.16	5	-11.78	-10.60	-11.84	77	78	40
38	659162	434	4.93	5	-10.50	-10.21	-11.31	46	25	28
39	659390	417	2.80	5	-8.72	-9.77	-9.62	28	42	18

 Table 13.
 Selected hits from virtual screening<sup>a</sup>

<sup>a</sup>39 compounds from NCI diversity set that match all *a*-cbtx templates (1YI5, 1CTX and 2CTX).

#### B. In Vitro Experiment

Thirty nine hits from virtual screening and rediocides (A and G) were investigated further for their binding capability. The binding between selected hits (39 compounds) and rediocides (A and G) with  $\alpha$ -cobratoxin were evaluated by (i) *in vitro* radioligand competition assay and (ii) sodium dodecyl sulfate polyacrylamide gel electrophoresis (SDS-PAGE). These experiments were carried out to substantiate the results from *in silico* experiments.

The AChBPs, a soluble extracellular binding domain of nAChR, came from snails in different species i.e. *Lymnaea stagnalis (Ls), Aplysia californica (Ac)* and *Bolunus truncatus (Bt)*. They were used for the investigation of the binding capability. AChBP from *Ac* shares 33% amino acid identity with those from *Ls* and less than 33% with *Bt.* AChBPs, *Ls* and *Ac*, share 24% amino acid identity with human nAChR  $\alpha$ 7 and 20% with human nAChR  $\alpha$ 1 while AChBP from *Bt* shares 22% amino acid identity with  $\alpha$ 7 and 17% with  $\alpha$ 1 (Table 14). The amino acid sequences of AChBP from snail and human nAChR ( $\alpha$ 7 and  $\alpha$ 1) were shown in Figure 20.

	Is AChDD		Human	nAChR
			α1	α7
Ac AChBP	33	< 33	24	20
Ls AChBP			24	20
Bt AChBP			22	17

Table 14. Percent amino acid identity of AChBPs



Figure 20. The amino acid sequences of AChBPs (*Ac*, *Ls* and *Bt*) and human nAChR ( $\alpha$ 7 and  $\alpha$ 1). Conserved residues are shown on a color background with bold denoting points of structural identity from an alignment of the crystallographic coordinates of the following complexes: AChBP from *L. stagnalis* with nicotine (PDB: 1UW6), AChBP from *A. californica* with epibatidine (PDB: 2BYQ), and AChBP from *B. truncatus* with CHAPS (3- (cyclohexylamino) propane-1-sulfonic acid, PDB: 2BJO). The plus (+) and minus (-) denote residues within the 4.0 Å radius of interaction at the subunit interface. Residues that are within 5 Å of all three molecules (nicotine, epibatidine and CHAPS) present in the binding site are italic ( $\downarrow$ ).

# 1. Radioligand Competition Assay

In the assay, the nAChR antagonist ( $^{125}I \alpha$ -bungarotoxin) and the nAChR agonist ( $^{3}H$  epibatidine) were the control as they completely bound with AChBP. [ $^{3}H$ ] epibatidine bound all three kinds of AChBPs but [ $^{125}I$ ]  $\alpha$ -bungarotoxin bound AChBP from *Ls* and *Ac* only. [ $^{3}H$ ] epibatidine was used for screening the hits that displaced its binding to AChBPs. AChBPs bound to the bead with the aid of the monoclonal anti-FLAG M2 antibody from mouse (Figure 21). The bead itself is made from polyvinyltoluene which serves as a scintillant. The bead emits light when there is a

high energy emitter nearby. The amount of emitted light from the samples were measured by a scintillation counter. The amount of light emitted by each sample is proportional to the amount of radioactive epibatidine or bungarotoxin bound to the binding protein on the bead which can then be related to the ability of a compound to compete with these radioligands. The redioligand binding to AChBP bound bead was displayed in Figure 22.



Figure 21. AChBPs bound to the bead by using the monoclonal anti-FLAG M2 antibody from mouse.



**Figure 22**. Radioactive  $[^{3}H]$  epibatidine or  $[^{125}I]$  bungarotoxin bound to the binding protein and emitted the light. Epibatidine or bungarotoxin is in blue, radioactive is in red and AChBP is in yellow. The size of each component is not to scale.

#### 1.1. Screening Test

The radioligand assay was initally run to screen the 39 hits prior to the determination of  $K_d$ . Ten micromolar of each compound was screened for their ability to displace the binding of  $[^{3}H]$  epibatidine to the three AChBPs (*Lymnaea stagnalis* (Ls), Aplysia californica (Ac) and Bolunus truncatus (Bt)) as measured by a scintillation proximity test (Figure 23). NCI36387 (compound 8), NCI42258 (compound 10), NCI121865 (compound 23) and NCI134754 (compound 27) showed the low fraction of  $[{}^{3}H]$  epibatidine bind to AChBPs (less than 0.6). The 4 hits can bind to the agonist [<sup>3</sup>H] epibatidine as strong as the AChBP which led to the reduction of redioligand binding to AChBP on the bead. The results of fraction over 1.0 were possibly due to the variation in number of AChBP binding to the beads or the interferences in light emission from the hits.



**Figure 23**. Screening test for the ability to displace the binding of [<sup>3</sup>H] epibatidine on AChBP from *Lymnaea stagnalis* (*Ls*), *Aplysia californica* (*Ac*) and *Bulinus truncatus* (*Bt*).

# **1.2.** Determination of $K_d$

The four hits were determined for the dissociation constant ( $K_d$ ).  $K_d$  is commonly used to describe the affinity between a ligand (4 hits) and a protein (AChBPs). The smaller the dissociation constant, the more tightly bound the ligand is, or the higher the affinity between ligand and protein. Four hits (compounds 8, 10, 23 and 27) from the screening test in varied concentrations from millimolar to nanomolar were added to the fixed concentrations of AChBPs (~500 pM binding sites), polyvinyltoluene anti-mouse SPA scintillation beads, monoclonal anti-FLAG M2 antibody from mouse. Then, agonist [<sup>3</sup>H] epibatidine (5 nM for *Ls* and *Bt*, 20 nM for *Ac*) or antagonist [<sup>125</sup>I]  $\alpha$ -bungarotoxin (10 nM for *Ls* and 20 nM for *Ac*) were added and allowed to equilibrate at room temperature. The plot between fraction of [<sup>3</sup>H] epibatidine or  $[^{125}I]$   $\alpha$ -bungarotoxin and the concentration of 4 hits were shown in Figure 24 and 25, respectively.



Figure 24. The ability to displace the binding of [<sup>3</sup>H] epibatidine on AChBPs (~500 pM binding site) from *Lymnaea stagnalis* (A), *Aplysia californica* (B) and *Bulinus truncatus* (C). X-axis is the log concentration of the 4 hits in molar, the concentrations of [<sup>3</sup>H] epibatidine were 5 nM for AChBPs derived from *Ls* and *Bt*, and 20 nM for *Ac*.

Maleeruk Utsintong



**Figure 25**. The ability to displace the binding of  $[^{125}I]$  *a*-bungarotoxin on AChBPs from *Lymnaea stagnalis* (A) and *Aplysia californica* (B). X-axis is the log concentration of the 4 hits in molar, 10 nM for AChBPs derived from *Ls* and 20 nM for *Ac*.

In the assay, four hits competitively displaced on antagonist (<sup>125</sup>I  $\alpha$ bungarotoxin) and the agonist (<sup>3</sup>H epibatidine) from their mutually exclusive binding sites on AChBPs from Lymnaea, Aplysia and Bulinus with the concentration from micromolar to nanomolar. Their chemical structures, NCI numbers, molecular weights,  $K_d$  values and estimated binding free energies were listed in Table 15 and 16. The hit with a better binding on AChBPs has a lower  $K_d$  value. According to [<sup>3</sup>H] epibatidine, NCI36387 (compound 8) was the most potent in binding with AChBPs from Ls and Bt while NCI121865 (compound 23) was most potent in binding with Ac. For the antagonist  $[^{125}I]$   $\alpha$ -bungarotoxin, NCI36387 was the most potent for binding with AChBP from Ls while NCI121865 most potent for binding with Ac AChBP. Interestingly, NCI36387 was found to be d-tubocurarine, the active ingredient in curare and the well known neuromuscular blocker. The results showed that the activity for competing with  $[^{3}H]$  epibatidine and  $[^{125}I]$  *a*-bungarotoxin for NCI121865 (compound 23) was better than NCI42258 (compound 10) and NCI134754 (compound 27), respectively. The graph has shifted into in left hand side. NCI121865 was the potent lead compound and the mechanism for the anti-cobratoxin may be competing with the toxin to bind the acetylcholine receptor. According to the virtual screening result (Table 13), the members in the lowest energy cluster of NCI42258 and NCI121865 are considerably high, 65 % and 72 %, respectively. Based on Binding Efficiency Index (BEI), the new parameter for filtering the drug like molecules, BEI value of NCI121865 is the best among the hits. BEI is calculated from pK<sub>i</sub> and molecular weight in kDa, the ideal BEI value is 27. It is 13.32 with  $[^{3}H]$  epibatidine and 32.83 with  $[^{125}I] \alpha$ -bungarotoxin on AChBP from Ac.

The amino acid of  $\alpha$ -cobratoxin interacted with 3 hits within 4 Å were shown in Figure 26-27 and Table 17. The active binding site of  $\alpha$ -cobratoxin is occupied between loop I and II in the presence of the hits and consequently the  $\alpha$ -cobratoxin can not bind to the acetylcholine receptor.

# Maleeruk Utsintong

Chemical structure	NCI number (cpd no.)	MW	Rotatable bonds	Cluster (% lowest energy clustering)	E <sub>docking</sub> (kcal/mol)	K <sup>a</sup> (nM)
MeO O O O H O H O H O H O H O H O H O H O	36387 (8)	696	4	4 (16)	-11.10	224
NH	42258 (10)	417	4	1 (65)	-11.23	54.2
HOOC HN N N OMe OMe	121865 (23)	412	5	1 (72)	-11.09	68.7
MeO MeO	134754 (27)	580	5	7 (44)	-10.20	289

# Table 15. NCI diversity set identified from virtual screening

<sup>a</sup>Dissociation constant or inhibition constant ( $K_i$ ) calculated by AutoDock program from the docking between the compound (NCI) and  $\alpha$ -cobratoxin template from 1YI5 (AChBP from *Lymnaea stagnalis*).

NCI		Radioligand					
number	AChBP <sup>a</sup>	[ <sup>3</sup> H] epibatidine		[ <sup>125</sup> Ι] α- bungarotoxin			
(cpd no.)		$K_d (\mathbf{nM})^{\mathbf{b}}$	BEI <sup>c</sup>	$K_d (\mathbf{nM})^{\mathbf{b}}$	BEI <sup>c</sup>		
36387 (8)	Ls	20.49	18.61	13.83	12.19		
	Ac	78.96	76.23	241.70	238.28		
	Bt	37.45	35.19				
42258 (10)	Ls	245.70	239.97	93.43	88.70		
	Ac	1147.00	1139.66	2913.00	2904.69		
	Bt	1890.00	1882.14				
121865 (23)	Ls	111.50	106.53	75.90	71.34		
	Ac	16.26	13.32	36.63	32.83		
	Bt	415.30	408.94				
134754 (27)	Ls	792.60	787.60	404.40	399.91		
	Ac	2366.00	2360.18	4620.00	4613.68		
	Bt	2629.00	2623.10				

**Table 16.**  $K_d$  of 4 hits on AChBPs

<sup>a</sup>Acetylcholine binding protein (AChBP) *from Lymnaea stagnalis* (*Ls*), *Aplysia californica* (*Ac*) and *Bulinus truncatus* (*Bt*),  ${}^{b}K_{d}$  calculated from the binding between hit and AChBPs,  ${}^{c}BEI = pKi / M.W.$  (kDa).





Figure 26. The dock orientations of NCI42258 (indigo blue), NCI121865 (light green) and NCI134754 (purple) (A) and crystallographic Ser182 – Tyr 192 of chain C from 1YI5 (purple) and control peptide (orange) (B). The amino acid residues of α-cobratoxin interacted with NCI42258 (C), NCI121865 (D) and NCI134754 (E).







**Figure 27.** The amino acid residues of  $\alpha$ -cobratoxin interacted with NCI42258 (A), NCI121865 (B) and NCI134754 (C) with distance < 4 Å.

NCI42258 - Cbtx	Distance (Å)	NCI121865 - Cbtx	Distance (Å)	NCI134754 - Cbtx	Distance (Å)
N – HO (Thr6)	2.994	NH – OC (Lys35)	2.273	N – HO (Thr6)	3.474
N – HO (Thr6)	3.988	NH – OC (Lys35)	3.505	O – HN (Phe65)	3.589
NH – OC (Lys35)	3.967	O – HN (Arg68)	3.187	O – HN (Phe65)	3.743
NH – OC (Val37)	3.342	CO – HN (Arg68)	2.987		
		CO – HN (Arg68)	3.310		
		OH – OC (Arg68)	3.830		

**Table 17.** Amino acid residues of  $\alpha$ -cobratoxin interacted with 3 hits

#### 2. Sodium Dodecyl Sulfate Polyacrylamide Gel Electrophoresis (SDS-PAGE)

#### 2.1. Effect of Rediocides on *a*-Cobratoxin and AChBP

In order to substantiate the molecular mechanism derived from the docking experiments, SDS-PAGE was carried out to detect the binding between the  $\alpha$ cobratoxin and rediocides. In this study, a 50:50 mixture of rediocides A and G was used as the 50:50 imitated the composition ratio of rediocides in the traditional regimen for the treatment of snake bites. The experiment was separated into 2 sections i.e. the study of interaction between rediocides and  $\alpha$ -cobratoxin as described in experimental section 2.1.1 and the binding between rediocides and  $\alpha$ -cobratoxin in the presence of AChBP as decribed in experimental section 2.1.2.

## 2.1.1. Effect of Rediocides on *a*-Cobratoxin

SDS-PAGE was performed using 15% acrylamide gels and the rediocides (A & G) were dissolved in DMSO. The sample mixtures were loaded into each gel after being mixed with the sample buffer, boiled for 5 minutes and centrifuged. Electrophoresis was run at 150 volts until the marker's dye front reached to the end of the gels. Electrophoretic profiles of crude venom, a mixtures of the rediocides (A:G-50:50) and crude venom are shown in Figure 28. The protein of cobra venom (lane 2) showed two predominant bands at 10 kDa ( $\alpha$ -cobratoxin) and 14 kDa. The SDS-PAGE electrophoretograms demonstrated that the bands of  $\alpha$ -cobratoxin in crude venom decreased with the increase in concentration of the rediocides. When the crude venom was mixed with serial dilutions of rediocides (lane 3-6), the  $\alpha$ -cobratoxin band from the crude venom was neutralized by the rediocides. From the result, it can be explained that rediocides bound  $\alpha$ -cobratoxin and diminished the  $\alpha$ -cobratoxin band. However, it is also possible that venom proteins were denatured or degraded and precipitated before loading on the gel. In order to clarify this notion, the binding between rediocides and  $\alpha$ -cobratoxin in the presence of AChBP was studied in the next section.



Figure 28. SDS-PAGE of the mixtures between crude venom and rediocides (A & G). Mixture of crude venom (50 μg) and rediocides with final concentrations of 2-fold dilution was applied to the gel, and electrophoresis was achieved with an electric current of 150 volts per gel. Lane 1: molecular weight marker, Lane 2: crude venom, Lanes 3-6: the mixtures between venom and rediocides in the final concentrations of 7.5, 15, 30 and 60 mg/8 ul, respectively.

#### 2.1.2. Effect of Rediocides on *a*-Cobratoxin and AChBP

In this experiment, the AChBP, a soluble extracellular binding domain of nAChRs, was added in order to study the binding between rediocides and  $\alpha$ -There are 3 AChBPs from snail, Lymnaea stagnalis (Ls), Aplysia cobratoxin. californica (Ac) and Bolunus truncatus (Bt). The AChBP from Ac was selected in this experiment because AChBP from Ac was the most stable and only a small amount of monomer was degraded from the AChBP pentamer in the gel runing condition. The conditions for running gel were modified to avoid the breakdown of the AChBP pentamer into monomers (25 kDa) by omiting the boiling of the sample before loading it to the gel. Electrophoresis was run at 100 volts until the marker's dye front reached to the end of the gels. Electrophoretic profiles of AChBP from Ac,  $\alpha$ -cobratoxin, and the rediocides (A:G-50:50) are shown in Figure 29. The AChBP protein showed a single band of pentamer at 100 kDa (lane 6). In the presence of  $\alpha$ -cobratoxin (lane 5) or rediocides (lane 2), the intensity of the AChBP pentamer band was increased substantially when compared with the band from AChBP alone (lane 6). The

pentamer band appeared in lane 2 is stronger in intensity than lane 6 because the rediocides bound with AChBP and made AChBP more stable. The result in lane 5 was in the same manner where the  $\alpha$ -cobratoxin bound with AChBP and stabilized AChBP. In the presence of  $\alpha$ -cobratoxin and rediocides (lane 4), the intensity of AChBP band (pentamer) decreased when compared with pentamer band in lane 5. It is because the rediocides bound to the  $\alpha$ -cobratoxin and prevented  $\alpha$ -cobratoxin from binding with AChBP. The results from SDS-PAGE electrophoretograms demonstrated that both  $\alpha$ -cobratoxin and rediocides bound with AChBP, as well as rediocides bound to both  $\alpha$ -cobratoxin and AChBP.



Figure 29. SDS-PAGE of AChBP from *Aplysia stagnalis* in the presence of α-cobratoxin and rediocides (A & G). Electrophoresis was achieved with an electric current of 100 volts. Lane 1: molecular weight marker, Lane 2: the mixture between AChBP and rediocides (A & G), Lane 3: α-cobratoxin, Lane 4: the mixture of AChBP, α-cobratoxin and rediocides (A & G), Lane 5: the mixture between AChBP and α-cobratoxin, Lane 6: AChBP. Concentrations of AChBP, α-cobratoxin and rediocides (A & G) were 200 µg/5 µl, 62.5 µg/5 µl and 2 mg/5 µl, respectively.

## 2.2. Effect of 3 Hits on *α*-Cobratoxin and AChBP

### 2.2.1. Effect of 3 Hits on *a*-Cobratoxin

SDS-PAGE was carried out to detect the binding between the 3 hits and  $\alpha$ cobratoxin from crude venom. SDS-PAGE was performed with 20% acrylamide gels and the hits were dissolved in 0.1% Tween 80. The sample mixtures were loaded into each gel after mixed with the sample buffer, boiled for 5 minutes and centrifuged. Electrophoresis was performed at 100 volts until the marker's dye front reached to the end of the gels. Electrophoretic profiles of crude venom, 3 hits from NCI diversity set (NCI42258, NCI121865 and NCI134754) and the mixtures of hits and crude venom are shown in Figure 30. The protein of crude cobra venom showed three predominant bands at 10 kDa ( $\alpha$ -cobratoxin), 12 and 14 kDa (lane 2). This variation in the number of protein bands, 3 bands versus 2 bands in section 2.1.1 came from the different lot number of crude cobra venom. In the presence of NCI42258 or NCI134754 (lanes 3 and 5), the electrophoretic profile of  $\alpha$ -cobratoxin was different from  $\alpha$ -cobratoxin (lane 2), the bands appeared to be more diffuse and broaden. It is possible that NCI42258 and NCI134754 may alter some characteristic of the  $\alpha$ -cobratoxin. In the presence of NCI121865 (lane 4), the intensity of the  $\alpha$ -cobratoxin band significantly increased when compared with those from  $\alpha$ -cobratoxin (lane 2). From the result, it can be explained that NCI121865 bind to  $\alpha$ -cobratoxin and changed or destroyed the tertiary and quarternary structure of  $\alpha$ -cobratoxin protein leading to broaden band. For this reason, the binding of 3 hits and  $\alpha$ -cobratoxin was studied in the presence of AChBP (experimental section 2.2.2).

#### Maleeruk Utsintong



Figure 30. SDS-PAGE of the mixtures between crude venom and 3 hits. Electrophoresis was achieved with an electric current of 100 volts per gel. Lane 1: molecular weight marker; Lane 2: crude venom (50 μg/5 μl), Lane 3: NCI42258 (50 μg/5 μl) and venom, Lane 4: NCI121865 (50 μg/5 μl) and venom, Lane 5: NCI134754 (50 μg/5 μl) and venom.

