

**Topic:** Development of Water Gas Shift Catalysts For Converting Synthesis Gas to Hydrogen Rich Gas

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## ABSTRACT

The water gas shift reaction is generally applied to produce hydrogen from the reaction of carbon monoxide and water. In the present work, the water gas shift reaction over several ceria-based catalysts (i.e. CeO<sub>2</sub>, GDC and YDC) was studied. It is noted that two catalyst preparation methods (i.e. precipitation and low temperature preparation) were compared. Furthermore, the effect of Ni impregnation to promote the reaction was also carried out.

From both preparation techniques, the fluorite oxide-type of CeO<sub>2</sub> was observed, according to XRD characterization. The Gd<sub>2</sub>O<sub>3</sub> and Y<sub>2</sub>O<sub>3</sub> peaks were not found in GDC and YDC patterns indicating that Gd<sup>3+</sup> and Y<sup>3+</sup> incorporated into subsurface region of CeO<sub>2</sub> lattice to form solid solution. An addition of Ni did not cause any change in XRD patterns, presuming that Ni was highly dispersed on the surface of the supports. From the water gas shift activity test, it was found that the impregnation of Ni significantly promoted the catalyst activity. Among the catalysts prepared by low temperature method, 5%Ni/20YDC showed the best activity, from which the CO conversion started at the temperature above 250 °C and reached 75% conversion at 350 °C before dropping at higher temperature as it reached the equilibrium condition. In case of catalysts prepared by co-precipitation method, 5%Ni/CeO<sub>2</sub> showed the highest activity, from which the CO conversion reached 92% conversion at 400 °C. The activation energy and reaction orders of H<sub>2</sub>, CO and CO<sub>2</sub> of 5%Ni/CeO<sub>2</sub> were predicted by the kinetics study. From the study, the activation energy (E<sub>a</sub>) of 5%Ni/CeO<sub>2</sub> was 27 kJ/mol, while the reaction orders of H<sub>2</sub>, CO and CO<sub>2</sub> were 0.69, -0.13 and -0.13, respectively. This indicates the weak negative effect of H<sub>2</sub> and CO<sub>2</sub> on the reaction due to the promotion of reverse water gas shift reaction by both compounds.

**Keywords:** CeO<sub>2</sub>; water gas shift reaction; catalyst; low temperature preparation technique