

Thesis Title

Synthesis and Characterisation of Copper(I) Thiourea
and Substituted Thiourea Complexes

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Abstract

Copper(I) thiourea complexes were synthesized by 3 methods according to the literature. Two methods claimed to obtain $\text{Cu}_2(\text{tu})_8\text{SO}_4 \cdot 2\text{H}_2\text{O}$ and the third method was reported to give $\text{Cu}_2(\text{tu})_5\text{SO}_4 \cdot 2\text{H}_2\text{O}$. However, the results of this investigation showed that the formulae for the complexes prepared by the first method should be $\text{Cu}_2(\text{tu})_5\text{SO}_4$ and the second method should be $\text{Cu}_2(\text{tu})_8\text{SO}_4$ and the third method gave the same formula as reported in the literature. It was found that when calcined and uncalcined Thai-kaolin clay, Mae-rim clay and English kaolin clay were added in the preparation of the complexes by the 3 methods, there was no change to the formulae of the complexes. Complexes of copper halide (CuX_2 ; $\text{X} = \text{F}, \text{Cl}$) and thiourea were also synthesized and $\text{Cu}(\text{tu})_2\text{F}$ and $\text{Cu}(\text{tu})_3\text{Cl}$ obtained. The synthesis of copper(I) substituted thiourea $\text{Cu}_2(\text{atu})_3\text{SO}_4 \cdot 2\text{H}_2\text{O}$, $\text{Cu}_2(\text{etu})_8\text{SO}_4$ and $\text{Cu}_2(\text{S-dptu})_4\text{SO}_4$ was also carried out. Infra-red spectra of the complexes showed that thiourea, acetylthiourea, sym-diphenylthiourea and ethylenethiourea used sulphur atom in the coordination with copper(I).