## 2.2.2. Effect of 3 Hits on *a*-Cobratoxin and AChBP

The AChBP was included in this experiment for studying the effect of 3 hits on the binding of  $\alpha$ -cobratoxin on AChBP or the anti-venom effect. SDS-PAGE was performed using 20% acrylamide gels and the hits were dissolved in 0.1% Tween 80. The conditions for running gel were modified to avoid the breakdown of the AChBP pentamer into monomers by omiting the sample boiling before loading. Electrophoretic profiles of AChBP from *Aplysia stagnalis*,  $\alpha$ -cobratoxin, mixtures of hits and these two proteins were shown in Figure 31. The proteins of AChBP showed a predominant band at 100 kDa (pentamer) and a small band of monomer at 25 kDa. Lanes 8-10 were the bands of AChBP in the presence of hits (NCI42258 or NCI121865 or NCI134754), the AChBP pentamer bands in lanes 8-10 were not increased in the intensity but also expanded and shift up. The monomer band only in lanes 4 and 8 decreased almost completely when compared with those of AChBP in lane 2. The increase in intensity and the shift up of the pentamer bands resulted from the binding between 3 hits and AChBP which led to the reduction in monomer breaking. The decrease in intensity of monomer band (25 kDa) in lane 8 emphasized that NCI42258 bound with AChBP and stabilized it from breaking. Lanes 4-6 were the lanes of  $\alpha$ -cobratoxin and AChBP in the presence of 3 hits, the intensity of the AChBP pentamer band in lanes 4-6 increased while AChBP monomer band in lane 4 decreased when compared with those from AChBP (lane 2). The  $\alpha$ -cobratoxin bands in lanes 4-6 shifted up when compared with  $\alpha$ -cobratoxin band (lane 7), the result confirmed that 3 hits bind the  $\alpha$ -cobratoxin and prevented the toxin to bind with the AChBP. The changes in the profile of  $\alpha$ -cobratoxin band i.e. band broadening, shifting and the increase in intensity demonstrated the interaction between hits and  $\alpha$ cobratoxin. The stronger in the diffuse and shift of pentamer bands in lanes 4-6 comparing to the pentamer band of AChBP in lane 2 showed that both hits also bound with AChBP. The result from the pentamer band in lane 3 was in the same manner where the  $\alpha$ -cobratoxin bound with AChBP and made AChBP more stable. All 3 hits bound to the  $\alpha$ -cobratoxin and prevented  $\alpha$ -cobratoxin from binding with AChBP. It can be concluded from SDS-PAGE result that 3 hits (NCI42258, NCI121865 and NCI134754) bound both  $\alpha$ -cobratoxin and AChBP.



**Figure 31.** SDS-PAGE of  $\alpha$ -cobratoxin and AChBP from *Aplysia stagnalis* in the presence of 3 hits. Electrophoresis was achieved with an electric current of 100 volts. Lane 1: molecular weight marker, Lane 2: the mixture between AChBP and  $\alpha$ -cobratoxin, Lane 3: the mixture of AChBP,  $\alpha$ -cobratoxin and NCI42258, Lane 4: the mixture of AChBP,  $\alpha$ -cobratoxin and NCI121865, Lane 5: the mixture of AChBP,  $\alpha$ -cobratoxin, Lane 7:  $\alpha$ -cobratoxin, Lane 8: the mixture of AChBP and  $\alpha$ -cobratoxin, Lane 7:  $\alpha$ -cobratoxin, Lane 8: the mixture of AChBP and NCI42258, Lane 9: the mixture of AChBP and NCI42258, Lane 9: the mixture of AChBP and NCI4258, Lane 9: the mixture of AChBP and NCI4258, Lane 9: the mixture of AChBP and NCI4254. Concentrations for AChBP,  $\alpha$ -cobratoxin, rediocides were 200 µg/5 µl, 62.5 µg/5 µl and 50 µg/5 µl, respectively. The blue letters are the quatity of the band in percentage using Syngene program.

#### C. In Vivo Experiment

An *in vivo* test was carried out to investigate the effect of rediocides and the three hits from NCI diversity against  $\alpha$ -cobratoxin. The median lethal dose (LD<sub>50</sub>) of  $\alpha$ -cobratoxin, the protection activity and the anti-cobratoxin were determined.

## 1. Determination of Median Lethal Dose

LD<sub>50</sub> is defined as the least amount of  $\alpha$ -cobratoxin which injected intramuscularly through the leg muscle of the mice, resulted in the death of half amount of mice within 24 h. The plot between % death and the concentration of  $\alpha$ cobratoxin was shown in Figure 32. A LD<sub>50</sub> of  $\alpha$ -cobratoxin was found to be 0.175 mg/kg (i.m.). Since 100% mice were death within 30 minutes at 3LD<sub>50</sub> dose (0.425 mg/kg), the dose of  $\alpha$ -cobratoxin in the experiment was 3LD<sub>50</sub> instead of 4LD<sub>50</sub>.



**Figure 32.** LD<sub>50</sub> of  $\alpha$ -cobratoxin (n=6).

#### 2. Effect of Rediocides and Three Hits on Mice

#### 2.1 The Protection Activity

In this protocol, rediocides or hits were tested for their protecting action. Rediocides or hits were injected to the mice before the injection of  $\alpha$ -cobratoxin. The sample solution, in 0.05 ml of 0.3% Tween 80, were prepared in varying concentrations. Each dose was injected to the mice intravenously (i.v.) at tail vein. The control group was injected with only the vehicle (0.3% Tween 80). The acute toxicity test of rediocides was determined prior to the test for protecting action. The result showed that the injection of rediocides (50 mg/kg, i.v.) resulted in 100% death within 24 hours and only 50% death at 1 mg/kg dose (Table 18). The maximum concentration of rediocides without acute toxicity was 0.5 mg/kg (i.v). All mice were alive at the maximum testing dose. The survived mice treated with 1 mg/kg showed the sign of toxicity at the tail vein after 24 hours (Figure 33).

Group	Dose (i.v) (mg/kg)	% death
Rediocides	50	100
	1	50
	0.5	0
	0.15	0
42258	10	0
	5	0
	1	0
121865	10	0
	5	0
	1	0
134754	10	0
	5	0
	1	0

# Table 18. Toxicity test of rediocides and 3 hits (n=6)



Figure 33. Toxicity of rediocides (A & G). The mouse was injected with rediocides (A) and NCI134754 (B).

For protection activity,  $\alpha$ -cobratoxin was injected before the injection of rediocides (A & G) or 3 hits from NCI diversity (NCI42258, NCI121865 and NCI134754). The testing dose of rediocides (A & G) was 0.5 mg/kg, i.v. and 1, 5 and 10 mg/kg for 3 hits. When 3LD<sub>50</sub> of  $\alpha$ -cobratoxin was injected after a rediocides injection (0.5 mg/kg, i.v.), the survival time was prolonged significantly (Table 19). The survival time of the control group was 27.1 ± 2.8 minutes while the survival time for rediocides (A & G) was 45.2 ± 9.7 minutes. The rediocides was the most potent for protecting mice from the  $\alpha$ -cobratoxin.

In the same way as rediocides (A & G), the hits solutions, in 0.05 ml of 0.3% Tween 80, were injected to the mice intravenously (i.v.) at tail vein. The control group was injected with only the vehicle, 0.3% Tween 80. When the hits were injected (i.v.) 30 minutes before  $\alpha$ -cobratoxin (3LD<sub>50</sub>, i.m.), the survival time increased significantly. The dixon's Q test was used to discard the outliers from the data. The Q<sub>exp</sub> values of the highest numbers (> 24 h and 111 minutes) at 5 mg/kg (NCI42298 and NCI121865) were higher than Q<sub>crit</sub> (CL: 95%). Therefore, these two numbers were discarded as outliers. The result showed that all hits can prolong the survival time of the mice (Table 19). The survival time for control was 27.1 ± 2.8 minutes while the survival time for NCI42258 and NCI121865 at 5 mg/kg dose were double, 49.6 ± 9.3 minutes and 50.6 ± 5.0 minutes, respectively. When the dose of hits were increased to 10 mg/kg, the survival times of the hit treated groups did not increase

further. The survival time of NCI42258, NCI121865 and NCI134754 at 10 mg/kg were  $43.2 \pm 7.6$ ,  $38.3 \pm 7.5$  and  $41.8 \pm 7.3$  minutes, respectively. The 5 mg/kg dose was the optimal dose and was choosen for the antitoxin activity.

Group	Dose (i.v.) (mg/kg)	Survival time (min)	Average (Mean + SD)
	(		(mean ± SD)
Control		21, 24, 25, 26, 27, 27,	$27.1 \pm 2.8$
		28, 28, 29, 29, 30, 31	
Rediocides	0.5	33, 35, 47, 48, 49, 59	45.2 ± 9.7*
42258	10	34, 39, 40, 41, 52, 53	$43.2 \pm 7.6*$
	5	37, 44, 50, 58, 59, >24 h <sup>b</sup>	49.6 ± 9.3*
	1	35, 37, 39, 45, 49, 50	$42.5 \pm 6.4*$
121865	10	30, 32, 35, 40, 43, 50	38.3 ± 7.5*
	5	44, 47, 52. 54. 56, 111 <sup>b</sup>	$50.6 \pm 5.0*$
	1	35, 36, 38, 51, 52, 54	44.3 ± 8.9*
134754	10	37, 37, 38, 40, 43, 56	41.8 ± 7.3*
	5	31, 37, 47, 48, 51, 53	44.5 ± 8.6*
	1	40, 40, 42, 44, 53, 60	46.5 ± 8.2*

**Table 19.** Protective effect of rediocides (A & G) and 3 hits against  $\alpha$ -cobratoxin<sup>a</sup>

<sup>a</sup>Hits was injected (i.v.) 30 minutes before injection of  $\alpha$ -cobratoxin (3LD<sub>50</sub>) (i.m.), n=6, <sup>b</sup>The outliers (Dixon's Q-test), \*p < 0.05.

## 2.2 The Antitoxin Activity

In this protocol, rediocides or hits were tested for the antitoxin action by injected after the injection of  $\alpha$ -cobratoxin.  $3LD_{50}$  of  $\alpha$ -cobratoxin in 0.1 ml of sterile water was injected (i.m.) before the injection of test compounds in 0.05 ml of 0.3% Tween 80 (i.v.). The survival time and number of deaths occurring within 24 hours

was recorded. Alpha-cobratoxin was injected before the injection of rediocides and resulted in the death of all the mice within 46 minutes, which was not significantly different from the control (Table 20). The result showed that the rediocides at 0.5 mg/kg could not protect the mice from the  $\alpha$ -cobratoxin when they were injected after the  $\alpha$ -cobratoxin. Inspite of better affinity than  $\alpha$ -cobratoxin (lower docking energy), the rediocides can not prolong the survival time when injected following the  $\alpha$ -cobratoxin. It is because the better accessibility of  $\alpha$ -cobratoxin to the AChBP based on the higher of % member in the cluster than those of rediocides. The docking energy between  $\alpha$ -cobratoxin and the control peptide from AChBP was -12.20 kcal/mol (with 100% member in the cluster) and the docking energy between  $\alpha$ -cobratoxin and the rediocides (A & G) was -14.17 kcal/mol (with 40% member in the cluster), respectively.

Group	Dose (i.v.)	Survival time (min)	Average (Mean ± SD)
Control	-	22, 23, 23, 24, 26, 32	$25.0\pm3.7$
Rediocides (A & G)	0.5 mg/kg	25, 26, 27, 31, 32, 46	$31.2 \pm 7.8$

**Table 20.** Anti-cobratoxin of rediocides against  $\alpha$ -cobratoxin

<sup>a</sup>Rediocides was injected (i.v.) immediately after injection of  $\alpha$ -cobratoxin (3LD<sub>50</sub>) (i.m.), n=6.

Two hits (NCI121865 and NCI134754) at 5 mg/kg were able to protect the mice from the  $\alpha$ -cobratoxin when they were injected after the  $\alpha$ -cobratoxin (Table 21). The survival times were  $45.2 \pm 5.2$  minutes and  $53.7 \pm 15.6$  minutes for NCI121865 and NCI134754, respectively. The NCI42258 at 5 mg/kg can prolong the survival time for 5 minutes but the prolonging time was not significant in statistic. Three hits can prolong the survival time of the mice if injected 30 minutes before injection with  $\alpha$ -cobratoxin, but only NCI121865 and NCI134754 can prolong the survival time of the mice when  $\alpha$ -cobratoxin was injected before injection of hits. NCI121865 and NCI134754 are the potential candidates as they have the antitoxin activity. The results from *in vivo* experiment supported the *in vitro* and *in silico* results.

Group	Survival time	Average
	(min)	(Mean ± SD)
Control	28, 31, 37, 38, 42, 45	$36.8 \pm 6.4$
42258	34, 34, 36, 42, 46, 56	$41.3 \pm 8.7$
121865	38, 41, 44. 48. 48, 52	45.2 ± 5.2*
134754	40, 41, 45, 55, 60, 81	53.7 ± 5.6*

**Table 21.** Anti-cobratoxin of 3 hits against  $\alpha$ -cobratoxin<sup>a</sup>

<sup>a</sup>Hits at 5 mg/kg was injected (i.v.) immediately after injection of  $\alpha$ -cobratoxin (3LD<sub>50</sub>) (i.m.), n=6, \*p < 0.05.

# **D.** Summary

Molecular docking was employed to investigate the binding mode of rediocides (A & G) and 1990 compounds in NCI diversity set were screened for its binding capability against  $\alpha$ -cobratoxin. This procedure is computationally intensive for analyzing a large database but provides the most detailed basis for determining which compounds are likely to be potent ligands. Thirty nine hits from virtual screening and rediocides (A & G) were further tested for its binding. It was found that 4 hits (NCI36387, NCI42258, NCI121865 and NCI134754) showed the low fraction of [<sup>3</sup>H] epibatidine bind to AChBPs. Four hits competitively displace on antagonist  $(^{125}\text{I} \alpha$ -bungarotoxin) and the agonist  $(^{3}\text{H epibatidine})$  from their mutually exclusive binding sites on Lymnaea, Aplysia and Bulinus AChBPs with the concentration from micromolar to nanomolar. One of the hits (NCI36387) is d-tubocurarine, the well known neuromuscular blocker. NCI121865 showed good competitive activity in displacing the binding of  $[{}^{3}H]$  epibatidine or  $[{}^{125}I]$   $\alpha$ -bungarotoxin to all types of AChBPs. The effect of three hits (NCI42258, NCI121865 and NCI134754) and rediocides (A & G) on  $\alpha$ -cobratoxin and AChBP were also determined by SDS-PAGE. Rediocides bound to  $\alpha$ -cobratoxin in crude venom and diminished the  $\alpha$ -cobratoxin band. NCI121865 also bound  $\alpha$ -cobratoxin in crude venom but increased the  $\alpha$ -
cobratoxin band. NCI42258 and NCI134754 changed some characteristic of the  $\alpha$ -cobratoxin because the  $\alpha$ -cobratoxin band appeared to be broaden.

When rediocides are added to the mixture of  $\alpha$ -cobratoxin and AChBP, the intensity of AChBP pentamer band decreased when compared with the pentamer band of AChBP without toxin. It is because the rediocides bound to the  $\alpha$ -cobratoxin and prevented  $\alpha$ -cobratoxin from binding with AChBP. Both  $\alpha$ -cobratoxin and rediocides bound with AChBP. In the presence of 3 hits in  $\alpha$ -cobratoxin and AChBP, the intensity of both AChBP pentamer band and the  $\alpha$ -cobratoxin band increased. The results demonstrated that three hits (NCI42258, NCI121865 and NCI134754) bound both  $\alpha$ -cobratoxin and AChBP.

Three hits (NCI42258, NCI121865 and NCI134754) and rediocides (A & G) were tested in the *in vivo* experiment. All hits and rediocides (A & G) can prolong the survival times of the mice if injected 30 minutes before  $\alpha$ -cobratoxin injection. Only NCI121865 and NCI134754 can prolong the survival time if injected immediately after  $\alpha$ -cobratoxin injection. The result from *in vitro* SDS-PAGE and binding assay as well as *in vivo* tests were in agreement with the *in silico* results. In clinical applications, NCI121865 (compound 23) and NCI134754 (compound 27) would be very useful potential leads for the treatment of snakebite victims.

# CHAPTER V CONCLUSION

# In search of an antidote of cobra venom by preventing the $\alpha$ -cobratoxin from binding to the nicotinic acetylcholine receptor, a comprehensive study was carried out involving *in silico* docking, virtual screening, and *in vivo* animal tests. First, molecular docking was employed to investigate the binding mode of rediocides (A & G) to $\alpha$ -cobratoxin by using the AutoDock program version 3.0.5. This study aimed to find the mechanism of action of rediocides against cobra venom at the molecular level. The rediocides and hits from virtual screening were investigated for anti-cobratoxin action both *in vitro* and *in vivo*. The performed studies are summarized below.

#### 1. Molecular Modeling (In Silico) Experiment

(i) A model of  $\alpha$ -cobratoxin was constructed and a control peptide was used to validate it by re-docking. The validation result with RMSD less than 2 Å indicated that the prepared  $\alpha$ -cobratoxin template was a good model for docking studies of rediocide binding modes and the diversity set.

(ii) The docking study on rediocides and  $\alpha$ -cobratoxin interaction was performed to investigate the binding modes of the rediocides. The rediocides were found to bind to the same location that the  $\alpha$ -cobratoxin bound to the nicotinic acetylcholine receptor. The active binding site of  $\alpha$ -cobratoxin is occupied in the presence of rediocides and consequently the  $\alpha$ -cobratoxin can not bind to the acetylcholine receptor. Thus, the *in silico* experiment revealed the detoxification mechanism of cobra venom at the molecular level.

(iii) Molecular docking was used in a virtual screening to investigate the binding of over 1990 compounds to  $\alpha$ -cobratoxin. There were 77 compounds that matched the three  $\alpha$ -cobratoxin templates and only nineteen of those compounds had "drug like" properties (molecular weight < 500, Log P < 5 and number of rotatable

bonds < 5). For experimental testing, 20 more compounds were included for *in vitro* evaluation on the basis of low docking energy and high %member in the cluster of lowest energy.

2. Radioligand Competition Assay

(i) A radioligand assay was initially run to screen the 39 hits and rediocides (A & G) prior to the determination of  $K_d$ . Four hits (NCI36387, NCI42258, NCI121865 and NCI134754) showed a low amount of fraction of [<sup>3</sup>H] epibatidine bind to AChBPs. The four hits either bind to the agonist [<sup>3</sup>H] epibatidine or the AChBP which led to the reduction of redioligand binding to AChBP on the bead.

(ii) The four hits were found to competitively displace antagonist (<sup>125</sup>I  $\alpha$ bungarotoxin) and the agonist (<sup>3</sup>H epibatidine) from their mutually exclusive binding sites on AChBPs with concentrations from micromolar to nanomolar.

3. Sodium Dodesyl Sulfate Gel Electrophoresis (SDS-PAGE)

SDS-PAGE was carried out to detect the binding between the  $\alpha$ -cobratoxin and rediocides and hits from virtual screening.

(i) When the crude venom was mixed with serial dilutions of rediocides, the  $\alpha$ -cobratoxin band from the crude venom was neutralized by the rediocides. All three hits bound to the  $\alpha$ -cobratoxin and changed the profile of  $\alpha$ -cobratoxin band to be broaden.

(ii) The acetylcholine binding protein (AChBP) was included for studying the effect on the binding of  $\alpha$ -cobratoxin on AChBP or the anti-venom effect. The rediocides or  $\alpha$ -cobratoxin bound with AChBP and made AChBP more stable. The rediocides were found to bind both  $\alpha$ -cobratoxin and AChBP which prevent  $\alpha$ cobratoxin from binding with AChBP. In the same way, all three hits bound to the  $\alpha$ cobratoxin and AChBP preventing  $\alpha$ -cobratoxin from binding with AChBP.

#### 4. In Vivo Experiment

An *in vivo* test was carried out to investigate the effect of rediocides and the three hits from NCI diversity against  $\alpha$ -cobratoxin.

(i) Rediocides (A & G) and the three hits were tested for their protecting action. Rediocides or hits were injected to the mice before the injection of  $\alpha$ -cobratoxin. When 3LD<sub>50</sub> of  $\alpha$ -cobratoxin was injected in the rediocides (0.5 mg/kg, i.v.) or 3 hits treated groups (0.5 mg/kg, i.v.), the survival time was prolonged significantly. Rediocides and the three hits prolonged the survival time of the mice if injected 30 minutes before injection with  $\alpha$ -cobratoxin.

(ii) Rediocides (A & G) and the three hits were tested for antitoxin action. Rediocides or hits were injected after the injection of  $\alpha$ -cobratoxin (3LD<sub>50</sub> dose) and the survival time and number of deaths occurring within 24 hours was recorded. In spite of the better affinity for  $\alpha$ -cobratoxin than the three hits, the rediocides did not prolong the survival time significantly when injected following the  $\alpha$ -cobratoxin. Only NCI121865 and NCI134754 prolonged the survival time of the mice significantly when  $\alpha$ -cobratoxin was injected before injection of hits. NCI121865 and NCI134754 are the best potential candidates as they have both protection activity and antitoxin activity.

*In silico* screening can help to identify the most promissing leads for anticobratoxin. The results from *in vitro* binding assay and SDS-PAGE as well as *in vivo* tests substantiate the *in silico* result. NCI121865 and NCI134754 serve as novel templates/scaffolds for development of more potent and specific anticobratoxin. In clinical applications, NCI121865 and NCI134754 would be very useful potential leads for the treatment of snakebite victims.

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Fac. of Grad. Studies, Mahidol Univ. Ph. D. (Pharmaceutical Chemistry and Phytochemistry) / 107

# APPENDIX

#### Appendix A

#### Using AutoDock for Virtual Screening

Using AutoDock for virtual screening was written by William Lindstrom, Garrett M. Moris, Christoph Weber and Ruth Huey (109). The paper will introduce you to the process of virtual screening using UNIX shell commands and python scripts in the AutoDock suite of program. There are nine steps to prepare a library of ligand files and corresponding AutoGrid and AutoDock parameter files for the library, use AutoGrid to calculate maps, launch AutoDock calculations for each ligand followed by analysis steps (Scheme A1).



Scheme A1. Virtual screening map.

The directories were created as shaded boxed with line to illustrate the data structure on the local computer (Scheme A2). The diversity in pdb file is available at TSRI. The diversity set and control peptide were put into Ligands directory.



Scheme A2. Virtual Screening data.

First, \$VSTROOT was set up. \$VSTROOT is a short cut to the directory in which the virtual screening activities will take place. Open the terminal and then type this at the UNIX or Linux prompt:

Type this: cd /usr/tmp/tutorial/VSTutorial source scripts/ex00.csh echo \$VSTROOT cd \$VSTROOT

The detail in scripts/ex00.csh

```
#!/bin/csh
# $Id: ex00.csh, v 1.3 2005/01/31 18:11:28 lindy Exp $
#
#
Because this script uses "pwd" to set VSROOT it matters where (which directory
# you run it from. This script should be run as "source ./scripts/ex00.csh" So, after you
# did your "cvs co VSTutorial" a "VSTutorial" directory was created and that's the
# one that should be your working directory when you source this script.
#
# set up the root directory of the Virtual Screening Tutorial
#
setenv VSTROOT `pwd`
```

The pdbq files were created by prepare\_ligand.py, a python script in the AutoDockTools module, and how to use it in a Unix foreach loop.

Type this:  $\implies$  source scripts/ex00.csh

The detail in scripts/ex02.csh

```
#!/bin/csh
#
# $Id: ex02.csh, v 1.2 2005/01/31 00:48:01 lindy Exp $
#
# use the prepare_ligand.py script to creat pdbq files
cd $VSTROOT/VirtualScreening/Ligands
foreach f (`\ls *.pdb`)
        echo $f
        ../../prepareligand.py -l $f -o ``$f``q -d ../etc/ligand_dict.py
end
```

The cbtx pdbqs files and grid parameter files were prepared using ADT, a graphical user interface. Each AutoGrid calculation can create up to 6 atom maps plus the electrostatics map. At this point, cbtx\_1.gpf (ACHNOS map types) and cbtx 2.gpf (cbFIP map types) were created.

Type this: autogrid3 -p cbtx\_1.gpf -l cbtx\_1.glg autogrid3 -p cbtx\_2.gpf -l cbtx\_2.glg Fac. of Grad. Studies, Mahidol Univ. Ph. D. (Pharmaceutical Chemistry and Phytochemistry) / 111

Docking directories and the parameter files for each ligand in a library and control peptide were prepared and the dpf files were created by prepare\_dpf.py and the virtual screening data after prepared the dpf files was shown in Scheem A3.

Type this:  $\implies$  source scripts/ex08.csh

The detail in scripts/ex08.csh

```
#!/bin/csh
# $Id: ex02.csh, v 1.4 2005/01/31 16:33:49 lindy Exp $
cd Dockings
# Creat a subdirectory named < ligand> cbtx and populate it with the docking input
# files: a) the pdbq from the Ligands directory will be copied directory; and b) the
# maps will be linked to the Receptor directory; and , c) the dpf file will be created
# using prepare dpf.py:
foreach f (`ls ../Ligands/*.pdbq`)
set name = `basename $f .pdbq`
echo $name
mkdir "$name" cbtx
cd "$name" cbtx
ln –s ../../Receptor/cbtx.pdbqs .
ln -s ../../Receptor/cbtx*map*.
../../prepare dpf.py-l basename f -r cbtx.pdbqs \
    -p ga num evals=1000000 \
    -p ga_pop_size=150 \
    -p ga run=1 \
     -p rmstol=2.0
cd ..
```

end



Scheem A3. The virtual screening data to prepare the dpf files.

#### Using the TSRI Cluster: Bluefish

All input file preparation have done on the local computer. The interactive head node on the bluefish cluster is used to transfer the files from the computer to the cluster where the calculations will be carried out. A tar file of VSTutorial directory tree was created on the local computer and then transferred to bluefish using SSH Secure Shell Client.

Type this: cd /usr/tmp/tutorial tar -czvf VSTutorial.tar.gz VSTutorial The envoroment on the bluefish cluster must be set so that autodock3 executable and the python script.py are in the path. Next uncompress the VSTutorial tree and then launch the jobs by using submit.py script.

Type this:  $\Longrightarrow$  set path = (\$path /bluefish/people-b/applications/autodock)

tar -xzvf VSTutorial.tar.gz

foreach f (`/bin/ls \$VSTROOT/VirtualScreening/Ligands/\*.pdbq`)

set name = `basename \$f .pdbq`

echo \$name

cd \$VSTROOT/VirtualScreening/Dockings/"\$name"\_cbtx

submit.py "\$name"\_cbtx 1

end

The first step of anylyzing the results is to build a list sorted by energy of the lowest energy docking for each ligand. To do this, first collect all the lig\_rec.NNN.dlg to get lig\_rec.energies and then sort the lig\_rec.energies to create the file lig\_rec.energies.sort

Type this:  $\longrightarrow$  source scripts/ex10.csh

The detail in scripts/ex10.csh

#!/bin/csh

```
# $Id: ex10.csh, v 1.3 2005/01/31 02:27:03 lindy Exp $
# Extract the Free Energy and Docked Energy from the dlg files. From the docking
# log files, use grep to extract the lines containing the binding energy and docked
# energy of each complex. Use sed and awk to process these lines into the final
# output:
cd $VSTROOT/VirtualScreening/Dockings
foreach d (`/bin/ls`)
    echo $d
     egrep "^USER Estimated Free Energy of Binding |
^USER Final Docked Energy" $d/sd.dlg | sed "N;s/ \n//" |
awk -v n=$d 'BEGIN {N=n} { print N" "$8" "$15}' > $d/$d.energies
     end
# Save the best energy fro each docking in a single file in the pwd directory called
# all energies.list:
touch ../etc/all_energies.list
foreach d ( `/bin/ls` )
     echo $d
     head –l $d/$d.energies >> ../etc/all energies.list
     end
# Sort the all energies.list file to find your best docking:
cd ../etc
sort -k3n all energies.list > all energies.sort
```

# Appendix B

1YI5		20	CTX	1CTX		
NGI	Docking	NCI	Docking	NCI	Docking	
NCI	energy	NCI	energy	NCI	energy	
number	(kcal/mol)	number	(kcal/mol)	number	(kcal/mol)	
140032 <sup>a</sup>	-24.65	1940 <sup>b</sup>	-24.62	140032 <sup>a</sup>	-24.62	
86374 <sup>a</sup>	-21.40	140032 <sup>a</sup>	-24.19	86374 <sup>a</sup>	-22.68	
625324 <sup>b</sup>	-20.20	625324 <sup>a</sup>	-23.54	625324 <sup>b</sup>	-22.65	
274547 <sup>°</sup>	-14.87	274547°	-13.46	274547°	-14.39	
371884	-13.86	49487	-12.74	11241	-13.59	
49487	-13.54	11241	-11.93	371884	-13.30	
11241	-13.32	371878	-11.91	23128	-13.10	
371878	-13.19	372280	-11.77	134755	-12.95	
371876	-12.72	128437	-11.67	371876	-12.52	
3354	-12.36	23904	-11.66	3354	-12.49	
23128	-12.23	371884	-11.60	372046	-12.48	
128437	-12.19	372294	-11.51	64111	-12.44	
14410	-12.11	64111	-11.39	49487	-12.28	
113491	-12.07	140936	-11.35	134754	-12.08	
95926	-11.83	15520	-11.27	56452	-12.04	
23904	-11.82	36387	-11.26	43628 <sup>d</sup>	-11.94	
26645	-11.82	87877	-11.19	166366	-11.84	
53396	-11.82	113491	-11.19	600067	-11.84	
211736	-11.82	74702	-11.12	89166	-11.77	
343040	-11.82	303812	-11.09	211736	-11.84	
600067	-11.78	131547	-10.95	81509	-11.74	
43628 <sup>d</sup>	-11.70	16168	-10.89	100857	-11.72	
64111	-11.69	600067	-10.87	371878	-11.61	
372280	-11.67	60043	-10.83	37245	-11.55	
13728	-11.59	48630	-10.83	26645	-11.52	
79050	-11.59	43628 <sup>d</sup>	-10.82	131547	-11.52	
234766	-11.58	14410	-10.81	13831	-11.49	
36387	-11.57	116654	-10.76	95090	-11.46	
95090	-11.57	150412	-10.76	128437	-11.43	
12181	-11.56	205511	-10.75	12181	-11.41	
23217	-11.54	95926	-10.73	36387	-11.39	
659162	-11.54	131615	-10.73	63875	-11.37	
322921	-11.53	106505	-10.71	172614	-11.37	
93354	-11.49	372046	-10.71	402959	-11.35	
63875	-11.48	3354	-10.70	101825	-11.33	
119847	-11.48	5069	-10.68	106221	-11.32	
172614	-11.47	680517	-10.68	350625	-11.32	
93674	-11.46	56452	-10.65	14410	-11.31	
37245	-11.45	134754	-10.64	659162	-11.31	
131547	-11.41	282027	-10.60	105687	-11.29	
163443	-11.41	371876	-10.57	18877	-11.28	
409664	-11 36	120288	-10.52	633406	-11 28	
361815	-11.35	255980	-10.54	106505	-11.25	
227147 <sup>e</sup>	-11.33	26273	-10.53	26273	-11.21	

Table B1. The top 175 compounds for each  $\alpha$ -cobratoxin

1	1YI5		СТХ	1CTX		
NCI	Docking	NCI	Docking	NCI	Docking	
NCI	energy	NCI	energy	NCI	energy	
number	(kcal/mol)	number	(kcal/mol)	number	(kcal/mol)	
106218	-11.32	134755	-10.52	106218	-11.20	
134755	-11.30	23128	-10.51	34238	-11.17	
93241	-11.30	93241	-10.46	150412	-11.17	
106221	-11.28	4895	-10.44	23922	-11.15	
45583	-11.24	159228	-10.44	116654	-11.13	
42258	-11.23	44585	-10.43	339601	-11.13	
81509	-11.21	45583	-10.43	48693	-11.10	
348401	-11.21	99550	-10.43	93241	-11.09	
95916	-11.19	310669	-10.40	99550	-11.09	
61806	-11.18	127917	-10.39	11237	-11.07	
96021	-11.18	299137	-10.38	17474	-11.07	
361814	-11.16	13728	-10.37	115448	-11.04	
56452	-11.12	12181	-10.35	163443	-11.04	
60043	-11 12	101825	-10.34	18891	-11.00	
371880	-11 12	119847	-10.34	95926	-10.98	
76026	-11.10	227147 <sup>e</sup>	-10.34	113491	-10.96	
121865	-11.09	241993	-10.34	127917	-10.96	
324065	-11.09	683770	-10.34	39863	-10.93	
372046	-11.07	43413	-10.31	43413	-10.93	
88915	-11.07	172614	-10.31	45208	-10.92	
17245	-11.05	95090	-10.30	372060	-10.92	
16168	-11.03	116702	-10.28	372280	-10.91	
150412	-11.02	62598	-10.27	348401	-10.89	
303812	-11.02	45582	-10.25	82147	-10.82	
282027	-11.01	147758	-10.25	101824	-10.81	
74702	-11.00	343040	-10 24	17245	-10.81	
77037	-10.97	45208	-10.23	45583	-10.80	
119886	-10.96	10458	-10 21	371872	-10 79	
120688	-10.96	106221	-10.21	186066	-10.78	
659390	-10.96	659162	-10 21	42384	-10.76	
326203	-10.95	69540	-10 20	186057	-10.75	
371684	-10.94	105687	-10.16	303812	-10.75	
134754	-10.94	37245	-10.15	306698 <sup>e</sup>	-10 74	
4292	-10.97	62594	-10.14	116490	-10.73	
136469	-10.92	81509	-10.13	23217	-10.72	
372294	-10.92	101824	-10.13	73109	-10.72	
18877	-10.89	100857	-10.12	73254	-10.72	
166366	-10.89	211736	-10.12	93674	-10.72	
13726	-10.87	23217	-10 11	87877	-10 71	
371872	-10.87	85372	-10.11	23904	-10 70	
41274	-10.86	668394	-10.10	42258	-10.69	
106505	-10.86	41309	-10.09	88915	-10.69	
25485	-10.84	117949	-10.09	112796	-10.69	
133488	-10.84	306698 <sup>e</sup>	-10.09	121855	-10.68	
5069	-10.77	9746	-10.07	136563	-10.63	

Table B1. The top 175 compounds for each  $\alpha$ -cobratoxin (Continued)

Fac. of Grad. Studies, Mahidol Univ. Ph. D. (Pharmaceutical Chemistry and Phytochemistry) / 117

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1YI5		20	CTX	1CTX		
NCI	Docking	NCI	Docking	NCI	Docking	
NCI h	energy		energy	NCI	energy	
number	(kcal/mol)	number	(kcal/mol)	number	(kcal/mol)	
295272	-10.77	47704	-10.05	227147 <sup>e</sup>	-10.63	
10458	-10.76	73109	-10.04	205511	-10.62	
402959	-10.76	163443	-10.04	295272	-10.62	
339601	-10.75	43131	-10.02	351814	-10.62	
69573	-10.72	638156	-10.02	99543	-10.59	
299137	-10.71	646824	-10.02	119847	-10.59	
101825	-10.66	154829	-10.01	36369	-10.55	
371682	-10.66	115448	-9.99	99671	-10.54	
39863	-10.65	327704	-9.99	99804	-10.54	
350625	-10.65	294149	-9.98	169453	-10.54	
99671	-10.64	26645	-9 97	190382	-10.54	
116654	-10.63	148354	-9.96	372294	-10 54	
14711	-10.62	402959	-9.96	62594	-10.52	
117614	-10.61	119886	-9.95	119886	-10.52	
18451	-10.60	13975	-9.91	159628	-10.51	
641691	-10.60	82147	-9.91	656202	-10 51	
116702	-10 59	86008	-9.90	53396	-10.50	
159228	-10.55	94914	-9.90	112675	-10.50	
7520	-10.54	120688	-9.90	34352	-10.49	
169534	-10.54	659390	-9.90	306711	-10.47	
48693	-10.53	53396	-9.88	13480	-10.46	
13/13	10.55	327702	0.88	69540	10.16	
100836	-10.51	18877	-9.88	121865	-10.40	
109850	-10.50	65828	-9.87	121805	-10.45	
13987	-10.30	120634	-9.87	978/15	-10.43	
373058	-10.49	18451	0.83	11///2	10.42	
24850	-10.48	10 <del>4</del> 51 99015	-9.85	114442	-10.42	
24839	-10.47	88913 41274	-9.83	14//38	-10.42	
109309	-10.47	412/4	-9.82	572045	-10.41	
3/1/3	-10.43	92828	-9.82	10424	-10.39	
102404	-10.43	42364	-9.81	74702	-10.39	
205511	-10.43	88135	-9.80	3/1684	-10.39	
14119	-10.42	326381	-9.79	6391/4	-10.39	
319434	-10.42	8/9/	-9./8	35489	-10.38	
1/4/4	-10.41	23922	-9.//	4//04	-10.38	
633406	-10.41	133488	-9.//	28136	-10.37	
85372	-10.40	211340	-9.77	44585	-10.37	
/524	-10.39	140905	-9.76	5069	-10.35	
14//58	-10.39	1/0328	-9.75	1545/2	-10.35	
03828	-10.3/	41003	-9./3	18883	-10.33	
527704 126472	-10.36	0391/4	-9./3	52445	-10.55	
1304/2	-10.34	131433	-9.72	154829	-10.32	
1/0328	-10.34	170008	-9.72	350363	-10.32	
84130	-10.33	1/245	-9./1	/6026	-10.31	
42384	-10.32	1120/3	-9./1	45582 84120f	-10.30	
326381	-10.33	81620	-9.70	84130	-10.30	
8/8//	-10.31	93916	-9.70	65828	-10.29	

Table B1. The top 175 compounds for each  $\alpha$ -cobratoxin (Continued)

1YI5		20	CTX	1CTX		
NCI	Docking	NCI	Docking	NCI	Docking	
NCI	energy	NCI	energy	NCI	energy	
number	(kcal/mol)	number	(kcal/mol)	number	(kcal/mol)	
48223	-10.30	117614	-9.68	351351	-10.29	
59620	-10.30	166366	-9.68	116491	-10.28	
143491	-10.30	372127	-9.68	294149	-10.28	
148354	-10.30	42258	-9.67	13987	-10.27	
357756	-10.30	321598	-9.67	41663	-10.27	
610930	-10.30	13987	-9.65	88135	-10.27	
52445	-10.27	23173	-9.65	98905	-10.27	
127917	-10.27	179703	-9.65	162404	-10.27	
31076	-10.26	51132	-9.64	282027	-10.26	
69540	-10.26	84130 <sup>f</sup>	-9.64	288391	-10.26	
5113	-10.25	109834	-9.64	343040	-10.26	
15909	-10.25	121847	-9.64	401366	-10.25	
23583	-10.25	372074	-9.64	186063	-10.24	
74860	-10.25	59620	-9.63	293161	-10.24	
43513	-10.23	13726	-9.61	372037	-10.24	
190382	-10.23	190382	-9.61	659390	-10.24	
18883	-10.22	48388	-9.60	35582	-10.22	
306698 <sup>e</sup>	-10.22	75749	-9.60	60043	-10.22	
310669	-10.22	106218	-9.60	79050	-10.22	
62349	-10.21	332429	-9.60	131815	-10.22	
176327	-10.21	361815	-9.60	176327	-10.22	
26273	-10.19	659694	-9.60	299137	-10.22	
82802	-10.19	99671	-9.59	45554	-10.21	
683770	-10.19	121865	-9.59	130830	-10.21	
90616	-10.18	45554	-9.58	10458	-10.20	
319095	-10.18	52445	-9.58	37881	-10.20	
407628	-10.18	326203	-9.57	150117	-10.20	
9608	-10.17	162404	-9.56	641691	-10.20	
82147	-10.17	31076	-9.55	74472	-10.19	
121137	-10.17	121855	-9.55	86467	-10.18	
668394	-10.17	333003	-9.55	99676	-10.18	
45208	-10.16	92601	-9.54	373058	-10.18	
99543	-10.16	409664	-9.54	631160	-10.18	
94820	-10.15	631160	-9.54	657704	-10.17	
34238	-10.14	11237	-9.53	88903	-10.16	
134244	-10.14	339601	-9.53	45236	-10.15	
306711	-10.14	641691	-9.53	201579	-10.15	
89166	-10.13	76064	-9.52	59486	-10.13	
3364	-10.11	89166	-9.52	140936	-10.13	
105687	-10.11	136469	-9.52	14143	-10.12	

Table B1. The top 175 compounds for each  $\alpha$ -cobratoxin (Continued)

<sup>a</sup>The structure has Se atom <sup>b</sup>The structure has Ag atom <sup>c</sup>The structure has Au atom <sup>d</sup>The structure has Cu atom

<sup>e</sup>The structure has Ni atom

<sup>f</sup>The structure has As atom

# Appendix C

		L	S			A	с			B	t	
Compound	Т	1	Т	2	Γ	1	Т	2	Т	1	Т	2
1	471	421	460	423	849	775	788	645	573	526	608	582
2	423	449	400	359	809	857	789	819	517	470	489	527
3	467	526	442	506	842	866	754	741	572	607	623	623
4	501	430	462	417	953	820	804	789	552	490	561	510
5	391	409	359	383	776	732	717	705	379	438	395	454
6	381	348	345	331	762	743	653	641	240	295	277	288
7	523	460	446	446	719	617	678	654	527	572	495	577
8	55	40	36	31	80	62	76	60	42	31	3	29
9	424	500	383	438	767	790	687	735	461	503	415	569
10	120	105	101	108	244	307	285	285	312	381	324	378
11	483	597	509	522	845	86	775	766	511	616	495	716
12	389	364	372	346	781	786	698	707	476	445	473	469
13	432	466	388	454	890	848	838	769	456	645	491	612
14	404	433	375	391	686	817	640	755	298	429	317	417
15	501	394	482	392	762	739	711	679	51	548	535	539
16	374	404	350	362	700	602	626	579	364	394	344	413
17	416	395	355	328	662	740	593	622	363	362	414	365
18	419	463	408	454	779	901	746	743	543	702	556	686
19	475	550	462	447	246	293	211	227	352	436	334	466
20	445	38	388	390	840	785	766	738	492	684	467	675
21	345	436	299	416	845	899	853	859	747	595	683	597
22	414	420	398	382	818	883	781	727	700	621	673	618
23	115	113	88	112	29	38	28	25	227	246	223	245
24	458	344	433	371	794	812	828	704	622	683	621	670
25	433	453	468	435	716	804	771	778	697	606	734	714
26	362	399	345	358	645	659	602	643	486	498	529	513
27	258	252	212	231	448	432	392	370	440	452	471	421
28	423	344	357	304	367	351	386	353	614	583	581	586
29	294	342	270	295	215	186	186	175	452	397	421	417
30	404	485	436	437	804	950	769	865	625	675	655	638
31	534	493	444	430	792	790	738	737	652	700	701	726
32	354	258	305	245	632	542	548	528	590	653	538	570
33	485	403	451	396	816	880	745	780	705	651	717	673
34	425	443	408	387	937	724	778	656	631	640	625	562
35	419	517	368	456	830	771	798	729	606	565	573	601
36	408	357	355	332	789	777	717	707	627	580	595	592
37	418	432	418	373	694	826	655	721	643	722	616	739
38	422	415	378	412	705	619	645	594	644	647	615	620
39	434	478	385	456	804	667	711	588	562	545	584	518
40	368	399	326	347	609	710	592	675	416	374	405	356

Table C1. The screening test data from radioligand competition assay

Commoniad		L	<b>S</b>			A	с			1	Bt	
Compound -	Т	1	Т	2	Т	<b>`1</b>	Т	2	Т	'1	Т	2
Negative control	428	403	295	337	837	838	770	795	700	736	671	750
(PBS)	478	473	440	404	790	747	778	669	701	761	684	808
	466	417	401	395	805	805	724	783	720	741	694	707
	442	456	421	363	840	972	690	860	694	673	709	664
Positive control	12	14	13	13	21	16	14	13	16	11	8	13
(MLA)	12	14	9	11	20	27	26	27	19	20	18	18
	16	11	13	9	16	15	17	21	15	19	20	19
	22	9	12	13	29	30	24	24	20	17	19	18

Table C1. The screening test data from radioligand competition assay (Continued)

# Appendix D

	$H_{3C} \xrightarrow{CH_{3}}_{H_{2}C} \xrightarrow{COOH}_{H_{2}}_{OOH}$		
	H <sub>3</sub> C CH <sub>2</sub> H <sub>2</sub> C COOH CH <sub>2</sub> H <sub>2</sub> C CH <sub>2</sub> COOH	H <sub>3</sub> C, CH <sub>3</sub> CH <sub>3</sub> OCH <sub>2</sub> C(CH <sub>3</sub> ) <sub>2</sub> NH <sub>2</sub> H 1199	
Сн <sub>2</sub> снимь,ссоон	H <sub>3</sub> C CH <sub>3</sub> CH <sub>3</sub> OCH <sub>2</sub> C(CH <sub>3</sub> ) <sub>2</sub> NO <sub>2</sub> H H 1482	HOLE HE CHECOCH	
	СН,СН,ССО№НИН₂ ССО№НИН₂ 1760		он-он но-он но-он но-он нс-он он-он он-он 1945
N SH 2006	CHOHCH,NED CHOHCH,NED CHOHCH,NED CHOHCH,NED	in the content of the	
ноосн <sub>2</sub> с н сн <sub>2</sub> соон   соон 2347	2518	OH OH OH OH 2561	(СH <sub>2</sub> ) <sub>2</sub> SO <sub>3</sub> H 2737
ноsон	но он н <sub>3</sub> с он но сн <sub>3</sub> 2805	H <sub>3</sub> C CH <sub>3</sub> CH <sub>3</sub> CH <sub>3</sub> CH <sub>2</sub> CH	Cl H <sub>3</sub> C Cl Cl Cl Cl Cl Cl Cl Cl Cl Cl Cl Cl Cl
CH <sub>2</sub> CO <sub>2</sub> H CH <sub>2</sub> CO <sub>2</sub> H CH <sub>3</sub> 3001	(CH <sub>2</sub> ) <sub>3</sub> CO <sub>2</sub> H 3036		
ALL	MEAT MILES 3323		$\underset{\substack{\mu, \mu}{\leftarrow} \int_{0,\pi} \int_{0,\pi}$

NCI Diversity Set

нл 3385	H <sub>3</sub> C, CH <sub>3</sub> (MNCH <sub>3</sub> ) CC(0)CH <sub>2</sub> SCN H 3535		HN HN HN H HN H H H H H H H H H H H H H
ньс-С-5-н-С-5-н-	H <sub>3</sub> C 3753	СООН ОН 3907	4265
2 4292	HG NO 0 4357	Me <sub>3</sub> CH <sub>2</sub> CO S OCH <sub>2</sub> CMe <sub>3</sub> 4622	HOOC
Let			он вын он 4972
осноснис области 5069	$H_{H,C} \rightarrow OH \rightarrow O$	с <u>С С С С С С С С С С С С С С С С С С С</u>	
CHECOOH		но о о он 5550	он он он о          нон₂с—с —с —с —сн₂мн₂ 5554
	$\begin{array}{c} & & & \\ & & & & \\ & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & &$	$\overset{\alpha}{\underset{H_{N}}{}{\underset{H_{2}}{\underset{H_{2}}{}{\underset{H_{2}}{\underset{H_{2}}{}{\underset{H_{2}}{\underset{H_{1}}{\underset{H_{2}}{\underset{H_{1}}{\underset{H_{2}}{\underset{H_{2}}{\underset{H_{1}}{H$	HBUL CHILD C
5992	CH <sub>2</sub> SCH <sub>2</sub> CH <sub>2</sub> CO <sub>2</sub> H	Br COOH Br COOH	
о <sub>2</sub> N-СH <sub>2</sub> ссон 6225	6239		соон 6497
S HN HOOC HN HOOC H	H <sub>b</sub> C H <sub>b</sub> N H <sub>b</sub> N NH <sub>b</sub> N NH <sub>b</sub> N NH <sub>2</sub> 7215		СН <sub>3</sub> СН <sub>3</sub> НО—N=ССН <sub>2</sub> СН <sub>2</sub> С-N—ОН 7307
0 0 0 0 0 0 0 0 0 0 0 0 0 0	0 	N <sub>2</sub> =CHCOCH <sub>2</sub> CH <sub>2</sub> CH <sub>2</sub> CHCO <sub>2</sub> H /NH <sub>2</sub> 7365	H <sub>2</sub> N SO <sub>2</sub> NH But But 7436

HO THE PERFORMANCE PERFORPERFORMANCE PERFORMANCE PERFO	Total Tota Total Total Tota	соон ноосн <sub>2</sub> с—с—снсоон 7616	
CH <sub>2</sub> COOH CH <sub>3</sub> CH <sub>3</sub> CH <sub>3</sub> 7734	H <sub>3</sub> C H <sub>3</sub> C	Соон соон соон 7810	но,5 — — — — — — — — — — — — — — — — — — —
ссегмег соон 7962	₩_ <sup>2</sup> 8090	HEN H NH NH NH NH NH 8174	
HOP HIGH HIGH	HO 3S NO2 8960	O2N OH O2N OH 9040	HCH <sub>2</sub> CH <sub>2</sub> C H <sub>3</sub> C H <sub>2</sub> CH <sub>2</sub> CH <sub>2</sub> CH H <sub>3</sub> C H <sub>2</sub> OH 9279
иоосң₂с—№—сң₂соон 9345	p p p p p p p p p p p p p p	стробов в 9608	но соон 9746
H <sub>2</sub> NSCH <sub>2</sub> CH <sub>2</sub> COOH 9906	SCH <sub>2</sub> CH <sub>2</sub> 9928	CH <sub>2</sub> CMe(NH <sub>2</sub> )COOH 9948	Н <sub>2</sub> Н <sub>2</sub> СООН С С СООН NH <sub>2</sub> 9959
C(CN) 99993	С(СС)2H)CH2CO2H 10134		
	С С С С С С С С С С С С С С С С С С С		H <sub>2</sub> C HO 10470
HO CHANNEH CHANNEH H H H H H H H H H H H H H H H H H H	$ \underset{(Me)_{E}CH_{0}(Me)_{2}}{\overset{OH}{\underset{C}{\mapsto}}} $ 10572	10573	
<sup>i-Pr</sup> но СН <sub>3</sub> 10776	S(CH <sub>2</sub> ) <sub>3</sub> -N NH 10777		но предоставляется на предоставл
NM62 HO,5 MAM2 HO,5 MAM2		NHCH <sub>2</sub> CH <sub>2</sub> OH H N N N N N N N 11590	$ \underset{NHCH_2CH_2CHMe_2}{\overset{H}{\longrightarrow}} 11607 $

Appendix / 124

		ноосн <sub>2</sub> с - <sup>N</sup> сн <sub>2</sub> соон	HO OH As NH <sub>2</sub>
11656	L 11668	11773	11825
CI CI CI CI CI	$\underset{H_2N}{\overset{O}{}_{\underset{H}{}}} \underset{H_1}{} \underset{H_2}{} \underset{H_2}{\overset{H_2}} \underset{H_2}{} \underset{H_2}} \underset{H_2}{\overset{H_2}{} \underset{H_2}{} \underset{H_2}{\overset{H_2}} \underset{H_2}} \underset{H_2}{\overset{H_2}} \underset{H_2}{\overset{H_2}$	$\overset{H,C}{\underset{NP_{b}}{\leftarrow}} \overset{N}{\underset{NP_{b}}{\leftarrow}} \overset{O}{\underset{NP_{b}}{\leftarrow}} \overset{O}{\underset{NP_{b}}{\leftarrow}} \overset{O}{\underset{NP_{b}}{\leftarrow}} \overset{OP_{b}}{\underset{NP_{b}}{\leftarrow}} O$	
11850		12155	но он 12161
			NHCOCH <sub>2</sub> CH <sub>2</sub> COOH
12181		12363	12418
			HO AS
۵ <sup>/</sup> 12492	0	12636	12639
		0=As	
0H 0 12955		SO <sub>2</sub> NHNHCO 13150	OF HOH OF HOH OH OH I3161
	NHCH <sub>2</sub> CH(OH)CH <sub>2</sub> OH 13220	G CH <sub>2</sub> NH <sub>2</sub> NH(CH <sub>2</sub> ) <sub>3</sub> CH <sub>3</sub> 13239	CI C
$\overset{H_{2}C}{\underset{0}{}} \xrightarrow{H_{1}} \overset{H_{1}}{\underset{N+}{}} \xrightarrow{H_{1}} \overset{H_{1}}{\underset{N+}{}} \xrightarrow{H_{1}} \overset{H_{1}}{\underset{Cl}{}} \xrightarrow{Cl} \overset{Cl}{\underset{Cl}{}} \xrightarrow{I3359}$	H <sub>3</sub> C, H, NHCOCH <sub>2</sub> S-C, a 0 13364		o A 13399
		CH3 H NH2 13545	NH H <sub>2</sub> N NH <sub>2</sub> SO <sub>2</sub> NH <sub>2</sub> 13572
CI- CI- SCH <sub>2</sub> COOH 13579	H <sub>3</sub> CO H <sub>3</sub> CO	ссн <sub>а</sub> соон 13597	он о
13726			ноос н 13770

OH SO2NHCH2CH2OH	о "Вь сн <del>с</del> нсоон	O <sub>2</sub> N OH	CH3
но брани в 13776	но он 13778	но то по	сн <sub>3</sub> но 13816
	13950	н <sub>2</sub> N SO <sub>3</sub> HE tO 13970	50,H N=N NH <sub>2</sub> 13975
		HO NH H OH	H <sub>3</sub> C N (CH <sub>2</sub> ) <sub>4</sub> COOH
13984	<sup>8</sup> 13987	13996	14119
		H <sub>2</sub> N NH <sub>2</sub> COOH	H <sub>4</sub> C H <sub>4</sub> C H <sub>4</sub> C H <sub>5</sub> C
сн <sub>3</sub> 14141	14143	14161	14163
NH CH <sub>2</sub> NH CH <sub>3</sub>	SCH <sub>2</sub> COOH	NHCOCH <sub>2</sub> NH <sub>2</sub>	H <sub>2</sub> C CH <sub>3</sub> Cl H <sub>2</sub> C C(0)OH <sub>2</sub> O H <sub>3</sub> C
сн <sub>2</sub> он 14179	14186	14288	14343
	Me <sub>2</sub> NH <sub>2</sub> C S S CH <sub>2</sub> NMe <sub>2</sub>		
14410	14543	он мн 14555	14703
ССН <sub>2</sub> ССН <sub>2</sub> ССН <sub>2</sub> ССН <sub>3</sub> СССН <sub>3</sub> СОСН СОСОН 14711	ноос	0 СООН H <sub>2</sub> N Н СООН 14983	(CH <sub>2</sub> ) <sub>3</sub> COOH
			Сна соон
₫н 15181	15226	15234	ноос он <sub>а</sub> 15520
HO OSB OH 15596			соон носн носн нос н <sub>о</sub> он нос н <sub>о</sub> он 15608
коон носн носн носн сн_он сн_он сн_он сн_он сн_о 15623	۲ ۱5784	сн(со <sub>2</sub> н)сн <sub>2</sub> соон 15889	
$ \begin{array}{c}                                     $	но <sub>3</sub> с 16163	ра	S(0)CH2000H

Appendix / 126

OH COOH CH-COOH	O <sub>2</sub> N Br H <sub>3</sub> O <sub>4</sub> O <sub>2</sub> N O <sub>2</sub> N O <sub>2</sub> N O <sub>2</sub> N O <sub>3</sub> O <sub>4</sub> O <sub>4</sub> O <sub>4</sub> O <sub>4</sub> O <sub>5</sub> O <sub>4</sub> O <sub>4</sub> O <sub>4</sub> O <sub>4</sub> O <sub>4</sub> O <sub>4</sub> O <sub>4</sub> O <sub>4</sub>	HO NH2 COOH	SO <sub>2</sub> OCH <sub>2</sub> COOH
16311	16414	16535	H <sub>3</sub> C 16555
HO, SH 16865	HO HO SH OH 16866	рот, субрат, 17061	еto
	HOOCH_2CH_2C COOH		Mest CityOrJanis 17383
	но он он 17391	$ \underbrace{ \begin{pmatrix} \rho^{r_{s}} \\ \rho \\$	NH NHCNHC(NH <sub>2</sub> )=NH HO <sub>3</sub> S
	H <sub>2</sub> N NH <sub>2</sub> CH <sub>3</sub> 17719		S-(CH <sub>2</sub> )2-N 17776
		SMe 17809	
	CH <sub>2</sub> CONHNH <sub>2</sub> 18355	H H H H H H H H H H H H 18360	
But H <sub>3</sub> C UCH <sub>23</sub> -N But 18430		H <sub>5</sub> C H <sub>5</sub> C H <sub>5</sub> C N-CH <sub>2</sub> CH <sub>2</sub> M N N N N N N N N N N N N N N N N N N N	
	ны=сомналин ССОмналин 18758	С С С С С С С С С С С С С С С С С С С	HOOC NHSO2CH2 18778
		E3MH2OH4G C S 18883	(CH2)MMe S
18997	Сснь 19039	$HN \xrightarrow{H} N \xrightarrow{H} N \xrightarrow{CH_3} H_3$ $HN \xrightarrow{H} N \xrightarrow{H} N \xrightarrow{CH_3} H_3$ $19097$	CH2CH,NB2 N N H5 19119
	19147		Соон 19496

$\wedge \rightarrow$	Ŷ	ę	
	N-CH <sub>2</sub> OC(0)Et	CH(CO <sub>2</sub> H)CH <sub>2</sub> Ph	
носо на 19509	٦ 196	30 19760	
$\sim$	HO	CH <sub>2</sub> C(CO <sub>2</sub> H) <sub>2</sub> NH <sub>2</sub>	
	N N N N N N N N N N N N N N N N N N N		
19976	199	95 20138	20207
	Соон	Соон	NH
HLCO A	СООН	Соон	CI S NH2
20410	207	07 20708	Сі 20878
	CH2NH2	СH3	
	HOCH <sub>2</sub> OH	CH <sub>3</sub>	Соон
H <sub>3</sub> C 21061	H <sub>3</sub> C N 212	78 CH <sub>2</sub> CH <sub>2</sub> 21434	0 <sub>2</sub> N <sup>2</sup> 21438
HBU CH <sub>2</sub> NMe <sub>2</sub>	HOOCH_CH_C		
Ę.	HO CH <sub>3</sub> Br		
Pri 21518	Br 215	88 21702	21706
ФОН		HOOC s H	H <sub>3</sub> C CH <sub>2</sub> NMe <sub>2</sub>
NH <sub>2</sub>			H <sub>1</sub> C CH <sub>2</sub> NMe <sub>2</sub>
- 21948	222	25 22242	22368
CH <sub>2</sub> CH <sub>2</sub> -N	HOOCH		
22819		54 22959	22976
N CH3			αφή το
СН3 СН3 22992		16 23082	23128
Br CHMeCH <sub>2</sub> CH <sub>2</sub> C(0)OMe		Сі, сңзсңалан	
CH3			
23159	231	73 CI 23183	23217
		CN CH2OMe	Соон
			H <sub>JC</sub> S(0)0-N
23478	235	83 <sup>I</sup> / <sub>Pr</sub> 23807	23833
		H <sub>3</sub> G <sub>R<sub>0</sub>, H (CH<sub>2</sub>)<sub>3</sub>CHMe<sub>2</sub> (CH<sub>2</sub>)<sub>3</sub>CHMe<sub>2</sub></sub>	
но			
23880	239	04 Haven 23922	23925

nct for			Et O SH
24037	24047	24048	الا 24076
	CH <sub>3</sub> NH CH <sub>3</sub> NH	S CH <sub>3</sub> CH <sub>3</sub> CH <sub>3</sub>	NH <sub>3</sub>
ά (ċH <sub>2</sub> ) <sub>3</sub> NMe <sub>2</sub> 24113	24205	24479	24515
	24859	CI S OH 24915	
H <sub>2</sub> N	HOOC CH <sub>3</sub> Br IIIIII Br H COOH	цо но	
25247	25303	25431	он он 25485
HN S S NH <sub>2</sub> 25561	H2NO2S H2N SO2NH2 SO2NH2 25857	HeNO28 CI CH3 25861	н <sub>2</sub> NO <sub>2</sub> S <sup>C),HH2</sup> H <sub>2</sub> NO <sub>2</sub> S <sup>C</sup> CH <sub>3</sub> 25869
н <sub>зс</sub> 26074	Соон 26079		0, COOH t-Bu Cl 26118
$\underset{O}{\overset{OH_3}{\underset{H}{\overset{O(0)OEt}{\overset{O(0)OEt}{\overset{O(1)}{\overset{O(1}{\overset{O(1)}{\overset{O(1)}{\overset{O(1)}{\overset{O(1}{\overset{O(1}{\overset{O(1)}{\overset{O(1}{\overset{O(1}{\overset{O(1}{\overset{O(1}{\overset{O(1}{\overset{O(1}{\overset{O(1}{\overset{O(1}{\overset{O(1}{\overset{O(1}{\overset{O(1}{\overset{O(1}{\overset{O(1}{\overset{O(1}{\overset{O(1}{\overset{O(1}{\overset{O(1}{\overset{O(1}}{\overset{O(1}}{\overset{O(1}}{\overset{O(1}{O$		$ \begin{array}{c} \begin{array}{c} \begin{array}{c} \begin{array}{c} \begin{array}{c} \end{array}\\ \end{array}\\ \end{array}\\ \end{array} \\ \begin{array}{c} \end{array}\\ \end{array} \\ \end{array} \\ \begin{array}{c} \end{array} \\ \end{array} \\ \end{array} \\ \end{array} \\ \begin{array}{c} \end{array} \\ \end{array} \\$	СН3 ССН3 ССОН ССН3 ССООН 26349
26382	HO CH <sub>0</sub> DC(0)CH <sub>2</sub> CH <sub>2</sub> Ph HO 26645	NH2 0CH3 H300 26669	Henry Colored
26720	CONFINET: 26904		CH <sub>3</sub> CH <sub>3</sub> CH <sub>3</sub> 27118
нос	CH <sub>2</sub> CH <sub>2</sub> CH <sub>2</sub> -N 27476	27515	Страниции и стр 27895
СП-СП-СП-СП-СП-СП-СП-СП-СП-СП-СП-СП-СП-С	27959	СН3 ССН3 ССОН NH2 28059	28081
	28136	28311	CH <sub>6</sub> H <sub>6</sub> C N (CH <sub>2</sub> N 28408
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СООН ССООН ССООН ССООН ССООН ССООН ССООН ССООН ССООН ССООН ССООН ССООН ССООН ССООН ССООН ССООНСКИ СТОРОСТИТИТИ СООНСКИ СТОРОСТИТИТИТИТИТИТИТИТИТИТИТИТИТИТИТИТИТИТИ	NH <sub>2</sub> 28527	HOOC HOOC 28620	HOOCSCOOH28743
SO2NHCH2CH2SO3H	ньс соон ньс 28992	н,с 29096	осн <sub>3</sub> соон <sub>H<sub>3</sub>C</sub> 29101
HN HN NH 29192	H <sub>3</sub> CO <sup>NHCH2</sup> CH2COOH H <sub>3</sub> CO	H <sub>B</sub> C <sub>S</sub> 29424	29448
но С С С С С С С С С С С С С С С С С С С			но-с-он ссон 30144
СH <sub>3</sub> 30154	ноос он но сосн 30188	N <sub>3</sub> OH N <sub>3</sub> 30408	н <sub>з</sub> с соон а 30534
HPr N=CH HO CH <sub>3</sub> HO 30712	ССООН 30799	₩ ₩ 31076	
GH_OC(0) CH_OC(0) CI	NH <sub>2</sub> Он 31660	NHAC 31698	
<sup>14</sup> + + + + + + + + + + + + + + + + + + +	но соон з2963	CH <sub>2</sub> F HN H S 33035	
сі — Сі — ЗЗЗІ45	33389	ньс с соон о с сна сна 33495	33571
	он 33761	N Ссоон Ссн <sub>3</sub> 34003	CI 34008

		CH <sub>3</sub> H <sub>3</sub> C	S S N NO2
34238	34240	34352	34444
		HN CH <sub>3</sub>	
34486	34506	34749	<sup>он</sup> 34824
сн з4908	34924	35049	стерия 35334
$\overset{\alpha}{\underset{\alpha}{\leftarrow}}\overset{\beta}{\underset{\alpha}{\leftarrow}}\overset{\mu_{\alpha}}{\underset{\alpha}{\leftarrow}}\overset{\alpha}{\underset{\alpha}{\leftarrow}}\overset{\beta}{\underset{\alpha}{\leftarrow}}\overset{\mu_{\alpha}}{\underset{\alpha}}$	35427	35446	л сн <sub>2</sub> с - сн <sub>2</sub> с - 35450
		HN, , , , , , , , , , , , , , , , , , ,	
35489	35582	35605	ö 35682
35755	H <sub>3</sub> C - 35761	соон лн з5790	Che Che Che Che Che Che Che Che Che Che
	0 <sub>2</sub> N NO <sub>2</sub> HO 35950	ет 35989	35991
36369		H <sub>3</sub> C N CH <sub>2</sub> OH <sub>2</sub> NMe <sub>2</sub> O N CH <sub>2</sub> OH O CH <sub>3</sub> CH <sub>3</sub> 36533	
HEN S S S S S S S S S S S S S S S S S S S		HG COOH HG CHB CHG COOH HO CHB CHG CHG COOH HO CHG	John John John John John John John John



H <sub>2</sub> N, O CH <sub>3</sub>		соон	SO <sub>2</sub> H OH
40245	ин(сн₂₃он 40380	40384	соон 40596
40614		$H_{3C} \rightarrow H_{4C} \rightarrow H$	$H_{2C}$ $H_{3}$ $H_{2C}$ $H_{3}$ $H_{2C}$ $H_{3}$ $H_{2C}$ $H_{3}$ $H_{2C}$ $H_{3}$ $H_{2C}$ $H_{2}$
СН <sub>3</sub> СН <sub>3</sub> H <sub>2</sub> N ОН 40792	но рон но рон 40837	соон H <sub>2</sub> NNH <sub>2</sub> 41117	
HOOC COOH 41309	H00C 41313	HOOC COOH 41355	ньс 41431
$ \underset{OH}{\overset{OH}{\longrightarrow}} \underset{41458}{\overset{OH}{\longrightarrow}} $	н <sub>зсо</sub> Он он 41499	41550	HOOC 41629
		ноос соон	
	$H_3C$	H <sub>2</sub> N NH <sub>2</sub>	н <sub>3</sub> с Он
<sup>II</sup> 42010	42069	42167	42199
		HOOC	HN SH
42255	42258	42325	42352
		HO HO HO HO	42636
42645	H CH3 CH3 CH3 CH3 CH3 CH3 CH3 CH3 CH3 CH		
о N сн <sub>3</sub> 42665	42679	H3C CH3 CH3 CH3 CH3 42688	42695



	OH OH OH OH	H <sub>2</sub> N S	H <sub>2</sub> N NH
<sup>c</sup> 45730	45741	45869	45870
HO NH2 HO H			H <sub>3</sub> C
46046	ноосень 46080	н,с Сн. 46480	Br 46529
<u>46538</u>		о соон 46647	н <sub>э</sub> со соон 46669
	H <sub>3</sub> C COOH CH <sub>3</sub> NH <sub>2</sub> 46709	Ссоон 47067	
о Соон 47083	о соон 47085	соон 47106	
СН <sub>3</sub> 47326		ноос соон 47621	н <sub>зсо</sub> 47652
H,C, N, H,C, H,C, H,C, H,C, H,C, H,C, H,	OH N=N SOJH 47716	$\downarrow_{t=1}^{CI} \qquad \qquad$	ОН ОН NH2 NH 47799
	$\begin{array}{c} \overset{OH}{\longleftarrow} & \overset{OH}{\longleftarrow} & \overset{OH}{\longleftarrow} & \overset{OH}{\longleftarrow} \\ \overset{CI}{\leftarrow} & \overset{OH}{\leftarrow} & \overset{OH}{\leftarrow} \\ 47932 \end{array}$	HEC CH6 HEC HCH6 HEC HCH6 HCH6 HCH6 HCH6 HCH6 HCH6 HCH6 HCH	47938
	ACO CH3 H H SH H H H H H H H H H H H H H H H	$\downarrow^{OH}_{H_{bC}} \downarrow^{OH}_{CH_{3}} \downarrow^{OH}_{H_{bC}} \downarrow^{OH}_{CH_{3}}$ $48151$	$H_{3C} \xrightarrow{CH_{3}}_{H_{3}C} \xrightarrow{N}_{H}$
	$\overset{\text{H}_{\text{H}}\text{C}}{\underset{\text{CH}_{3}}{\overset{\text{OH}}{\overset{\text{H}}{\underset{\text{H}}}}}+\underset{\text{H}_{0}}{\overset{\text{OH}}{\underset{\text{CH}_{5}}{\overset{\text{H}_{6}}{\underset{\text{CH}_{5}}{\overset{\text{OH}}{\underset{\text{CH}_{5}}{\overset{\text{H}}{\underset{\text{CH}_{5}}{\overset{\text{OH}}{\underset{\text{CH}_{5}}{\overset{\text{H}_{6}}{\underset{\text{CH}_{5}}{\overset{\text{OH}}{\underset{\text{OH}}}}}}}}}}}}}}}}}$	H <sub>3</sub> CH <sub>3</sub> H <sub>3</sub> CH	$\xrightarrow{H_bC}_{H_bN} \xrightarrow{P_b}_{N} \xrightarrow{P_b}_{N+2} \xrightarrow{H_b}_{N+2} H$

	-		
		$\xrightarrow{H_{0} \subset H_{0}}_{H_{0} \subset H_{0}} \xrightarrow{H_{0} \subset H_{0}}_{H_{0} \subset H_{0}} \xrightarrow{H_{0} \subset H_{0}}_{48458}$	$\xrightarrow{H_{5}C_{H_{5}}CH_{5}}_{H_{5}C_{H_{5}}CH_{5}}$
Н SCH <sub>2</sub> CO <sub>2</sub> H 48521	HO HO HO HO HO HO HO HO HO HO HO HO HO H	$\begin{array}{c} \overset{OH}{\underset{H_2}{\overset{H_{H_2}{\overset{H}}{\overset{H}}{\overset{H}}{\overset{H}}{\overset{H}}{\overset{H}}{\overset{H}}{\overset{H}}{\overset{H}}{\overset{H}}{\overset{H}}{\overset{H}}{\overset{H}}{\overset{H}}{\overset{H}}}{\overset{H}}}{\overset{H}}}}{\overset{H}}}{\overset{H}}}{\overset{H}}}}}}}}$	
		N S CH2OCH2 48747	
			t-Bu CH_3NMeCH2 HO HO HO HO HO HO HO HO HO HO
48949	CH <sub>2</sub> CH(OH)C(CO)OEt 48961	CH_CH_CHOO_2H NHAC 49124	Ет-S-CH <sub>2</sub> -CH-CO <sub>2</sub> H NH <sub>2</sub> 49244
$ \begin{array}{c}                                     $	A9460	49487	NHCOCHEGHCO2H 49689
	HC <sub>2</sub> O N N NH <sub>2</sub> SCH O A9713		49789
СН <sub>2</sub> СН(ОН)СООН 49848		CH(SO <sub>3</sub> H)CH <sub>2</sub> CH	
	$\bigcup_{i=1}^{n_{i}} \bigcup_{j=1}^{n_{i}} \bigcup_{j=1}^{n_{i}} \sum_{j=1}^{n_{i}} \sum_{j=1}^{n_{$	$H_{3C} \xrightarrow{N} H_{3} \xrightarrow{H} N_{1} \xrightarrow{NH_2} $ $H_{3C} \xrightarrow{V} 50570$	CH <sub>2</sub> CH <sub>2</sub> NHCH <sub>2</sub> COOH 50603
	$ \begin{array}{c} \overset{CH_{2}CO_{2}}{\bigvee} & & \\ & & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & & \\ & & & \\ & & & \\ & & & \\ & & & & \\ & & & & \\ & & & & \\ & &$	CMe(OH)CH <sub>2</sub> NHSO <sub>2</sub>	н <sub>2</sub> с соон нус 51132

СООН СН3 СООН 51135	CHI CHI	$HO \xrightarrow{F}_{O} \xrightarrow{F}_{F} \xrightarrow{F}_{F} \xrightarrow{F}_{F} \xrightarrow{O}_{OH}$ 51277	н s соон 51357
рания (соон на соон на соон он, 51470	00N C C C C C C C C C C C C C C C C C C	C(NH <sub>2</sub> )(CO <sub>2</sub> H)CHMe <sub>2</sub> 51810	51857
NH <sub>2</sub> H <sub>2</sub> N CH <sub>3</sub> 51867	Br Contraction of the second s	CI NICH <sub>2</sub> CH <sub>2</sub> NH <sub>2</sub> NHCH <sub>2</sub> CH <sub>2</sub> NH <sub>2</sub> 52075	Me_CHICH(CO_H)NH H N N N S2343
ни от не ни с с с с с с с с с с с с с с с с с с	52440	hyter N 52445	солнин <sub>2</sub> но н но н н о сн <sub>2</sub> он 52902
$\bigcup_{I \\ I \\$	52929	CONHNHCO N 52961	MeC CH2CH2NH2 OMe 53069
53346	S3396	$H_2C \xrightarrow{N} H_3 = Br$ $GH_3 = Br$ $53441$	Br 53454
CH <sub>3</sub> CH	HOLE HANDER	53935	CH CH <sub>5</sub> H CH <sub>5</sub> 54044
HOOC	$\overset{H_{5}C}{\underset{N}{}}\overset{CH_{5}}{\underset{N}{}}\overset{H_{4}C}{\underset{N}{}}\overset{N}{\underset{N}{}}\overset{CH_{5}}{\underset{N}{}}\overset{H_{5}C}{\underset{N}{}}\overset{N}{\underset{N}{}}\overset{CH_{5}}{\underset{N}{}}$		$\begin{array}{c} H_{3}C \\ H_{3}C \\ H_{3}C \\ \\ H_{2}C \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\$
54088 H2N 54297	54101 H_N CH <sub>b</sub> CH <sub>b</sub> 54340	<sup>тон</sup> 54251	<i>∽</i> 54278
54671	54672	$H_{3C}$ $O$ $H_{3C}$ $CH_{3}$	он сі 54854

H <sub>3</sub> C N (CH <sub>2</sub> ) <sub>3</sub> Br 54970	HO $HO$ $HO$ $HO$ $HO$ $HO$ $HO$ $HO$ $H$	S55462	a store a stor
55810	56227	H <sub>2</sub> N N <sub>3</sub>	Me <sub>2</sub> NH <sub>2</sub> CH <sub>2</sub> C <sub>y</sub> H Cl 56378
Contraction of the second seco		S6681	
	CI SCH(CO <sub>2</sub> H)S-CI	H <sub>2</sub> H <sub>0</sub> H <sub>2</sub> H <sub>2</sub> H <sub>2</sub> H <sub>2</sub> H <sub>2</sub> H <sub>2</sub> H <sub>2</sub> H <sub>2</sub>	
57345		$ \begin{array}{c} & & & \\ & & & & \\ & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & $	Me <sub>2</sub> NSO <sub>2</sub> N 57788
$HN \xrightarrow{H}_{NH_2} H \xrightarrow{COOH}_{COH_3} 57810$		$H_2N$ $H_2N$ $H_2N$ $H_2N$ $H_2$ $H_2N$ $H_2$	S7975
out ← ← ← ← ← ← ← ← ← ← ← ← ← ← ← ← ← ← ←	CONH COLH 58025	HO HIN 158184	58255
HO <sub>2</sub> CH <sub>2</sub> CO <sub>2</sub> H 58250	o <sub>2</sub> N CH <sub>CHCO2</sub> H NH <sub>b</sub> 58612	Me CI CI CI CI S8739	Me Me 59275
source for the source of the s	H <sub>2</sub> C H <sub>2</sub> C H <sub>3</sub> C H <sub>2</sub> C H <sub>3</sub> CH <sub>3</sub> CH <sub>2</sub> 59288	59349	59413
59486	59620	CH(MHMe)CH(NHMe) N 60036	
H <sub>3</sub> C-N/NH <sub>2</sub> 60215	С(0)0Рг соон 60379	Мс Л - ос(о)Сң(он) - 60600	H <sub>2</sub> N H <sub>2</sub> N CH <sub>2</sub> CH <sub>2</sub> CH <sub>2</sub> COOH 61297

61735	HC=CHCH2NHSO2 H N N 61747	H H H H H H H H H H H H H H H H H H H	Hand Hand Hand Hand Hand Hand Hand Hand
H H H H H H H H H H H H H H H H H H H	Me Ph Ph Ph Ph Ph CHCH <sub>2</sub> OH 61810	HO HO HO HO HO HO HO HO HO HO HO HO HO H	Constraints for the second sec
ост. П. Сн. Сн. Сн. Сн. Сн. Сн. Сн. Сн. Сн. Сн	$ \begin{array}{c} \begin{array}{c} \begin{array}{c} \begin{array}{c} \begin{array}{c} \end{array}\\ \end{array} \\ \end{array} \\  \end{array} \\  \end{array} \\  \end{array} \\  \end{array} \\  \end{array} \\  \end{array}$	Me NH Me NH Me CSH <sub>2</sub> COOH Me 62586	$\underset{Me}{\overset{0}{}_{\underset{Me}{}}} \underset{Me}{\overset{0}{}_{\underset{Me}{}}} \underset{Me}{\overset{Me}{\overset{0}}} \underset{Me}{\overset{Me}{\overset{Me}{}} \underset{Me}{\overset{Me}{\overset{Me}{}}} \underset{Me}{\overset{Me}{$
	HN CH <sub>2</sub> NMe <sub>2</sub> 62749	но странования и простория и прост Напри и простория	H <sub>2</sub> N S NH <sub>2</sub> . NH 62857
$\underset{\substack{H_{C} \\ H_{C} \\ $	$\begin{array}{c c} & & & NH_2 & & NH_2 \\ & & & Me & & N^* & & (CH_2)_3 & & N^* & & Me \\ & & & & Me & & \\ & & & & Me & & \\ & & & & & 63052 \end{array}$		СО <sub>2</sub> H NH <sub>2</sub> NHCHOH 63331
HOW HIGH, CD)OMe	C NNHCSNH <sub>2</sub> 63786	63875	$\begin{array}{cccc} Ph & Ph \\ & &   \\ Ph & P^{*} - C & -C & -P^{*} - Ph \\ & &   \\ Ph & Ph \\ Ph & Ph \\ \end{array}$ $\begin{array}{c} 64111 \\ \end{array}$
	HC=NNHCOCH <sub>2</sub> -N <sup>*</sup> -Me Me 64408	HC=NNHCOCH <sub>2</sub> -N <sup>Me</sup> Me Me 64617	Me NN=H NN=H HOOC 64678
	t-Bu-(CH <sub>2</sub> ) <sub>3</sub> NMe <sub>2</sub>	a + b + b + b + b + b + b + b + b + b +	CH <sub>3</sub> Pri H <sub>3</sub> C CH <sub>2</sub> NH <sub>2</sub> 65238
	CH <sub>2</sub> NH(CH <sub>2</sub> ) <sub>3</sub> OH	CH <sub>2</sub> NH(CH <sub>2</sub> ) <sub>3</sub> OH	Me H H H H H H H H H H H H H H H H H H H
65537 H <sub>1</sub> ,с, , , , , , , , , , , , , , , , , , ,	65650 о <sub>2</sub> м, , , , , , , , , , , , , , , , , , ,	65652	65683
50 <sub>3</sub> H H <sub>2</sub> NS0 <sub>2</sub> N 65990	N CH <sub>2</sub>	CH2NHCH2CH2·N NH	СH <sub>2</sub> NHOH <sub>2</sub> CH <sub>2</sub> -N NH 66184

CH2NHCH2CH2NN	COOH COOH		NH OSO2-
66185	соон соон 66207	<sup>но́ ј</sup> н 66380	66426
	NHCOCH <sub>2</sub> NH <sub>2</sub> COOH	$ \underset{Br}{\overset{OH}{}}  \underset{Br}{\overset{Ho}{}}  \underset{Br}{\overset{Ho}{\overset{Ho}{}}  \underset{Br}{\overset{Ho}{\overset{Ho}{}}  \underset{Br}{\overset{Ho}{\overset{H}{H$	H <sub>3</sub> C, N, H, NH <sub>2</sub> 0, NH 67323
HO HO G7485	NNHC(NH <sub>2</sub> )=NH HO OH 67608		Me Me Me 67852
Me Me H Me Me 67856	HO HO HO HO HO HO HO HO HO HO HO HO HO H	$\begin{array}{c} & \overset{\text{Me}}{\underset{OH}{\longrightarrow}} \overset{OH}{\underset{OH}{\longrightarrow}} \overset{\text{NMe}_2}{\underset{OH}{\longrightarrow}} \overset{OH}{\underset{OH}{\longrightarrow}} \overset{OH}{\underset{OH}{\longrightarrow}} \overset{OH}{\underset{OH}{\longrightarrow}} \\ & 69343 \end{array}$	Aco 69540
69573			
HOCC ME 70194	N СН <sub>2</sub> СООН 70769	CH <sub>2</sub> M(OH)CH <sub>2</sub> CH <sub>2</sub>	CH <sub>2</sub> N(OH)CH <sub>2</sub> CH <sub>2</sub>
$() \qquad \qquad$		NH2 HEN N CF3	
71284	н 🍑 71426	CF3 71669	71881
		HN=C(H <sub>2</sub> NNHNHC NH	HOOC HOOC
и 72234	72254	72528	72533
ме-Л в соорональствон 72861	C(O)OMe Very o C(O)Me 72914	۲2938	72939
CONHCH <sub>2</sub> C(NH <sub>2</sub> NH H <sub>2</sub> N 72942	NH2 73013	но,с соон 73100	

	Ph_c_k_N_NCHs		
	Рһн <sub>2</sub> с 0 73254	73300	нс-ё-ч- б-чч
H <sub>3</sub> C <sup>H3</sup> NH NH	HOOC /hurn	HO CH_2NMe_2	
73646	HOUC 73712	73721	73748
OH 73827	F 73990	NO2N3 74390	CI 74420
NHCSNH C	NH NHCNHCN	CH <sub>2</sub> NHCH <sub>2</sub> CH <sub>2</sub> NH <sub>2</sub>	
74429	a 74463	74472	74702
ссю ин-ссо со осно			CH3 NNHC(NH2)=NH
74860	75140	75503	I сн <sub>з</sub> 75513
Me H H H T T T T T T T T T T T T T T T T	HO <sub>2</sub> CH <sub>2</sub> CH <sub>2</sub> CH <sub>2</sub> C	Ph COOH HO <sub>2</sub> C 75627	N N CHt.)xCOCH 75747
NH N H N (CH <sub>2</sub> ) ССH <sub>2</sub> ) ССН <sub>2</sub> ССОН Т5749		HOCH THE TRANSPORT	соон H <sub>b</sub> S Me COOH Коон Коон Коон
	C C C C C C C C C C C C C C C C C C C		CI NHCH2CH2OH
77037	77053	77393	NH2 77427
HOOH CH <sub>3</sub> 77552	0H H <sub>2</sub> N NH <sub>2</sub> 77554	HN NHC(0)0 H H T7597	но от снион но от снион нонистрон нонистрон 77963
	MezNH2CH2CHC 78206	но СНЕКМе <sub>2</sub> соон 10 78296	SMe N CH <sub>2b</sub> Me 78508
	CONHCHCOOH NH2 <sup>Me</sup>		
78674	78857	78864	ة T8871

		OSOMe	
79050	° 79103	79422	79529
		C(O)OCH <sub>2</sub> Ph NH <sub>2</sub> 80126	м сн <sub>2</sub> сн <sub>2</sub> соон 80177
	HOOC OH HOOC OH 81463	H <sub>2</sub> C <sub>N</sub> H <sub>1</sub> CH <sub>2</sub> h <sub>2</sub> H <sub>1</sub> CH <sub>2</sub> h <sub>2</sub> H <sub>1</sub> of H <sub>3</sub> B1509	
ноос-Ссоон 81620		кала кала кала кала кала кала кала кала	HOOC HINNER BELLEVILLE
82339			$H_{H_{2}} \xrightarrow{OH} \xrightarrow{OH} \xrightarrow{CH_{3}} \xrightarrow{I} \xrightarrow{I} \xrightarrow{OCH_{3}} \xrightarrow{I} \xrightarrow{OCH_{3}} \xrightarrow{I} \xrightarrow{OCH_{3}} \xrightarrow{I} \xrightarrow{OCH_{3}} \xrightarrow{I} \xrightarrow{OCH_{3}} \xrightarrow{OH} \xrightarrow{O} \xrightarrow{OH} \xrightarrow{OH} \xrightarrow{OH} \xrightarrow{OH} \xrightarrow{OH} \xrightarrow{OH} \xrightarrow{OH} \xrightarrow{O} \xrightarrow{OH} \xrightarrow{OH} O$
	H <sub>2</sub> NO <sub>2</sub> S H <sub>2</sub> NO <sub>2</sub> S H <sub>2</sub> NO <sub>2</sub> S H <sub>2</sub> NO <sub>2</sub> S H <sub>2</sub> H <sub>3</sub> NO <sub>2</sub> S	$H_{2N}$ $H_{N}$ $SH$ $SH$ $S3224$	
Ме Ме ме стъсн,ссон 83436		Me <sub>2</sub> HCH <sub>2</sub> CHCOCHN H <sub>2</sub> N 83633	0 N H 83960
ньс он в 4093	K K K K K K K K K K K K K K K K K K K	н <sub>2</sub> N N=N (ОН <sub>2</sub> O H <sub>2</sub> N NH <sub>2</sub> Aq(OH <sub>2</sub> O 84130	Стустуссон 84145
н <sub>з</sub> с s соон 84216	H HOOCH <sub>2</sub> C H 84221	Me, I, NN=CMeCOOH	As(OH)2=0 N3 84460
N <sub>3</sub>	HOOC N-S <sup>2</sup> HOOC 85010		с(CO <sub>2</sub> H)=C(OH)COOH i,Pr 85176
			CH <sub>2</sub> C(CO <sub>2</sub> H)=NOH 85359

C(CN)=C(OH)COOH	H <sub>2</sub> NH <sub>2</sub> C-COOH	CH(OH)CH <sub>2</sub> CH <sub>2</sub> NHPri	
85372	сн <sub>а</sub> 85420	85431	Br 85433
но осна-сна-ко 85459	$ \underbrace{ \begin{array}{c} & & \\ &$		SO <sub>2</sub> NHCH <sub>2</sub> CH <sub>2</sub> OH
Е1-О-С-S-СН <sub>2</sub> -С-ОН 85743		Me CH2COOH	CH <sub>2</sub> NHCH <sub>2</sub> 86140
HOOC 86153	ED H CHECH CHECH CHECH		Me-C-COOH HNH <sub>2</sub> C OMe 86380
ноосньоо	MezCH2NOCHN MezCH2NOCHN Me N Me Me Me Me Me Me Me Me Me Me	Me <sub>2</sub> C, N, CSNH Me <sub>2</sub> C, CH <sub>2</sub> OH 87008	
HOH <sub>2</sub> CH <sub>2</sub> C	HOOCH <sub>2</sub> CH <sub>2</sub> C-N-Me 87119	НСОСН <sub>2</sub> СН <sub>2</sub> СООН 87229	
CONHCH <sub>2</sub> COOH OH 87566	HOSE NON SOUTH 87877	Н <sub>3</sub> С ОН 87882	CH2NMeCH2CH2OH OH
	CH2CH(NH2)COOH		HOCH <sub>2</sub> —CHCONHCH <sub>2</sub> COOH   NH <sub>2</sub>
ме снуон он 88135	88416	88418	88482
г СН <sub>2</sub> соон 88616	(CH <sub>2</sub> ) <sub>3</sub> NMe <sub>2</sub> 88622	COM HE SERVICE	NHCO- 88795
	Ho Br	NH2 (CH2)1 H2N Me 88903	
	88999	(CH <sub>2)3</sub> -NH <sub>2</sub> 89110	HO35 HO35 HO35 HO35 HO35 HO35 HO35 HO35

(CH <sub>2</sub> ) <sub>3</sub> COOH	ОРО <sub>3</sub> Н <sub>2</sub>   МеНС— СНСО <sub>2</sub> Н	OCH2CH(OH)-CH2N	CH2CH(OH)-CH2O
NH <sub>2</sub> 89250	 NH₂ 89296	a 89419	89435
-0-C-C-N 89457	NH2 (CH2) H2N NH2 NH2 NH2 89818	°	90318
	СH <sub>2</sub> CH(NH <sub>2</sub> )СООН	CI Me CO-CH_2OPOJH2 Me H MICH H Me MiCH H Me MiCH	
90328	90367 Ì	90615 ÇH <sub>3</sub> ÇООН	90616
CH(OH)CH2		H <sub>3</sub> C N SH	H <sub>3</sub> C-
90630	90737	н <sub>3</sub> с Сн <sub>3</sub> 90798	90810
Соон		CH <sub>3</sub> CH <sub>3</sub> CH <sub>3</sub>	
90823	90829	90831	H <sub>3</sub> C OSO <sub>2</sub> CH <sub>3</sub> 90841
H <sub>3</sub> C H <sub>N</sub> C H <sub>N</sub> C S C H <sub>3</sub> C C H <sub>3</sub> C S C H <sub>3</sub> C C S C S C S C S C S C S C S C S C S C		0 0 91750	
	H <sub>3</sub> C NO <sub>2</sub>		
91884	92378	92412	92577
HOOC	HLND 25		°
92828	но он в 92893	Соон соон 92896	NH2 92987
N S S			H <sub>2</sub> N
92988	93033	но 93241	93277
93317		Phi and Phi an	HO CH <sub>3</sub> O CH <sub>3</sub> O CH <sub>3</sub> O CH <sub>3</sub> O O O O O O O O O O O O O O O O O O O

Fac. of Grad. Studies, Mahidol Univ.

Ph. D. (Pharmaceutical Chemistry and Phytochemistry) / 143

	HOOOH <sub>3</sub> 93807		HO HO SHOT
<sup>5</sup> <sup>5</sup> <sup>6</sup> <sup>6</sup> <sup>6</sup> <sup>6</sup> <sup>6</sup> <sup>6</sup> <sup>7</sup> <sup>7</sup> <sup>8</sup> <sup>94783</sup>	94810	94820	ноос соон 94914
о	$H_{H_{1}C_{n}} \rightarrow C^{CH_{5}}$	$(1) \qquad \qquad$	95490
95501	95503	95570	
Страни и простории и простор 95666	H <sub>3</sub> C CH <sub>3</sub> H <sub>3</sub> C 95676	H <sub>3</sub> C N H O N N CH <sub>5</sub> N SH 95865	H <sub>3</sub> C <sub>N</sub> N CH <sub>5</sub> 95910
HIG HIGH NO2 OF HIGH NO2 OF HIGH NO2 OF HIGH NO2	His	Aco PH3 CH3 CH3 CH3 CH3 CH3 CH3 CH3 CH3 CH3 C	H <sub>3</sub> C H <sub>3</sub> C
н <sub>5</sub> со <sup>NH2</sup> соон 96396	Сталини и стал И сталини и с	СН <sub>3</sub> 96606	NH <sub>2</sub> Соон 96678
н <sub>2</sub> N н <sub>2</sub> N н <sub>2</sub> N н <sub></sub>	96694	( ) = ( )	H <sub>3</sub> C N N CH <sub>3</sub> 97064
° + 0 + 49 <sup>12</sup> - N - CH <sub>2</sub> CO <sub>2</sub> . N - CH <sub>2</sub> CO <sub>2</sub> . N - CH <sub>2</sub> CO <sub>2</sub> . 97345	н <sub>я</sub> с <sub>н<sub>1</sub></sub> 97845	H <sub>3</sub> CO H <sub>3</sub> CO H <sub>3</sub> CO H <sub>3</sub> CO CH <sub>3</sub> CH <sub>3</sub> CH <sub>3</sub> CH <sub>3</sub> CH <sub>3</sub> CH <sub>3</sub>	H <sub>3</sub> C H <sub>3</sub> C
м соон H <sub>2</sub> N H CH <sub>3</sub> 97918	97923	н <sub>3</sub> с s соон 97927	

ОН 98019		H <sub>3</sub> C H <sub>3</sub> C CH <sub>3</sub> CH <sub>3</sub> CH <sub>3</sub> 98332	в, соон 98601
N S O CH3 98828		<i>C</i> →- <sup>8</sup> - <sup>H2</sup> -s- <i>C</i> →-S→98905	Соон 0 н <sub>3</sub> С сн <sub>3</sub> 99241
н <sub>3</sub> с , соон сн <sub>3</sub> , но 99419	HO OH OF NH2 OF N NH2 HO 99445	H <sub>3</sub> C H <sub>3</sub> C H <sub>3</sub> C H <sub>3</sub> H <sub>3</sub> C H <sub>3</sub> 99504	99515
99526	99543	99550	$\underset{CH_{5}}{\overset{S}{\underset{H_{5}}{}}} \underset{CH_{5}}{\overset{H}{\underset{H_{5}}{}}} \underset{N}{\overset{S}{\underset{H_{5}}{}}} \underset{N}{\overset{N}{\underset{H_{5}}{}}} \underset{N}{\overset{N}{\underset{H_{5}}{}} \underset{N}{}} \underset{N}{\overset{N}{}} \underset{N}{}} \underset{N}{\overset{N}{}} \underset{N}{}} \underset{N}{}} \underset{N}{}} \underset{N}{} \underset{N}{}} \underset{N}{} \underset{N}{} \underset{N}{}} \underset{N}{}} \underset{N}{} \underset{N}{}} \underset{N}{} \underset{N}{} \underset{N}{}} \underset{N}{} \underset{N}{} \underset{N}{}} \underset{N}{}} \underset{N}{} \underset{N}{} \underset{N}{}} \underset{N}{} \underset{N}{} \underset{N}{}} \underset{N}{} \underset{N}{}} \underset{N}{} \underset{N}{} \underset{N}{}} \underset{N}{} \underset{N}{} \underset{N}{}} \underset{N}{} \underset{N}{} \underset{N}{} \underset{N}{} \underset{N}{}} \underset{N}{} \underset{N}{} \underset{N}{} \underset{N}{} \underset{N}{}} \underset{N}{} $
H <sub>3</sub> C H S 99663	H <sub>3</sub> C <sub>N</sub> s <sub>CH<sub>3</sub></sub> H <sub>3</sub> C <sub>N</sub> S H <sub>3</sub> S 99667	Huch N H S N	H <sub>3</sub> C <sub>N</sub> H <sub>3</sub> C <sub>N</sub> H <sub>3</sub> H <sub>3</sub> H <sub>3</sub> 99676
HOOC 99785	но он но он с он он он 99799	He H	$H_2N \xrightarrow{N} H_2 = N \xrightarrow{O} H_3$
$\underset{(CH_2)_2NMe_2}{\overset{H_3C}{\underset{(CH_2)_2NMe_2}{}}}$ 100297	FaC OH CHE MMe2 100708	H <sub>a</sub> C H <sub>a</sub> C H <sub>a</sub> C H <sub>a</sub> C	CH3 100762
(CH <sub>2</sub> ) <sub>3</sub> NH <sub>2</sub> NH NH 100857	HO H COOH HO H HAN HAN COOH OH 100858	С С С С С С С С С С С С С С С С С С С	H <sub>3</sub> CO O <sub>2</sub> N COCH <sub>3</sub> COOH 100936
ACHN 100942	СH <sub>3</sub> СH <sub>3</sub> СH <sub>3</sub> 100972		Соон Соон 101127
	$\stackrel{OH}{\xrightarrow{H}}_{H_3}$	101751	$\overset{H_2N}{\underset{NH_2}{\overset{N}{}}} \overset{N}{\underset{NH_2}{}} \overset{(CH_2)_3}{\underset{NH_2}{\overset{N}{}}} \overset{N}{\underset{NH_2}{\overset{NH_2}{}}} \overset{NH_2}{\underset{101824}{\overset{NH_2}{}}}$

$\overset{H_2N}{\underset{NH_2}{\longrightarrow}}\overset{N}{\underset{H_2}{\longrightarrow}}\overset{(CH_2)_k}{\underset{NH_2}{\longrightarrow}}\overset{N}{\underset{H_2}{\longrightarrow}}\overset{NH_2}{\underset{H_2}{\longrightarrow}}$	H <sub>2</sub> N N COOH N N N COOH NH <sub>2</sub> 101828	N 5 СООН NO2 101844	Me(ON)N-C-E-E-N 101984
102728		H <sub>3</sub> C <sub>4</sub> H <sub>3</sub> C <sub>4</sub> CH <sub>2</sub> CH(OH)Me Me(HO)HCH <sub>2</sub> C 103019	HOOC N=N N-S 103773
$\underset{CH_{5}}{\overset{(OOH)}{\underset{CH_{5}}{\overset{(OOH)}{\underset{H_{5}}{\underset{H_{5}}{\overset{(OOH)}{\underset{H_{5}}{\underset{H_{5}}{\overset{(OOH)}{\underset{H_{5}}{H_{5}}{H_{5}}{H_{5}}{H_{5}}{H_{5}}{H_{5}}{H_{5}}{H_{1}{H_{1}}{$	H <sub>3</sub> C N H <sub>1</sub> C OH	$(f_{i}) = (f_{i}) = (f_{$	Страния страна с
HOOC H H H H H H H H H H H H H H H H		но ОН 105534	
105597			Соон
	$ \underset{\substack{\mu \in \mathcal{H}_{h} \\ 0 \\ 0 \\ 0 \\ 0 \\ 0 \\ 0 \\ 0 \\ 0 \\ 0 \\ $	HOOC H H H H H H H H H H H H H H H H H H	
NHCH <sub>2</sub> CH=CMe <sub>2</sub> N N N N N N N N N N N N N N N N N N N			
	$H_{3C}$ H	о сн <sub>3</sub> но сн <sub>3</sub> 108225	
		HO NH <sub>2</sub> HO NH <sub>2</sub> 108731	Соон СН3 108802
ноос 108895	108944	сн <sub>3</sub> сн <sub>3</sub> 109128	
H <sub>3</sub> C H <sub>3</sub> C H <sub>3</sub> C N 109174	$N \rightarrow N \rightarrow$	H <sub>3</sub> C H <sub>3</sub> C H <sub>3</sub> C H <sub>3</sub> C H <sub>3</sub> C CH <sub>3</sub> CH <sub>3</sub> C	

	$CI \qquad \qquad$	H <sub>9</sub> CO H	H <sub>G</sub> Ch H <sub>G</sub> H
		AcHN, , , , , , , , , , , , , , , , , , ,	
Hall		$\overbrace{CH_{5}}^{N} \xrightarrow{NH_{2}} 110275$	
$\underset{n,c}{\overset{n,c}{\longrightarrow}},\underset{n,c}{\overset{n,c}{\to}},\underset{n,c}{\overset{n,c}{\to}},\underset{n,c}{\overset{n,c}{\to}},\underset{n,c}{\overset{n,c}{\to}},\underset{n,c}{\overset{n,c}{\to}},\underset{n,c}{\overset{n,c}{\overset{n,c}{\to}},\underset{n,c}{\overset{n,c}{\to}},\underset{n,c}{\overset{n,c}{\to}},\underset{n,c}{\overset{n,c}{\to}},\underset{n,c}{\overset{n,c}{\to}},\underset{n,c}{\overset{n,c}{\to}},\underset{n,c}{\overset{n,c}{\to}},\underset{n,c}{\overset{n,c}{\to}},\underset{n,c}{\overset{n,c}{\to}},\underset{n,c}{\overset{n,c}{\to}},\underset{n,c}{\overset{n,c}{\to}},\underset{n,c}{\overset{n,c}{\to}},\underset{n,c}{\overset{n,c}{\to}},\underset{n,c}{\overset{n,c}{\to}},\underset{n,c}{\overset{n,c}{\to}},\underset{n,c}{\overset{n,c}{\to}},\underset{n,c}{\overset{n,c}{\to}},\underset{n,c}{\overset{n,c}{\to}},\underset{n,c}{\overset{n,c}{\to}},\underset{n,c}{\overset{n,c}{\to}},n$		$CH_3 \\ Br \\ CH_3 \\ CH_3 $ 112200	
	() + () + () + () + () + () + () + () +	одс 112671	он Сосн 112675
соон 112702	но СНЬ ОН ОН ОН 112737		112778
	H <sub>2</sub> N S N S N S N S N S N S N S N S N S N S	П 112796	AC CH <sub>3</sub> 112898
ноос 112983	$ \overset{\circ}{\rightarrow} $	$\overset{N_{2}}{\underset{A}{\overset{N_{2}}{\overset{C}{\underset{A}{\atop\\{A}}{\overset{C}{\underset{A}{\overset{C}{\underset{A}{\atop\\{A}}{\overset{C}{\underset{A}{\atop\\{A}}{\atop\\{A}}{\underset{C}{\atop\\{A}}{\atop\\{A}}{\atop\\{A}}{\atop\\{A}}{\atop\\{A}}{\atop\\{A}}{{\\$	HN HN N HN 113989
Соон 113997	СН3 СООН 114213	обе соон 114215	$\overset{H_{3}C}{\underset{N}{\overset{S}{\underset{O}{\overset{S}{\underset{O}{\overset{N}{\underset{O}{\underset{O}{\overset{N}{\underset{O}{\overset{N}{\underset{O}{\overset{N}{\underset{O}{\overset{N}{\underset{O}{\overset{N}{\underset{O}{\underset{O}{\overset{N}{\underset{O}{\underset{O}{\overset{N}{\underset{O}{\underset{O}{\underset{O}{\overset{N}{\underset{O}{\underset{O}{\underset{O}{\underset{O}{\underset{O}{\underset{O}{\underset{O}{\underset$
$H_{AN} \xrightarrow{Br} G_{O} \xrightarrow{H_{b}N} H_{$	но	$\overset{H_{\mathcal{G}}}{\underset{OH_{3}}{\overset{H_{1}}}}}}}}}}}}}}}}}}}}}}}}}}}}}}}}}}}$	$(\mathbf{r}_{t_{1}})$

	$H_{3}C \xrightarrow{NH_{2}} N \xrightarrow{N}_{N} O \xrightarrow{S} NH_{2}$	0 <sub>2</sub> N S C C 114528	How of the Horizon Hor
HO HO 114609			о м н м н м н м н м н м н 114766
HOOC HI4/92	114831 Соорнания 115787	114885 NH2 HEN N N COOH 115883	115448 но н <sub>2</sub> NH <sub>2</sub> 116064
$H_2N H_2 H_2N H_2$	C <sup>145</sup> C <sup></sup>		U U U U U U U U U U U U U U U U U U U
(NH) N) H 116533			
116709	H <sub>2</sub> N , NO <sub>2</sub> CH <sub>3</sub> 116805	$\alpha \xrightarrow{CH_3} CH_3 \xrightarrow{CH_3} CH_3 \xrightarrow{CH_3} CH_3 \xrightarrow{CH_3} CH_3 \xrightarrow{CH_3} 116933$	NH2 H3C N 116977
	$\overset{HO}{\xrightarrow{}}_{H,N\\ O\\ O\\$		
	H <sub>b</sub> N <sub>y</sub>	$\overset{HeH}{\underset{Met}{\overset{H}}} \xrightarrow{\overset{H}{\underset{Met}{\overset{H}}}} \xrightarrow{\overset{H}{\underset{Met}{\overset{H}}}} \xrightarrow{\overset{H}{\underset{Met}{\overset{H}}}} \overset{H}{\underset{Met}{\overset{H}}} \xrightarrow{\overset{H}{\underset{Met}{\overset{H}}}} 117269$	

	$\underset{H_2N}{\overset{NH_2}{\underset{H_2N}{\overset{N}{\underset{N}{\underset{N}{\underset{N}{\underset{N}{\underset{N}{\underset{N}{$	он соон н <sub>2</sub> N NN NH <sub>2</sub> 117285	117369
	н соон 117429	Hooc 	Соон H <sub>2</sub> N Соон 117441
Соон нN 117462	COOH HN 117470	ноос ноос соон <sup>о</sup> 117489	117554
$ \underbrace{ \left( \begin{array}{c} \begin{array}{c} \\ \\ \end{array} \right) \\ + \end{array} \right) \\ + \\ + \\ \end{array} \right) \\ + \\ H \\ H$	$\underset{h \in \mathbb{N}}{\overset{NH_2}{\underset{h \leftarrow 0}{\overset{r}{\underset{h \leftarrow 0}{\underset{h \atop1}{\underset{h \leftarrow 0}{\underset{h \leftarrow 0}{\underset{h \leftarrow 0}{\underset{h \leftarrow 0}{\underset{h \leftarrow 0}{\underset{h \atop1}{\underset{h \leftarrow 0}{\underset{h \atop1}{\underset{h \atop1}{\underset{h \leftarrow 0}{\underset{h \atop1}{\underset{h \atop1}{\underset{h \atop1}{\underset{h \atop1}{\underset{h \atop1}{\atop1}{\underset{h \atop1}{\underset{h \atop1}{\underset{h \atop1}{\atop1}{\underset{h \atop1}{\atop1}{\atop1}{\underset{h \atop1}{\atop1}{\atop1}{\atop1}{\atop1}{\atop1}{\atop1}{\atop1}{\atop1}{\atop1}{$	117908	
н <sub>2</sub> N OH ОН 118217	HOOC NH <sub>2</sub> HOOC 118360	$ \bigcup_{HN}^{O} \bigcup_{NH}^{N} \bigcup_{NH_2}^{NH_2} 118396 $	
$\overset{H_{3}C}{\underset{OH}{\leftarrow}} \overset{O}{\underset{OH}{\leftarrow}} \overset{O}{\underset{COOH}{\leftarrow}} \overset{O}{\underset{COOH}{\leftarrow}} $		H <sub>3</sub> C-N + C H <sub>3</sub> C-N + C H <sub>3</sub> C-N + C H <sub>3</sub> C-N + C H <sub>3</sub> C	
СООН СН <sub>3</sub> 119531	н <sub>з</sub> с о он он 119621	$H_{3}C_{N}$ $H_{4}N$ $H_{2}N$ $H_{2}N$ $H_{2}N$ $H_{2}N$ $H_{2}N$ $H_{3}N$ $H_{4}N$	$ \overset{\text{HO}}{\underset{\substack{\leftarrow}\\\leftarrow}{\overset{\leftarrow}\\\leftarrow}} \overset{\text{OH}}{\underset{\leftarrow}{\overset{\leftarrow}\\\leftarrow}} \overset{\text{OH}}{\underset{\leftarrow}{\overset{\leftarrow}\\\leftarrow}} $ 119886
Br + C + C + C + C + C + C + C + C + C +		ностория и проведения и проведе	
	$H_3C$	ССООН 120025	

N S S N COOH HOOC 120288	остосносности 120495	Соон 120501	
HOOC HOOC HOOC HOOC HOOC HOOC HOOC HOOC	N H H H H H H H H H H H H H H H H H H H		
		СН3 ОН 120917	$\overset{\text{cH}_{3}}{\underset{s}{\overset{\text{cH}_{3}}}{\overset{\text{cH}_{3}}}{\overset{\text{cH}_{3}}}{\overset{\text{cH}_{3}}}{\overset{\text{cH}_{3}}}{\overset{\text{cH}_{3}}}{\overset{\text{cH}_{3}}}{\overset{\text{cH}_{3}}}{\overset{\text{cH}_{3}}}{\overset{sH}_{3}}}{\overset{sH}_{3}}}{\overset{sH}_{3}}}{\overset{sH}_{3}}}{\overset{sH}_{3}}}{\overset{sH}_{3}}}{\overset{sH}_{3}}}{\overset{sH}_{3}}}{\overset{sH}_{3}}}{\overset{sH}_{3}}}{\overset{sH}_{3}}}{\overset{sH}_{3}}}{\overset{sH}_{3}}}{\overset{sH}_{3}}}{\overset{sH}_{3}}}{\overset{sH}_{3}}}{\overset{sH}_{3}}}{\overset{sH}_{3}}}{\overset{sH}_{3}}}}}}}}}}}}}}}}}}}}}}}}}}$
H <sub>5</sub> Ch H <sub>5</sub> CH <sub>2</sub> H <sub>5</sub> CH <sub>2</sub> CH <sub>5</sub> CH <sub>2</sub> CH <sub>5</sub> CH <sub>2</sub> OH 121145	$ \underset{H_{2}N}{\overset{NH_{2}}{\underset{H_{2}N}{\overset{N}{\underset{H_{2}N}{\overset{P}{\underset{H_{2}N}{\underset{H_{2}N}{\overset{P}{\underset{H_{2}N}{\underset{H_{2}N}{\underset{H_{2}N}{\underset{H_{2}N}{\underset{H_{2}N}{\underset{H_{1}N}{I}N}{\underset{H_{1}N}{I}N}{I}}}}}}}}}}}}}}}}}}}}}}}}}}}}$	121304	Н300 Н 121384
HN N N SO <sub>3</sub> H 121603	N N 121771	H <sub>2</sub> C <sub>H<sub>3</sub></sub> CH <sub>3</sub> CH <sub>3</sub> CH <sub>3</sub>	
орон	но но но но но но но но но но но но но н	H <sub>9</sub> C <sup>O</sup> + H <sub>9</sub> C <sup>O</sup> + H <sub>9</sub> C <sup>O</sup> + H <sub>9</sub> C <sup>O</sup> + H <sub>9</sub> + H <sub>9</sub> C <sup>O</sup> + H <sub>9</sub>	
	$H_{3}C_{0}$	CN CH3 CH3 CH3 121912	
он соон 122281	122335		H <sub>2</sub> C + S + CH <sub>3</sub> H <sub>2</sub> C + S + CH <sub>3</sub> H <sub>2</sub> C + C + S + CH <sub>3</sub> N + C + C + S + C + S + C + S + C + S + C + S + C + S + C + S + C + S + S
с	н <sub>2</sub> N соон 5 122405	ноос	
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но	но строборание с с с с с с с с с с с с с с с с с с с		
$ \underset{\alpha}{\overset{(\alpha)}{\underset{\alpha}{\overset{(\alpha)}}}}{\overset{(\alpha)}}{\overset{(\alpha)}{\overset{(\alpha)}{\overset{(\alpha)}{\overset{(\alpha}{\overset{(\alpha)}{\overset{(\alpha)}{$	н 130830	H <sub>3</sub> C, H <sub></sub>	
	CH <sub>3</sub> H <sub>3</sub> C <sup>-N</sup> CH <sub>3</sub> N CH <sub>3</sub> CH <sub>3</sub> 130915		HO <sub>2</sub> C HO <sub>2</sub> C CO <sub>2</sub> H 131453
HO $H_{H_{3}}$ $H_{3}$		HO <sub>2</sub> C N N 131614	NO <sub>2</sub> H O <sub>2</sub> N CO <sub>2</sub> H CO <sub>2</sub> H 131615
131734	C CH <sub>3</sub> I 31815		$HO_2C \xrightarrow{0} VH_2$
"HERE CONTROL OF THE SECOND SE		ССС <sup>5</sup> 25 ССС <sup>5</sup> 132693	
133074	$ \underset{\substack{\text{HOR} \\ \text{HOC} \\ HOC$		
		HO PH H2 PH C H2N CH2 H2N 133811	отрон он 133837
СО <sub>2</sub> Н НО НН2 133887		$\overbrace{}^{\circ} \overbrace{}^{\circ} \overbrace{}^{\circ} \underset{CH_{3}}{\overset{\circ}} \overbrace{}^{\circ} \overbrace{}^{\circ} \underset{}{\overset{\circ}} \overbrace{}^{\circ} \underset{}{\overset{}} \underset{134120}{\overset{}}$	
		ны – – – – – – – – – – – – – – – – – – –	н_м

	HO <sub>2</sub> C N H	H500 HC L L L L L L L L L L L L L L L L L L	H <sub>3</sub> C O
со <sub>2</sub> н 134244	134478	[ N 134664	CN 134666
$H_{3}CO \xrightarrow{(CN)} H_{2}N \xrightarrow{(CN)} H_{3}CO \xrightarrow{(CN)} H_{2}N \xrightarrow{(CN)} H_$		H <sub>9</sub> 00 H <sub>9</sub> 00 H <sub>3</sub> C 134755	135371
$\begin{array}{c} \begin{array}{c} & & & \\ & & $		$H_{bC} \xrightarrow{CH_{3}} H_{bC} \xrightarrow{F}$ $H_{bC} \xrightarrow{H_{bC}} H_{bC} \xrightarrow{F}$ $H_{bC} \xrightarrow$	135900
HN2 CO2H		NHAC	MeO OMe
136035	<sup>со</sup> лн 136469	<sup>со</sup> ₂н 136472	<sup>со</sup> 2 <sup>н</sup> 136473
		HN N O HE 136727	осн <sub>2</sub> сн <sub>2</sub> он <sub>002</sub> н
N		$\sim$	NH2
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$ \begin{array}{c} \begin{array}{c} \begin{array}{c} \begin{array}{c} \begin{array}{c} \\ \end{array} \\ \end{array} \\ \end{array} \\ \end{array} \\ \begin{array}{c} \end{array} \\ \begin{array}{c} \end{array} \\ \begin{array}{c} \end{array} \\ \end{array} \\ \begin{array}{c} \end{array} \\ \begin{array}{c} \end{array} \\ \begin{array}{c} \end{array} \\ \end{array} \\ \begin{array}{c} \end{array} \\ \end{array} \\ \begin{array}{c} \end{array} \\ \end{array} \\ \begin{array}{c} \end{array} \\ \end{array} \\ \begin{array}{c} \end{array} \\ \end{array} \\ \end{array} \\ \end{array} \\ \begin{array}{c} \end{array} \\ \end{array} \\ \end{array} \\ \end{array} \\ \end{array} \\ \begin{array}{c} \end{array} \\ \end{array} $	$ \begin{array}{c}                                     $	(f) = (f)	$ \begin{array}{c} \begin{array}{c} \begin{array}{c} \begin{array}{c} \begin{array}{c} \begin{array}{c} \end{array}\\ \end{array}\\ \end{array} \\ \end{array} \\ \begin{array}{c} \end{array} \\ \end{array} \\ \end{array} \\ \begin{array}{c} \end{array} \\ \end{array} \\ \end{array} \\ \end{array} \\ \begin{array}{c} \end{array} \\ \end{array} \\ \end{array} \\ \end{array} $ \left( \begin{array}{c} \end{array} \\ \end{array} \\ \end{array} \\ \end{array} \\ \end{array} \\ \left( \begin{array}{c} \end{array} \\ \end{array} \\ \end{array} \\ \end{array} \\ \left( \begin{array}{c} \end{array} \\ \end{array} \\ \end{array} \\ \end{array} \\ \left( \begin{array}{c} \end{array} \\ \end{array} \\ \end{array} \\ \end{array} \\ \left( \begin{array}{c} \end{array} \\ \end{array} \\ \end{array} \\ \left( \begin{array}{c} \end{array} \\ \end{array} \\ \end{array} \\ \end{array}  \left( \begin{array}{c} \end{array} \\ \end{array} \\ \end{array} \\ \end{array}  \left( \begin{array}{c} \end{array} \\ \end{array} \\ \end{array} \\ \left( \begin{array}{c} \end{array} \\ \end{array} \\ \end{array}  \left( \begin{array}{c} \end{array} \\ \end{array} \\ \end{array}  \left( \begin{array}{c} \end{array} \\ \end{array} \\ \end{array}  \left( \begin{array}{c} \end{array} \\ \end{array}  \left( \end{array} \\ \bigg) \\ \left( \end{array} \\ \end{array}  \left( \end{array} \\ \bigg) \\ \left( \end{array} \\ \left) \\ \left( \end{array} \\ \bigg) \\ \left( \end{array} \\ \left( \end{array} \\ \bigg) \\ \left( \end{array} \\ \bigg) \\ \left( \end{array} \\ \left( \end{array} \\ \bigg) \\ \left( \end{array} \\ \left( \end{array} \\ \bigg) \\ \left( \end{array} \\ \bigg) \\ \left( \end{array} \\ \left( \end{array} \\ \left) \\ \left( \end{array} \\ \left
$(\downarrow)_{C,C} \downarrow_{C} $	$\downarrow \downarrow_{H_2}$ $\downarrow \downarrow_{H_3}$ $\downarrow \downarrow_{H_3}$ $\downarrow \downarrow \downarrow_{H_3}$ $\downarrow \downarrow $	(f) = (f)	$ \begin{array}{c} \underset{h_{h_{S}C}}{\overset{O}{\underset{O}{\overset{O}{\underset{H_{S}C}}}}} \\ \underset{M_{F}}{\overset{O}{\underset{O}{\overset{O}{\underset{H_{S}C}}}}} \\ \underset{M_{F}}{\overset{O}{\underset{I}{\overset{O}{\underset{I}{\underset{S}{\underset{O}{\underset{S}{\underset{O}{\underset{S}{\underset{O}{\underset{S}{\underset{O}{\underset{I}}{\underset{I}{\underset{I}{\underset{I}}{\underset{I}{\underset{I}{\underset{I}{\underset{I}{\underset{I}{\underset{I}{\underset{I}{\underset{I}{\underset{I}{\underset{I}}{\underset{I}{\underset{I}{\underset{I}}{\underset{I}{\underset{I}{\underset{I}{\atopI}}{\underset{I}}}}}}}}}}}}}}}}}}}}}{} \\ \\ \\ \underset{I{1}}}{}  \overset{I}{\underset{I}{\underset{I}{\underset{I}{\underset{I}}{\underset{I}}}}}}}}}$

			C-H-H-S-ONE
143125	он мн <sub>2</sub> 143491	143677	143722
	HO <sub>2</sub> CH <sub>2</sub> CS		(CH <sub>2b</sub> -OHMeCO <sub>2</sub> H
HN H2N H2N H2N H2N H2N H2N H2N H2N H2N H	$ \underset{k \in \mathcal{O}^{H}}{\overset{H}{\underset{h}}} \underset{k \in \mathcal{O}^{H}}{\overset{H}{\underset{h}}} \underset{H}{\overset{H}{\underset{h}}} \underset{H}}{\overset{H}{\underset{H}}} \underset{H}{\overset{H}{\underset{H}}} \underset{H}}{\overset{H}{\underset{H}}} \underset{H}}{\overset{H}{\underset{H}}} \underset{H}}{\overset{H}{\underset{H}}} \underset{H}}{\overset{H}{\underset{H}}} \underset{H}}{\overset{H}{\underset{H}}} \underset{H}}{\overset{H}{\underset{H}}} \underset{H}}{\overset{H}{\underset{H}}} \underset{H}}{\overset{H}}} \underset{H}}{\overset{H}{\underset{H}}} \underset{H}}{\overset{H}}} \underset{H}}{\overset{H}} \underset{H}}{\overset{H}}} \underset{H}}{\overset{H}} \underset{H}}{\overset{H}}} \underset{H}}{\overset{H}}} \underset{H}}{\overset{H}}} \underset{H}}{\overset{H}} \underset{H}}{\overset{H}}} \underset{H}}{\overset{H}}} \underset{H}}{\overset{H}}} \underset{H}}{\overset{H}}} $	$H_{C,NH_{2}C} \xrightarrow{H_{0}} H_{$	CH2-SC(NH2)=NH
ноос		Me	CH_CCO_H Me Me MeO-CI
146443	146878	147109	<sup>8</sup> 147737
CH <sub>2</sub> (CH <sub>2</sub> ) <sub>2</sub> NMe <sub>2</sub> CI 147744	но <sub>2</sub> с	нс соон 148230	С С С С С С С С С С С С С С С С С С С
	Me CH=CHOMe=CHCO <sub>2</sub> H OH Me 149707	$\overbrace{}^{Me} \overbrace{}^{OH} \phantom{aaaaaaaaaaaaaaaaaaaaaaaaaaaaaaaaaaa$	
	Соон 150277	of the second se	
$H_{H_{0}}^{0} \xrightarrow{CH} N_{H_{0}}^{0} \xrightarrow{H_{0}} H_{H_{0}}^{0} \xrightarrow{P_{H_{0}}} H_{H_{0}}^{0} \xrightarrow{P_{H_{0}}} H_{0}^{0} P_{$	ст нон 	H S S 150554	$\overset{H_{0}}{\rightarrow} \overset{H_{0}}{\rightarrow} H_$
° √ + − ↓ ↓ ↓ ↓ ↓ ↓ ↓ ↓ ↓ ↓ ↓ ↓ ↓ ↓ ↓ ↓ ↓ ↓	но Н Он 151063	HO $\rightarrow$	$\xrightarrow{o}_{HO} \xrightarrow{v}_{o} \xrightarrow{u}_{HO} \xrightarrow{v}_{HO} \xrightarrow{v}_$

H <sub>2</sub> C <sup>Q</sup> H <sub>2</sub> C <sub>Q</sub>	С С С С С С С С С С С С С С С С С С С		NH2 HEC N
<sup>сн</sup> <sub>э</sub> 152686	153308	s Цана 153330	H <sub>G</sub> C 0 153533
	$H_{O} \xrightarrow{QH} N_{O} \xrightarrow{NH_2} N_{CH_5}$ $H_{O} \xrightarrow{W} O \xrightarrow{N} N_{CH_5}$ $154020$	ت درب 154389	но Н с 154572
он 154652	HO-NH HO-NH HO-NH HO HO HO HO HO HO HO HO HO HO HO HO HO	H <sub>2</sub> N O H54966	CH <sub>3</sub> S 154983
но	N + + + + + + + + + + + + + + + + + + +	$ \underset{HN}{\overset{HN}{\longrightarrow}} \overset{H}{\overset{O}{\longrightarrow}} \overset{O}{\overset{O}{\longrightarrow}} $	$ \begin{array}{c} & & & \\ & & & \\ & & & \\ & & $
	но с булине 157421	$\underset{\substack{H_{0}C\\H_{$	H5N K K K K K K K K K K K K K K K K K K K
		HOS 158362	
но н	$ (f_{i}) = (f_{i}) + (f_$	$H_{H_{0}} \xrightarrow{H_{0}} H_{0} \xrightarrow{H_{0}} H_{0} \xrightarrow{H_{1}} $	$\downarrow_{n_{0}}^{C^{H}} \downarrow_{n_{0}}^{C^{H}} \downarrow_{n_{0}}^$
	$\downarrow \downarrow $	F F HO H <sub>3</sub> C I62215	NH <sub>2</sub> HN/H H
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H3C N			но
н <sub>эс</sub> 163160	163173	163326	о но 163339
	ну со 163376	N 163398	С,
H <sub>3</sub> C, , , , , , , , , , , , , , , , , , ,	он 163865		но Н Н Н Н Н Н Н Н Н Н Н Н Н Н Н Н Н Н Н
			$H_2N \xrightarrow{N}_{N+2} H_2C \xrightarrow{\alpha}_{N+2} 0$
но 164892			
	$H_{1}C_{1} \xrightarrow{0} (1 - 1) (1 -$	<del>с с с с с с с с с с с с с с с с с с с </del>	но в 167452
	С <sup>45</sup> с с с с с с с с с с с с с с с с с с с		Hec
CH <sub>3</sub> CH <sub>3</sub> CH <sub>3</sub> CH <sub>3</sub> CH <sub>3</sub> CH <sub>3</sub> CH <sub>3</sub> CH <sub>3</sub>	$\begin{array}{c} \overbrace{H_{3}C}^{s} \overbrace{H_{3}C}^{s} \overbrace{H_{3}}^{s} \\ H_{3} \overbrace{H_{3}}^{s} \overbrace{H_{3}}^{s} \end{array} \\ 169471 \end{array}$	$ \underset{\substack{H \subseteq H \\ H \subseteq H \\ H \subseteq H_{2}}{\overset{H}{\longrightarrow}} \underbrace{ \begin{array}{c} G \\ G $	C <sup>Hb</sup> , , , , , , , , , , , , , , , , , , ,

H <sub>3</sub> C H <sub>3</sub> C H <sub>3</sub> C H <sub>3</sub> C H <sub>4</sub> C H <sub>3</sub> C H <sub>4</sub> C H <sub>3</sub> C H <sub>4</sub> C	$H_{3}C \rightarrow 0 \qquad OH \qquad CH_{3} \rightarrow 0 \qquad H \qquad CH_$	$ \begin{array}{c} \overset{O}{\underset{H_{3}C}{\leftarrow}} \overset{O}{\underset{H_{3}C}{\underset{H_{3}C}{\leftarrow}} \overset{O}{\underset{H_{3}C}{\leftarrow}} \overset{O}{\underset{H_{3}C}$	$H_{3C} \xrightarrow{P_{0}} H_{3C} \xrightarrow{P_{0}} H_{1} \xrightarrow{P_{0}^{e^{2}}} H_{3C} \xrightarrow{P_{0}^{e^{2}}} H_{3C} \xrightarrow{P_{0}^{e^{2}}} H_{3} P$
$\begin{array}{c} H_{3C} & CH_{6} & H_{3}C & CH_{3} \\ \hline 0 & N & OHHO - N & CH_{2} \\ HO - N & CH_{2}^{2} & 0 - N & CH_{2}^{2} \\ HO - N & CH_{3} & H_{3}C & CH_{6} \\ H_{3}C & CH_{3} & H_{3}C & CH_{6} \\ \hline 169942 \end{array}$	$\begin{array}{c} H_{3}C & CH_{5} \\ 0 \\ -N \\ H_{5}C \\ -O \\ H_{5}C \\ -O \\ -N \\ H_{5}C \\ -O \\ -N \\ -O \\ -O \\ -N \\ -O \\ -O \\ -N \\ -O \\ -O$	$\begin{array}{c} \overset{H_{3}}{\underset{O}{\overset{H}{\underset{O}{\overset{OH}{\overset{OH}{\underset{O}{\overset{OH}{\underset{O}{\overset{OH}{\underset{O}{\overset{OH}{\overset{OH}{\underset{O}{\overset{OH}{\overset{OH}{\underset{O}{\overset{OH}{\overset{OH}{\underset{O}{\overset{OH}{\overset{OH}{\underset{O}{\overset{OH}{\overset{OH}{\overset{OH}{\overset{OH}{\underset{O}{\overset{OH}{OH}{\overset{OH}{OH}{\overset{OH}{O}{\overset{OH}{\overset{OH}{\overset{OH}{\overset{OH}{\overset{OH}{&}{\overset{OH}{\overset{OH}{\overset{OH}{&}}{\overset{Oh}{\overset{Oh}{\overset{Oh}{\overset{Oh}{\overset{Oh}{Oh}{\overset{Oh}{&}}{\overset{Oh}{\overset{Oh}{\\$	
	H <sub>3</sub> C <sub>N</sub> NN NNN 170561	170578	г, г , г, г , г,
HO HO THE HOLD THE HO	нс, от от то	ОН 0 N-OH3 172251	$H_{D} \xrightarrow{H_{2}C} N_{NO_{2}} \xrightarrow{H_{2}C} N_{NO_{2}} \xrightarrow{H_{2}C} N_{NO_{2}} \xrightarrow{H_{2}C} N_{NO_{2}} \xrightarrow{H_{2}C} 172614$
сн <sub>3</sub> с , сн <sub>5</sub> н <sub>3</sub> с , сн <sub>5</sub> , сн <sub>5</sub>	N NH <sub>2</sub> N CO <sub>2</sub> H 173723		СH <sub>3</sub> HO <sub>2</sub> C 174267
	$\bigcup_{\substack{H_2N \\ NH}} NO_2$ 175650	H <sub>S</sub> C H HN N N N N N N N N N N N N N N N N N	$n_{i_{1}} \rightarrow (i_{1}, \dots, i_{n}) \rightarrow (i_{n_{1}}, \dots, i_{n_{n_{n_{n_{n_{n_{n_{n_{n_{n_{n_{n_{n_$
$\downarrow_{CH_{5}}^{CH_{5}} \qquad $	$ \begin{array}{c}                                     $		
H H 2 C H H 2 C H H 2 C H H 2 C H H H 2 C H H H H		H <sub>3</sub> C H <sub>3</sub> CH <sub>3</sub> H <sub>3</sub> C H <sub>3</sub> CH <sub>3</sub> H <sub>3</sub> C CH <sub>3</sub> H <sub>3</sub> C CH <sub>3</sub> 177764	
$HO_{2C} \xrightarrow{HO_{2}C} \xrightarrow{HO_{2}} HO_$	$HO \rightarrow H \rightarrow$	HC HC HC OCH 181486	, , , , , , , , , , , , , , , , , , ,
но, С в страниции страници			

H <sub>3</sub> C <sup>-N</sup>	H.C. N.		
CH <sub>b</sub>		HO <sub>2</sub> C NH <sub>2</sub>	NH <sub>2</sub>
186063	186066	186915	186919
		H <sub>3</sub> C H	$\underbrace{\begin{array}{c} 0\\ H_2N, \\ 0\\ S \\ 0\\ H_3\end{array}}^{\text{NH}_3} \underbrace{\begin{array}{c} 0\\ NH_3\\ NH_3$
он 187494	187510	187538	187613
	Br		NO2 S S
№—‴ №н 188491	190336	190382	190577
	H <sub>3</sub> C	S CH3	
190694	191260	191411	193457
		$ \underset{CH_3}{\overset{\circ}{\underset{Br}{\overset{\bullet}{\underset{Br}{\overset{Br}{\underset{Br}{\underset{Br}{\overset{Br}{\underset{Br}{\underset{Br}{\overset{Br}{\underset{Br}{\underset{Br}{\overset{Br}{\underset{Br}{\underset{Br}{\overset{Br}{\underset{Br}{\underset{Br}{\underset{Br}{\overset{Br}{\underset{Br}{\underset{Br}{\underset{Br}{\underset{Br}{\underset{Br}{\atopBr}{\underset{Br}{\atopBr}{\underset{Br}{\atopBr}{\atopBr}{\atopBr}{\atopBr}{\atopBr}{\atopBr}{\atopBr}{$	
	H <sub>3</sub> C	сн.	()
H02C 195952		N O CH <sub>3</sub> 196148	201430
			201873
	H <sub>2</sub> NS	H <sub>3</sub> C N S OH	
ньс 201980	202409	202604	202695
H <sub>3</sub> C HO O CH <sub>3</sub>	H <sub>1</sub> C H <sub>2</sub> C H <sub>3</sub> C	HO HO Br OH	CI C
202789	× 203328	203337	204103
		он 204794	нострука и пользования и пользов И пользования и

о N СН <sub>3</sub> 204976		<sup>o</sup> , <sup>NH<sub>2</sub></sup> <sup>o</sup> , <sup>o</sup> ,	NH2 O H <sub>3</sub> C H <sub>3</sub> C
остория и сила и сил 205862	$\overset{HO}{\rightarrow} \overset{HP_{2}}{\longleftrightarrow} \overset{0}{\longleftrightarrow} \overset{0}{\longleftrightarrow$		
$H_{N}^{\text{H}_{3}C} \xrightarrow{CH_{3}}_{N-\overline{0}} \xrightarrow{H_{3}C}_{H_{3}} \xrightarrow{H_{3}C}_{H_{3}} \xrightarrow{CH_{3}}_{D-OH} \xrightarrow{H_{3}C}_{CH_{3}} \xrightarrow{208823}$	209910	a + f + a + s + s + s + s + s + s + s + s + s	
H <sub>3</sub> C OH CH <sub>3</sub> H <sub>3</sub> C OH CH <sub>3</sub> H <sub>3</sub> C OH CH <sub>3</sub> CH 210595	$H_{3}C \rightarrow O_{H_{3}}C \rightarrow O_{H_{3$		н <sub>2</sub> N Н С С С С С С С С С С С С С С С С С С
Br COOH HO Br 210786	H <sub>2</sub> C <sup>H<sub>3</sub></sup> H <sub>3</sub> C <sup>H<sub>3</sub></sup> C <sup>H<sub>3</sub> C<sup>H<sub>3</sub></sup> C<sup>H<sub>3</sub></sup> C<sup>H<sub>3</sub> C<sup>H<sub>3</sub></sup> C<sup>H<sub>3</sub> C<sup>H<sub>3</sub></sup> C<sup>H<sub>3</sub> C<sup>H<sub>3</sub> C<sup>H<sub>3</sub> C<sup>H<sub>3</sub> C<sup>H<sub>3</sub> C<sup>H<sub>3</sub> C<sup>H<sub>3</sub> C<sup>H<sub>3</sub> C<sup>H<sub>3</sub> C<sup>H<sub>3</sub> C<sup>H<sub>3</sub> C<sup>H<sub>3</sub> C<sup>H<sub>3</sub> C<sup>H<sub>3</sub> C<sup>H<sub>3</sub> C<sup>H<sub>3</sub> C<sup>H<sub>3</sub> C<sup>H<sub>3</sub> C<sup>H<sub>3</sub> C<sup>H<sub>3</sub> C<sup>H<sub>3</sub> C<sup>H<sub>3</sub> C<sup>H<sub>3</sub> C<sup>H<sub>3</sub> C<sup>H<sub>3</sub> C<sup>H<sub>3</sub> C<sup>H<sub>3</sub> C<sup>H<sub>3</sub> C<sup>H<sub>3</sub> C<sup>H<sub>3</sub> C<sup>H<sub>3</sub></sup></sup></sup></sup></sup></sup></sup></sup></sup></sup></sup></sup></sup></sup></sup></sup></sup></sup></sup></sup></sup></sup></sup></sup></sup></sup></sup></sup></sup></sup></sup></sup></sup></sup>	H <sub>4</sub> C 0 0 0 0 0 0 0 0 0 0 0 0 0	
	н <sub>з</sub> с <u>р</u> <u>он</u> <u>он</u> <u>211685</u>	Litrain 211736	
$\overbrace{C_{I}}^{C_{I}} \overbrace{C_{I}}^{H_{J}C_{I}} \overbrace{H_{J}C_{I}}^{H_{J}C_{I}} \overbrace{O}^{CH_{J}} \overbrace{O}^{CH_{J}}$ 216633	но СН <sub>3</sub> Он но Но Он но 220204		$H_{3}C$ $H_{3}C$ $H_{3}C$ $H_{3}C$ $H_{3}C$ $H_{3}C$ $H_{3}N$ $H$
H <sub>3</sub> C H <sub>3</sub> C SH 227265		231596	$\overbrace{OH}^{CH_{5}}_{NH} \xrightarrow{N}_{OH} \xrightarrow{N}_{H_{5}C}$ 231643
	234766		$H_{3}C$ $H$
Ho Ho 241993	H <sub>3</sub> C CH <sub>3</sub> H <sub>3</sub> C CH <sub>3</sub> 245053	н <sub>а</sub> стания 246981	ОН НN HN 247535



ИН ОН ОН 289336	() → → → → → → → → → → → → →	но-состатование с состатование с состато С состатование с соста С состатование с сост	$\overset{\circ}{\underset{\substack{ \overset{\bullet}{ \overset{\bullet}}{ \overset{\bullet}}$
$H_{20} \xrightarrow{CI} H_{20} \xrightarrow{PI - S} N \xrightarrow{CH_{3}} CH_{3}$ 292596	HLN WH	HOLE CONTRACT OF THE STREET STRE	
и страна и страна сна 293958	$ \begin{array}{c} \begin{array}{c} \begin{array}{c} \begin{array}{c} \end{array} \\ \end{array} \\ \end{array} \\ \end{array} \\ \begin{array}{c} \end{array} \\ \end{array} \\ \end{array} \\ \begin{array}{c} \end{array} \\ \end{array} \\ \end{array} \\ \begin{array}{c} \end{array} \\ \end{array} \\ \end{array} \\ \end{array} \\ \begin{array}{c} \end{array} \\ \end{array} \\ \end{array} \\ \end{array} \\ \end{array} \\ \begin{array}{c} \end{array} \\ \end{array} \\ \end{array} \\ \end{array} \\ \end{array} \\ \end{array} $ } \\ \end{array} \\ \end{array}  } \\ \end{array} \\ \end{array} \\ \end{array}  }  } \\ \end{array}  } \\ \end{array}  } \\ }  } \\ }  } \\ \end{array}  } \\ }  } \\ }  } \\ }  } \\ }  } \\ }  } \\ }  } \\ }  } \\ }  }  } \\ }  } \\ }  } \\ }  } \\ }  }  }  } \\ }  }  }  } \\ }  }  } \\ }  }  } \\ }  }  } \\ }  }  } \\ }  }  } \\ }  } \\ \rangle  }  }  }  }  }  }  }  }  }  }	$ \begin{array}{c}                                     $	N A CH3 N
$ \overset{H_{0}}{\underset{n}{}} \overset{H_{0}}{\underset{n}{\overset{H_{0}}}} \overset{H_{0}}{\underset{n}{}} \overset{H_{0}}{\underset{n}{\overset{H_{0}}}} \overset{H_{0}}{\underset{n}{}} \overset{H_{0}}{\underset{n}{}} \overset{H_{0}}{\underset{n}{}} \overset{H_{0}}{\underset{n}{}} \overset{H_{0}}}{\overset{H_{0}}} \overset{H_{0}}{\underset{n}{}} \overset{H_{0}}}{\overset{H_{0}}} \overset{H_{0}}{\underset{n}{}} \overset{H_{0}}{\underset{n}{}} \overset{H_{0}}{\overset{H_{0}}}} \overset{H_{0}}{\overset{H_{0}}} \overset{H_{0}}{\overset{H_{0}}}} \overset{H_{0}}}{\overset{H_{0}}} \overset{H_{0}}{\overset{H_{0}}} \overset{H_{0}}}{\overset{H_{0}}} \overset{H_{0}}{\overset{H_{0}}} \overset{H_{0}}}{\overset{H_{0}}} \overset{H_{0}}}{\overset{H_{0}}} \overset{H_{0}}}{\overset{H_{0}}} \overset{H_{0}}} \overset{H_{0}}}{\overset{H_{0}}} \overset{H_{0}}}{\overset{H_{0}}}} \overset{H_{0}}} \overset{H_{0}}}{\overset{H_{0}}} \overset{H_{0}}} H_$	$\downarrow_{\mu_{s} \leftarrow CH_{s}}^{\mu_{s}} \qquad $	$H_3C \xrightarrow{O} \xrightarrow{H} \xrightarrow{I} \xrightarrow{O} \xrightarrow{O} \xrightarrow{I} \xrightarrow{I} \xrightarrow{O} \xrightarrow{I} \xrightarrow{O} \xrightarrow{I} \xrightarrow{O} \xrightarrow{I} \xrightarrow{O} \xrightarrow{I} \xrightarrow{I} \xrightarrow{O} \xrightarrow{I} \xrightarrow{O} \xrightarrow{I} \xrightarrow{I} \xrightarrow{I} \xrightarrow{O} \xrightarrow{I} \xrightarrow{I} \xrightarrow{I} \xrightarrow{O} \xrightarrow{I} \xrightarrow{I} \xrightarrow{I} \xrightarrow{O} \xrightarrow{I} \xrightarrow{I} \xrightarrow{I} \xrightarrow{I} \xrightarrow{I} \xrightarrow{I} \xrightarrow{I} I$	а
140		H,co H,co H,co H,co H,co H,co H,co H,co	299130
299137	299209	299210	
	ньс оснь 302042	NO <sub>2</sub> N, CO <sub>2</sub> H 302988	H <sub>2</sub> N S H <sub>N</sub> 303288
сл NH 303289	H <sub>3</sub> C - (	о , , , , , , , , , , , , , , , , , , ,	$H_{3}C$ $H_{0_{2}}C$ $H_{0_{2}}C$ $H_{3}$ $H_{0_{2}}C$ $H_{3}$ $H_{0_{2}}C$ $H_{3}$ $H_{0}C$ $H_{1}C$ $H_{1$
303612		HOWM HE CH2OC(O)CH2NH2 HOWM HE CH2OC (O)CH2NH2 HOWM HE CH2OC (O)CH2NH2 HOW HE CH2OC (O)CH2NH2 HE CH2OC (O)	NH H2 СС NH NH2 303851
BE- 		305787	$(r_{3})^{\text{CH}} (r_{3})^{\text{CH}} (r_{3})^{\text{CH}}$

305821	305831	<sup>bo;</sup> 306698	306711
Рр. H <sub>2</sub> со	H <sub>4</sub> C <sub>0</sub> OH OF OH OH OF OH OH OF OH OH OH OH OH OH OH OH OH		H <sub>2</sub> N Se Se NH <sub>2</sub> 307711
NCH2-CH2NEt H2N 308820		СF <sub>3</sub> 309874	$ \begin{array}{c} \begin{array}{c} \begin{array}{c} \begin{array}{c} \begin{array}{c} \end{array}\\ \end{array}\\ \end{array} \\ \end{array} \\ \begin{array}{c} \end{array} \\ \end{array} \\ \begin{array}{c} \end{array} \\ \begin{array}{c} \end{array} \\ \end{array} \\ \end{array} \\ \end{array} \\ \begin{array}{c} \end{array} \\ \end{array} $ } \\ \end{array} \\ \end{array} \\ \end{array} \\ \end{array} \\ \end{array} \\ \end{array}  } \\ \end{array} \\ \end{array} \\ \end{array} \\ \end{array} \\ \end{array} \\ \end{array}  } \\ \end{array} \\ \end{array} \\ \end{array} \\ \end{array} \\ \end{array}  } \\ \end{array}  } \\ \end{array}  } \\ \end{array}
	$\overset{\alpha}{\underset{H}{\leftarrow}}\overset{\alpha}{\underset{CN}{\leftarrow}}\overset{\alpha}{\underset{N}{\leftarrow}}\overset{\alpha}{\underset{K}{\leftarrow}}\overset{\alpha}{\underset{N}{\leftarrow}}$	HO <sub>2</sub> C N S 309939	CH <sub>3</sub> CH <sub>5</sub> CH <sub>5</sub> CH <sub>5</sub> CH <sub>5</sub> CO <sub>2</sub> H 310038
HO <sub>2</sub> C N CH <sub>3</sub> 310099	HEN S CON CI HIN S CON CI 310342	$\downarrow_{HSC} \xrightarrow{OL} 0 \qquad HN \qquad $	a 310352
s cr s s s s s s s s s s s s s s s s s s	310366		HEGGE CHARACTER STREET
310839	$ \underset{H_{i}}{\overset{H_{i}}{\underset{H_{i}}{\overset{H_{i}}{\underset{H_{i}}{\overset{H_{i}}{\underset{H_{i}}{\overset{H_{i}}{\underset{H_{i}}{\overset{H_{i}}{\underset{H_{i}}{\overset{H_{i}}{\underset{H_{i}}{\overset{H_{i}}{\underset{H_{i}}{\underset{H_{i}}{\overset{H_{i}}{\underset{H_{i}}{\underset{H_{i}}{\overset{H_{i}}{\underset{H_{H_{i}}{\underset{H_{i}}{H_{H$	$\downarrow_{HO} \xrightarrow{(H_{0})}_{CH_{0}} \xrightarrow{(H_{0})}_{311153}$	$ \underbrace{ \begin{array}{c} & & \\ &$
$\begin{array}{c} \hline \\ \\ \hline \\$	$H_{2}C_{H_{3}}$	NH2 NH2 NH2 314885	CH <sub>3</sub> H S S S H 318538
СІ СІ ОН О Н Рh 318824	СО <sub>2</sub> H 319095	$a_{+} \underbrace{ \begin{pmatrix} c \\ c$	ни сли в страна в стран
н <sub>ус</sub>	H <sub>2</sub> N H <sub>3</sub> C OH OH 319443	о <sub>сн3</sub> о <sub>н2</sub> N 319447	

CH Me OH OH OH OH OH OH OH OH OH OH	$H_{C} \xrightarrow{\alpha} (f) = (f) + (f) +$		$ \underset{o}{\overset{H,C}{\underset{o}{\leftarrow}}} \underset{o}{\overset{O}{\underset{o}{\leftarrow}}} \underset{o}{\overset{O}{\underset{o}{\leftarrow}}} \underset{o}{\overset{O}{\underset{o}{\leftarrow}}} \underset{OH}{\overset{OH}{\underset{o}{\leftarrow}}} $ 320209
ну н з20217	NH <sub>2</sub> О ОН ОН 320308	H <sub>3</sub> C 0 H <sub>3</sub> C 0 H <sub>0</sub> O H <sub>0</sub> O	$HO \begin{array}{c} OH \\ HO $
		Суркание и сила и с З 20870	
но но сн 321237		a states and the states of the	() + () + () + () + () + () + () + () +
CH2 CH2 OH2 OH2 OH2 OH2 OH2 OH2 OH2 OH2 OH2 O	$H_{3}C \xrightarrow{CH_{3}}CH_{3}$ $H_{3}C \xrightarrow{CH_{3}}CH_{3}$ $H_{3}C \xrightarrow{CH_{3}}CH_{3}$ $321239$	$\overset{H_{2}C}{\underset{HC}{\leftarrow}}\overset{H_{3}CH_{3}}{\underset{HC}{\leftarrow}}\overset{H_{3}CH_{3}}{\underset{HC}{\leftarrow}}\overset{H_{3}CH_{3}}{\underset{HC}{\leftarrow}}$	$ \begin{array}{c} ^{\downarrow}_{H_{c}} \\ \downarrow_{H_{c}} \\ \downarrow_{H$
$\begin{array}{c} & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ &$	н <sub>6</sub> с- н <sub>6</sub> с- н <sub>6</sub> с- сн <sub>6</sub> 321568	$ \underset{l \in \mathcal{C}_{H_{0}}}{\overset{H_{1} \subset \mathcal{C}_{H_{0}}}{\mapsto}} \underset{l \in \mathcal{C}_{H_{0}}}{\overset{H_{1} \subset \mathcal{C}_{H_{0}}}{\mapsto}} \underset{l \in \mathcal{C}_{H_{0}}}{\overset{H_{1} \subset \mathcal{C}_{H_{0}}}{\mapsto}} 321578 $	HN HNN NH NN NH 321596
CO <sub>2</sub> H 0 CO <sub>2</sub> H 0 CO <sub>2</sub> H 321598	H <sub>3</sub> C NC NC 321785	CI FF CI FF HO HO 322661	$\overbrace{}^{}_{} \downarrow \downarrow$
, , , , , , , , , , , , , , , , , , ,	signature for the second secon	$H_{S}C$ $H_{N}C$ H	$ \underset{c_{2}}{\overset{H_{G}}{\underset{D_{2}}{\leftarrow}}} \underset{C_{2}}{\overset{H_{G}}{\underset{D_{2}}{\leftarrow}}} \underset{D_{2}}{\overset{H_{G}}{\underset{D_{2}}{\leftarrow}}} $
		о	s CH <sub>3</sub> 327447


H <sub>2</sub> N, S, P, OH NH <sub>2</sub>			
350191	сн <sub>а</sub> 350363	но 350378	<sup>0</sup> 350625
HS N 351112	351123		Сурнание и страна и стр
	H <sub>3</sub> C <sup>0</sup> H	$\underset{_{H,H}}{\overset{H}}\overset{H}{\overset{H}$	H <sub>3</sub> C CH <sub>3</sub> H <sub>3</sub> C H <sub>3</sub> C OH H <sub>3</sub> C OH H <sub>3</sub> C OH H OH OH OH OH OH OH
CH <sub>5</sub> H <sub>C</sub> H OH OH 355081	H_N K N CH3 356363	$H^{\bullet} \xrightarrow{O}_{O_2C} \xrightarrow{O}_{O_2} \xrightarrow{H_2} \xrightarrow{H_2}$	$ \begin{array}{c}                                     $
$(t_{ij}, t_{ij}) = (t_{ij}, t_{ij})$		H <sub>5</sub> C, NH <sub>2</sub> OH OH OH 359098	Hyc N Hyc 359449
со <sub>2</sub> н 360211	HO <sub>2</sub> C HO	н <sub>а</sub> с-С-С-К-С-К-С-К-З 360218	СЦСА З60494
	s s s d 1814	° → , , , , , , , , , , , , , , , , , ,	н <sub>3</sub> с NH <sub>2</sub> NH <sub>2</sub> ОН 362121
$ \begin{array}{c} \begin{array}{c} & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ \end{array} \right)  $		Signal Si	H <sub>2</sub> C-N <sub>CH3</sub> S H <sub>3</sub> C H <sub>3</sub> 363916
	ССС ,	HO N N N N N NO2 366163	с
500012	50/40/	506270	500755

## Maleeruk Utsintong

Appendix / 166



H <sub>3</sub> C, H <sub>3</sub> S, S, C, C, C, H <sub>3</sub>	$\underset{H_{0}C}{\overset{H_{0}C}{\underset{H_{0}C}{}{\underset{F}{F$		
379543	379546	379552	379553
$ \underset{Ci}{\overset{N=N}{\underset{N+e}{}}} \underset{N+e}{\overset{N=N}{\underset{N+e}{}}} \underset{379640}{\overset{S}{\underset{N+e}{}}} $			
		HO <sub>2</sub> C, CI	
382098	H <sub>2</sub> C 00 <sub>2</sub> H 400159	400464	
401077	H <sub>3</sub> C <sub>N</sub> H H H 401366	$\overset{H_{3}C}{} \overset{G}{} \overset{H_{2}N}{} \overset{O}{} \overset{CH_{3}}{} \overset{G}{} \overset{G}{} \overset{G}{} \overset{G}{} \overset{H_{3}N}{} \overset{G}{} \overset{G}$	
HIC, NOT ON 403225	$HO_2$	s=c=nsссе_2н 403376	H <sub>2</sub> N S OH NH 403734
СН СН <sub>3</sub> 403822	HO	407628	
	$\overset{CH_3}{\xrightarrow{H_0}}_{HO} \overset{O}{\xrightarrow{H_0}}_{HO} \overset{O}{\xrightarrow{408456}}$	409455	409664
	но с с с с с с с с с с с с с с с с с с с		HOOC NH <sub>2</sub> 521717
HO H	HN NH <sub>2</sub> 523303	$\begin{array}{c} \overset{u_{\mathcal{C}_{n}}}{\longrightarrow} u_{\mathcal$	н <sub>2</sub> N S S OH 528168
	N HO HO CH3		H <sub>5</sub> C NW <sup>M</sup>
600067	601351	601364	602670

## Maleeruk Utsintong

	$H_{O} \xrightarrow{H_{3}C} H_{O} \xrightarrow{H_{3}C} H_{3} H_{3$	(f) = (f) + (f)	
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н <sub>я</sub> с он вr 647137			$(H_{1,C}, M_{1,C}) \xrightarrow{OH}_{H_{1,C}} (H_{1,C}) (H_$
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Biography / 170

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	Commission on Higher Education	